

FRITZ-HABER-INSTITUT
MAX-PLANCK-GESELLSCHAFT



Big Data Summer

A summer school of the BiGmax Network

Platja d'Aro, Spain, September 9 – 13, 2019

Clean data acquisition in oxidation catalysis

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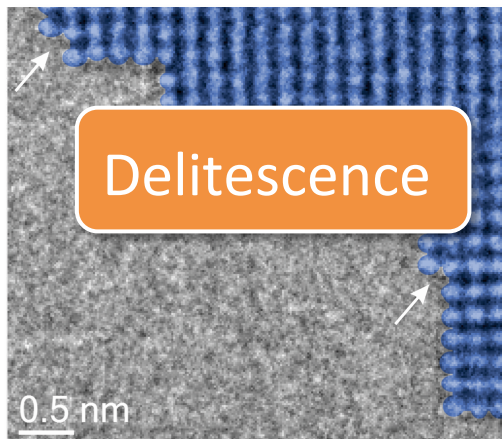
Data: Raw material to generate understanding

- Data will have to be
 - Well documented by meta data
 - Stored in a findable way
 - Documented and stored in a re-useable form
 - Clean enough to be interoperable with later and external analysis
 - Complete enough to allow reconstruction of the experiment



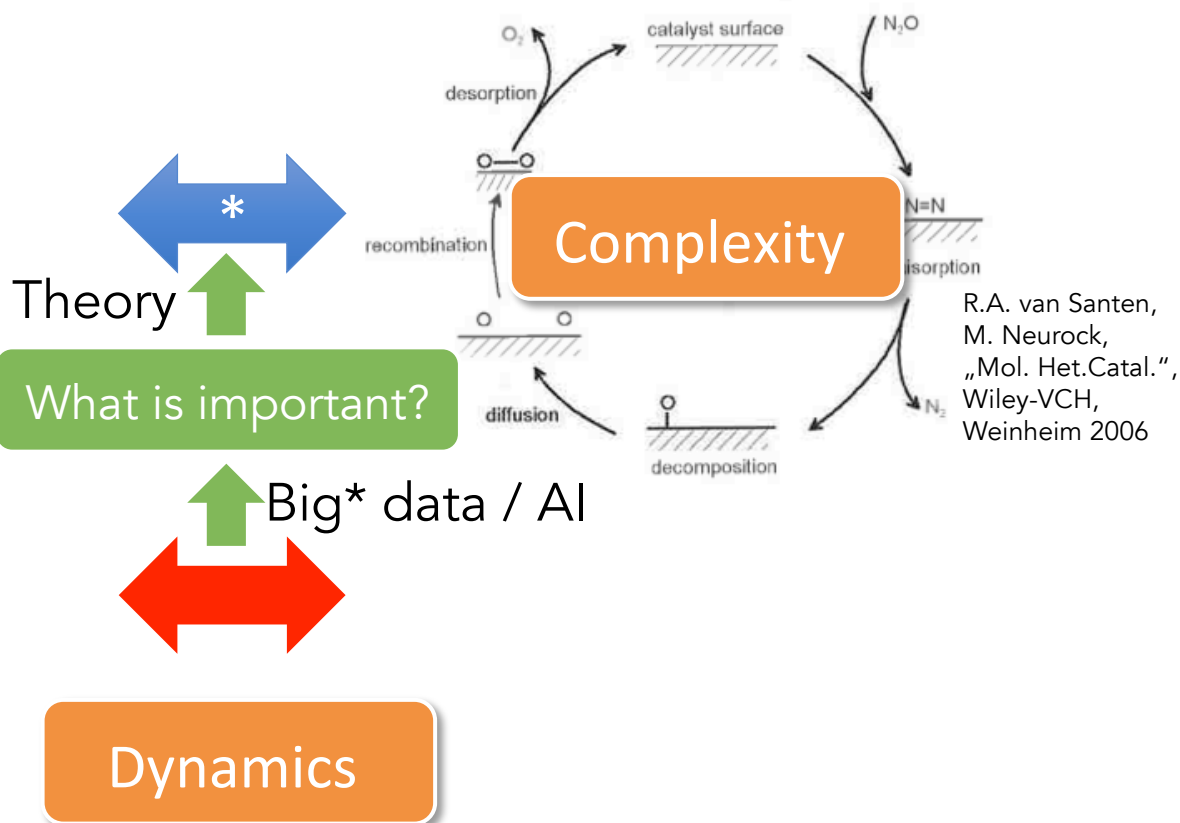
Challenges and chances – types of data

Active site



Journal of Catalysis 2015, 326, 560-573.

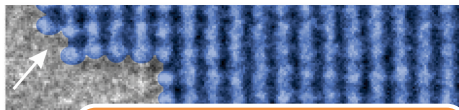
Catalytic cycle / mechanism



*but with much less data compared to YouTube (10^9 GB)

Challenges and chances

Active site



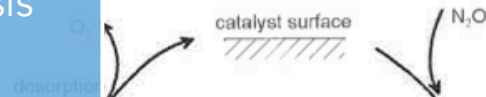
Characterization of activated and spent catalysts

- Bulk and surface analysis of powders
- Synthesis conditions including raw materials and history

Model investigations

- Surface science analysis
- DFT calculations

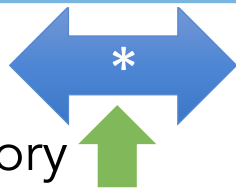
Catalytic cycle / mechanism



Reaction kinetics

- Process conditions and reactor type
- Analytical methods of concentration determination
- Method of calculations applied in activity/selectivity determination

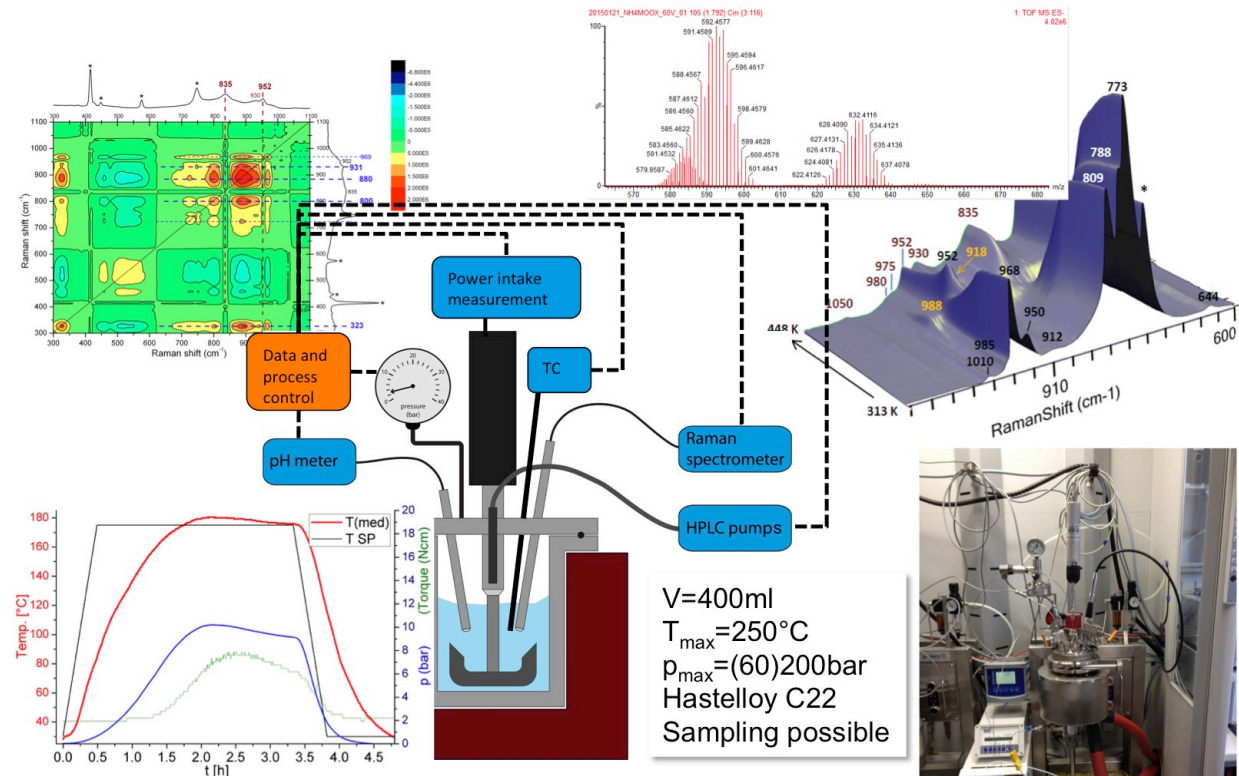
Theory



Operando spectroscopy

- Simultaneous measurement of chemical/physical properties and functional properties in one experiment

Reproducible synthesis

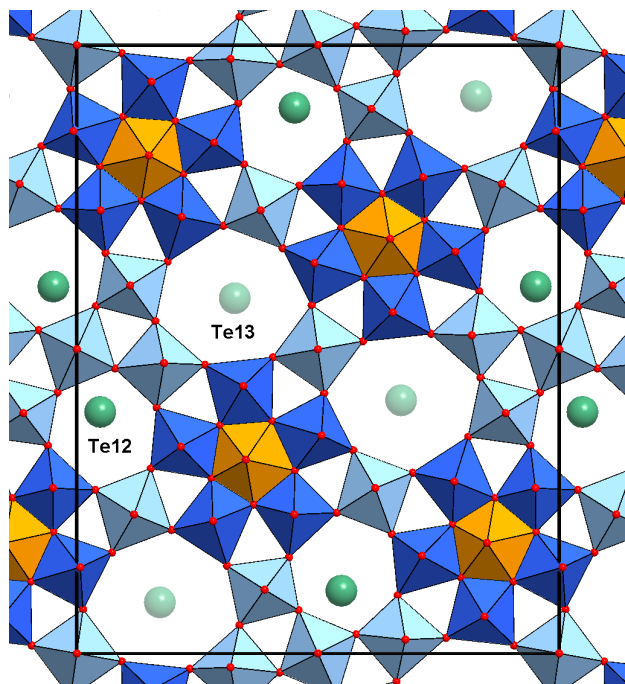


HPM-PT-040 (Premex Reactor GmbH)
Kaiser Optics Raman Spectrometer RXN1

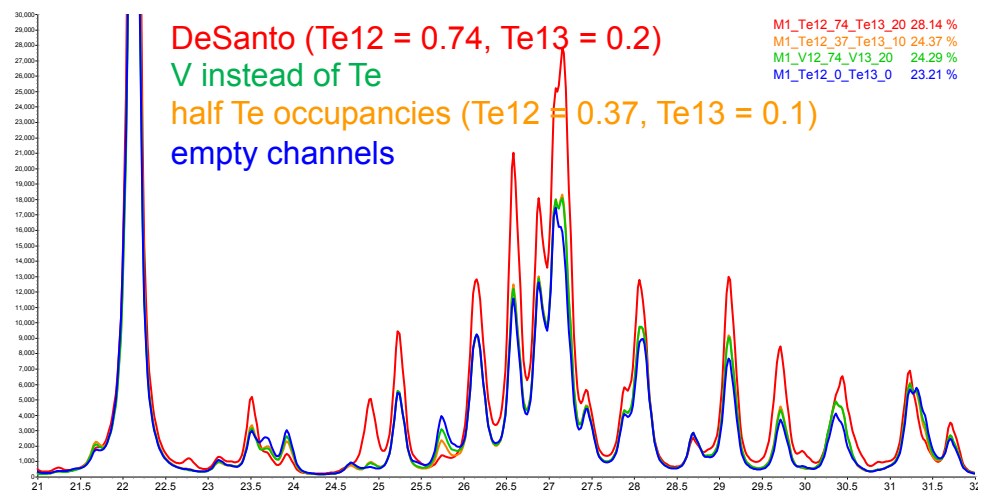
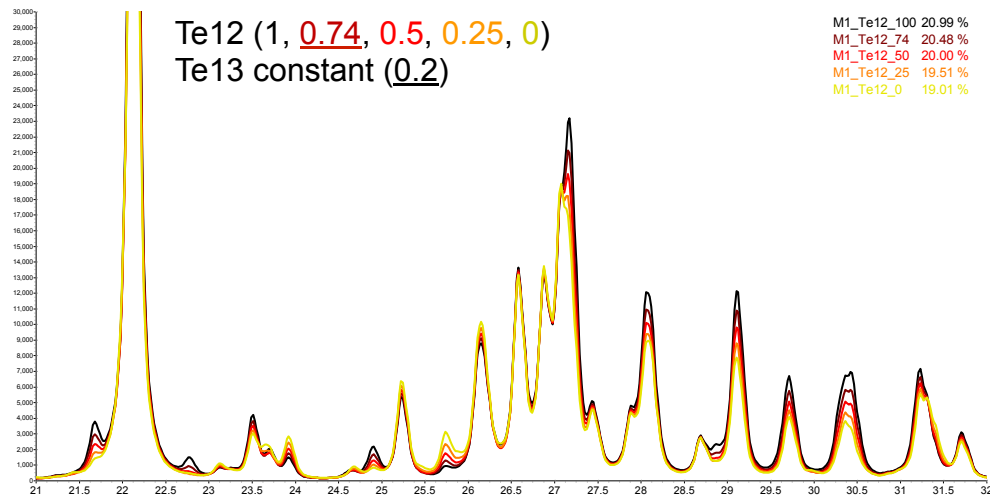


- Automated synthesis reactors deliver synthesis metadata
- Difficulties in automation:
 - Separation and washing steps
 - Activation procedures

Reproducible synthesis

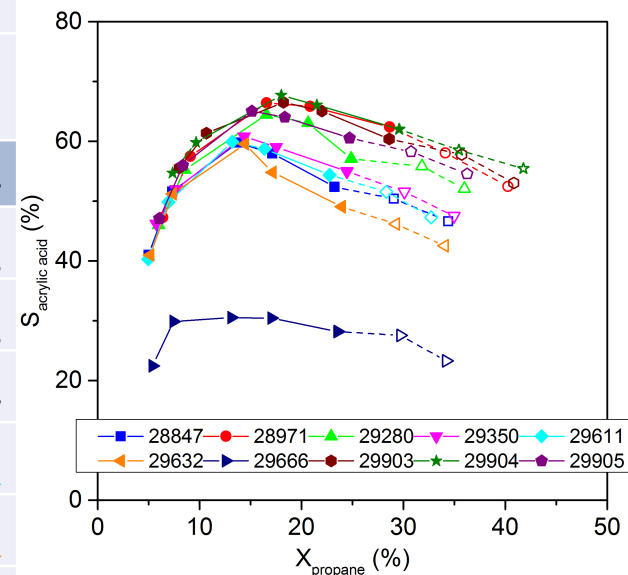


Simulations: Frank Girgsdies

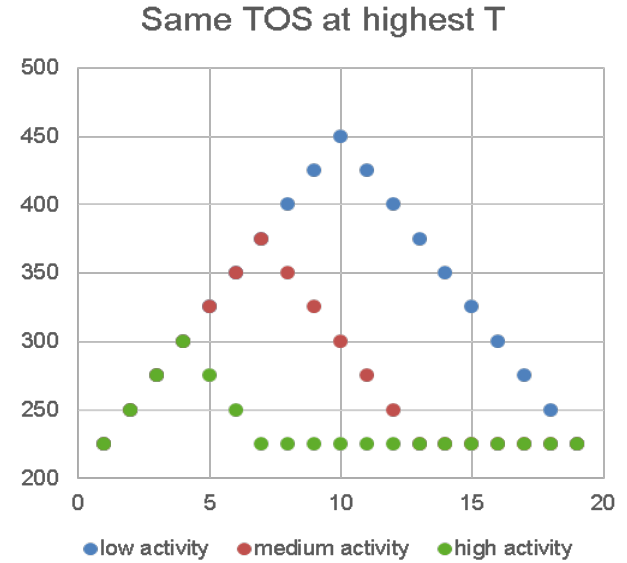
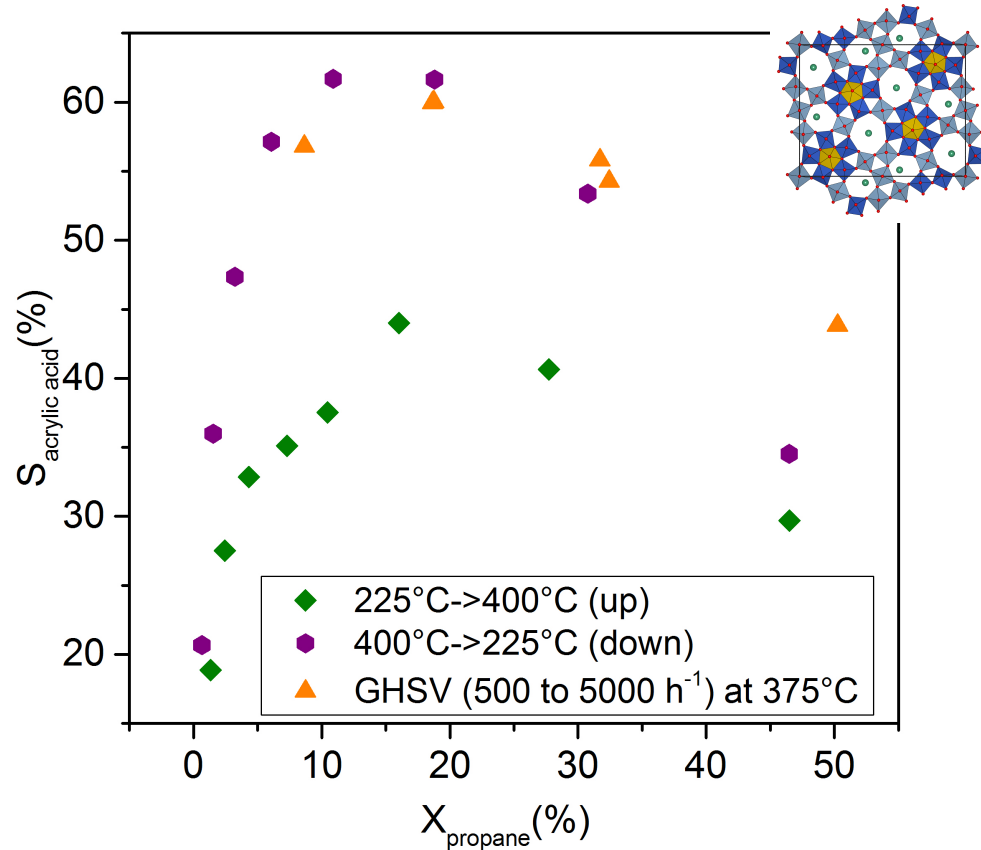


"Reproducible" synthesis

SN	m (g)	XRD lattice parameter			XRD	XRF			
		a	b	c		M1 phase (%)	Mo	V	Te
29200	130.11	Precursor			-	0.4474	0.0978	0.0813	0.0997
29258 ^A	1.69	21.1682790	26.6586529	4.0158877	97.69	0.4579	0.0994	0.0837	0.0873
29350 ^B	8.59	21.1667252	26.6579321	4.0164262	97.74	0.4546	0.0979	0.0835	0.0913
29425 ^A	1.71	21.1713859	26.6656641	4.0171937	97.57	0.4566	0.1004	0.0840	0.0877
29611 ^B	8.10	21.1723662	26.6627860	4.0170173	97.96	0.4575	0.0995	0.0840	0.0874
29632 ^B	8.34	21.1715227	26.6655441	4.0171018	97.66	0.4522	0.1006	0.0829	0.0924
29666 ^B	8.30	21.1699864	26.6621774	4.0164017	97.89	0.4522	0.1010	0.0828	0.0924

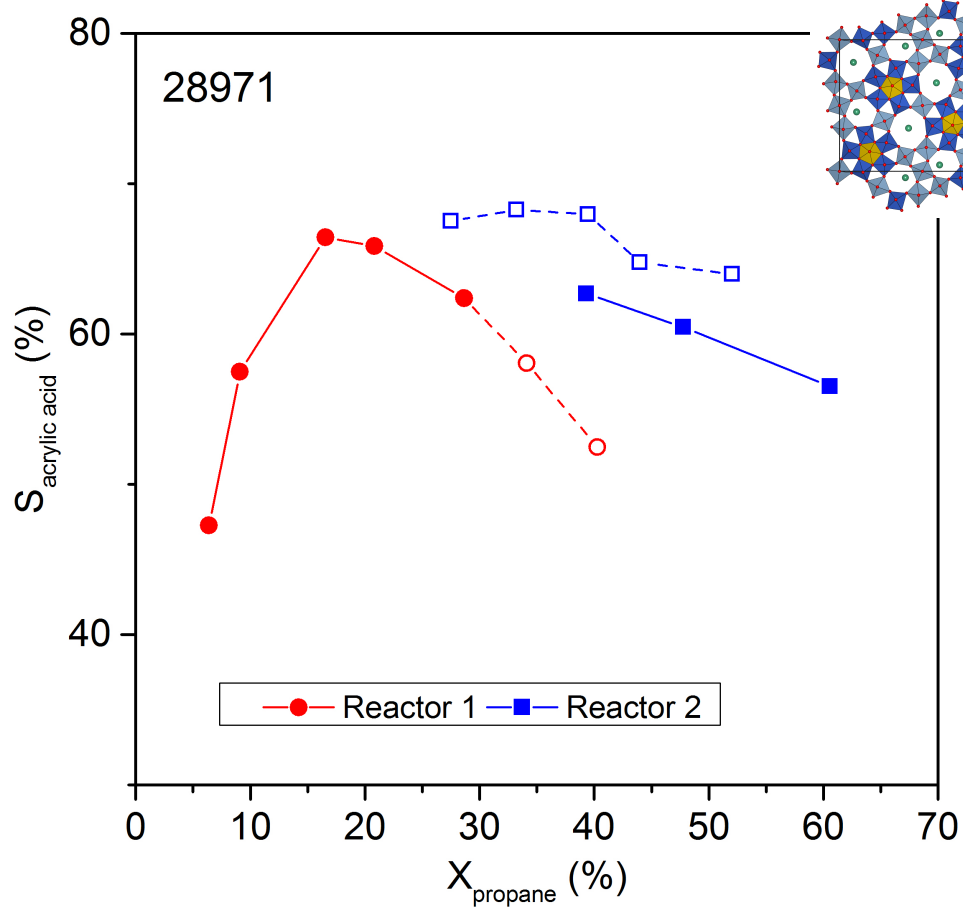


Sequence of catalytic test experiment



- Sequence matters due to the chemical memory of catalysts
- Catalysts needs to be equilibrated at first

Heat and mass transport phenomena



■ Pressure drops, carbon balance,...

Calculation of activity

Conversion X

$$X_i = \frac{n_{i,0} - n_i}{n_{i,0}}$$

Reaction rate r

$$r = \frac{1}{\nu_i m_{cat}} \frac{dn_i}{dt}$$

Turn over frequency

$$TOF [s^{-1}] = \frac{N_i}{N_{sites} \cdot t} [s^{-1}]$$

- Nature and number of active sites usually unknown
- Difficult / impossible to measure TOF in heterogeneous catalysis !!!

Calculation of selectivity

	key compound consumed	sum of products
stoichiometric factors	formally correct, requires high knowledge level S_{cf}	S_{pf}
number of carbon atoms	S_{cn}	most convenient, requires minimum of knowledge S_{pn}

$$S_{cf}(\text{product}) = \frac{\frac{\nu(\text{key compound})}{\nu(\text{product})} \cdot c(\text{product})_{out}}{c(\text{key compound})_{in} - c(\text{key compound})_{out}}$$

$$S_{pn}(\text{product}) = \frac{\frac{n_{\text{carbon atoms}}(\text{product})}{n_{\text{carbon atoms}}(\text{key compound})} \cdot c(\text{product})}{\sum \frac{n_{\text{carbon atoms}}(\text{product})}{n_{\text{carbon atoms}}(\text{key compound})} \cdot c(\text{product})}$$

- Preferred raw data is the concentration as a function of time

Developing quality control



THEORY DEPARTMENT
Fritz-Haber-Institut, MPG



1. Definition of clear rules to test and characterize catalysts in form of an experimental **handbook**
2. Synthesis of 10 well-known catalysts in selective oxidation that show **diversity**

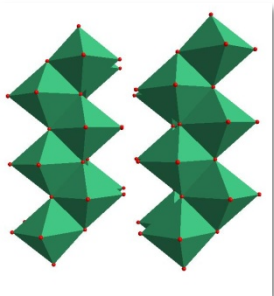
Data Science Project in Oxidation
Catalysis

Generation of a Reference Data Set

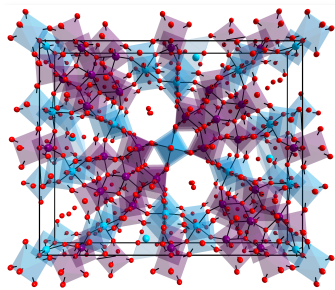
3. Testing (C2-C4 oxidation) and characterization of the catalysts according to the workflow described in the handbook involving **activated and spent** catalysts
4. Data evaluation – finding a continuous function $P(\mathbf{d})$ that generalizes the discrete data sets $\{P_i, \mathbf{d}_i\}$

Ghiringhelli, L. M.; Vybiral, J.; Levchenko, S. V.; Draxl, C.; Scheffler, M., Big Data of Materials Science: Critical Role of the Descriptor. *Physical Review Letters* **2015**, *114*, 105503.

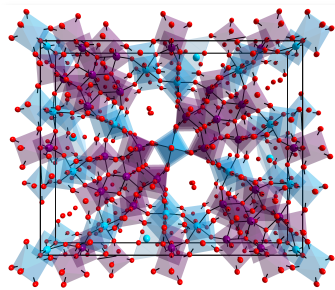
Catalyst platform for training in ethane, propane, *n*-butane oxidation



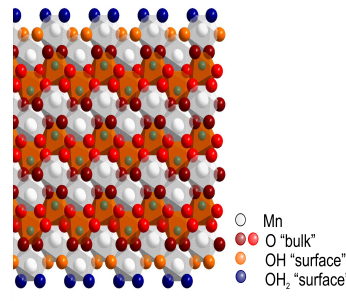
MoO₃



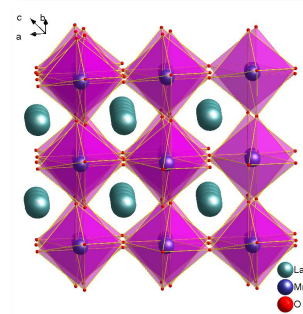
Mo-V oxide



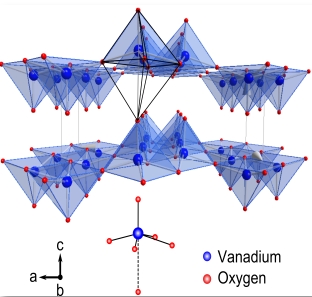
Mo-V-Te-Nb oxide



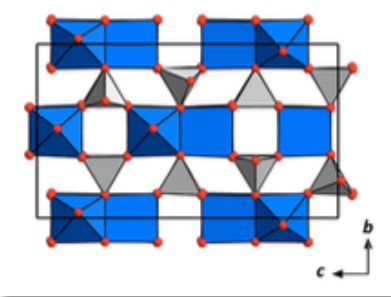
MnWO₄



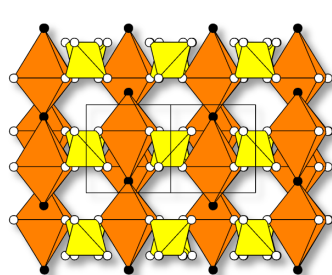
SmMnO₃



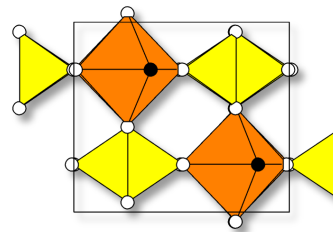
V₂O₅



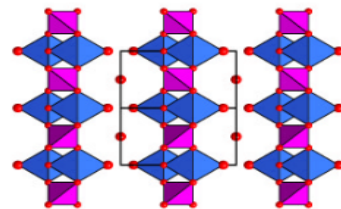
(V^{IVO})₂P₂O₇



α_{II} -VOPO₄



β -VOPO₄



VOPO₄*2H₂O

10 g of each, pressed and sieved

Kinetic data

Catalyst ID	Catalyst preparation			Characterization				Catalysis				
	Synthesis technique	...	Activation T_{max} [°C]	...	ICSD	a [Å]	n-butane oxidation			...
								Feed	r (300°C) [mmol g ⁻¹ h ⁻¹]	r (300°C) [mmol m ⁻² h ⁻¹]	E_a [kJ mol ⁻¹]	
1	HT		650					Standard				
								Less O ₂	-	-	-	
								ODH	-	-	-	
								Fuel rich	-	-	-	
								5% water	-	-	-	
								10% water	-	-	-	
2	P							Standard				
								Less O ₂	-	-	-	
								ODH	-	-	-	
								Fuel rich	-	-	-	
								5% water	-	-	-	
								10% water	-	-	-	
20% water	-	-	-									
...												

- Single rate not useful
- Broad parameter space and different reactions increase the number of data
- How to optimize selectivity?
For example:

$$\frac{1}{r_{C_3H_8}} = f(r_{CO_2})$$

Chemical and physical properties

Property	Physical value	Methods
Bulk crystal structure (ICSD#)	Lattice constants a, b, c [Å], α, β, γ [°] Unit cell volume V [nm ³] crystallographic density $\rho_{crystal}$ [g/cm ³]	XRD, TEM
Bulk chemical composition	c_M, c_O [at%]	XRF, ICP-OES, EDX
Surface composition	c_M, c_O [at%] surface active site density [nm ⁻²]	XPS
Binding energy, surface oxidation state	B.E. [eV], M^{n+}	XPS
Valence band onset, M nd onset, E secondary electron cutoff, work function, maximum band bending at the surface "s" (surface potential barrier), change in electron affinity	$E_{VB}, E_{nd}, E_{cutoff}, \phi, eV_s, \Delta X$ [eV]	XPS, NAP-XPS, ResPES
Absorption edge energy	O K-Edge, V L ₃ -Edge [eV]	NEXAFS

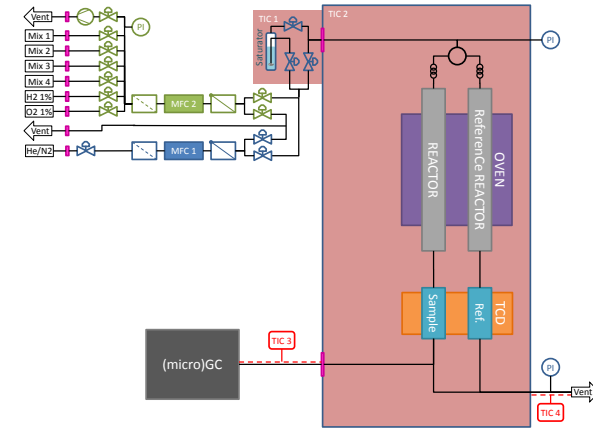
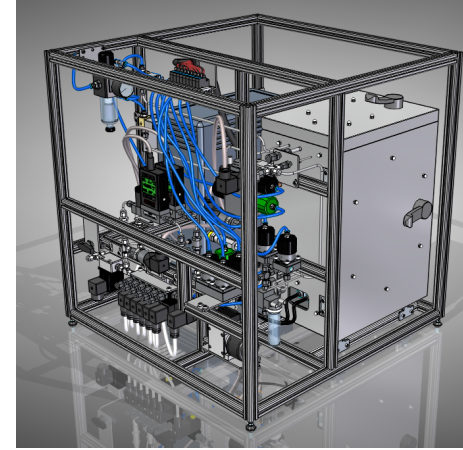
Chemical and physical properties

Property	Physical value	Methods
Specific surface area (BET) Total pore volume, mesopore volume	S_{BET} [m ² /g] V_{total}, V_{mp} [m ³ /g]	Nitrogen adsorption
Heat of adsorption Ads. capacity of reactants (C _n H _{2n+2} , O ₂)	ΔH_{ads} [kJ mol ⁻¹] n [mol/g]	Microcalorimetry
Conductivity, real part of permittivity ϵ' , imaginary part of permittivity ϵ'' , Apparent activation energy of conduction	σ [S/m], ϵ' , ϵ'' , E_c [eV]	MCPT
Optical edge energy	E_{edge} [eV]	UV/Vis
Oxidation/reduction equivalents	Oxygen defect density [V _o ·/nm ²]	TPR/TPD
Desorption temperature	T [K]	TPD
¹⁸ O exchange temperature as function of feed, M-O stretching frequencies	T [K] ν [eV]	Raman
Activity CO oxidation	T_{10}, T_{50} [K]	Fixed bed reactor

Hardware

Pierre Kube

- Kinetic reaction experiments in steady state and pulse operation with temperature-programmed reactions / desorption experiments allows to investigate the very same specimen without the need to transfer the material
- Advanced programming operation of the central unit allows the generation of experiment trains in recipe style probing relevant properties in consecutive order
- Kinetic experiments with pre-programmed parameters are interlaced with activation steps and determination steps of redox/acid-base properties; Fully automated operation allows the execution of long-term experiment series (typically up to 100 hours)



Data management

Peter Kraus

- **Tree based structure**

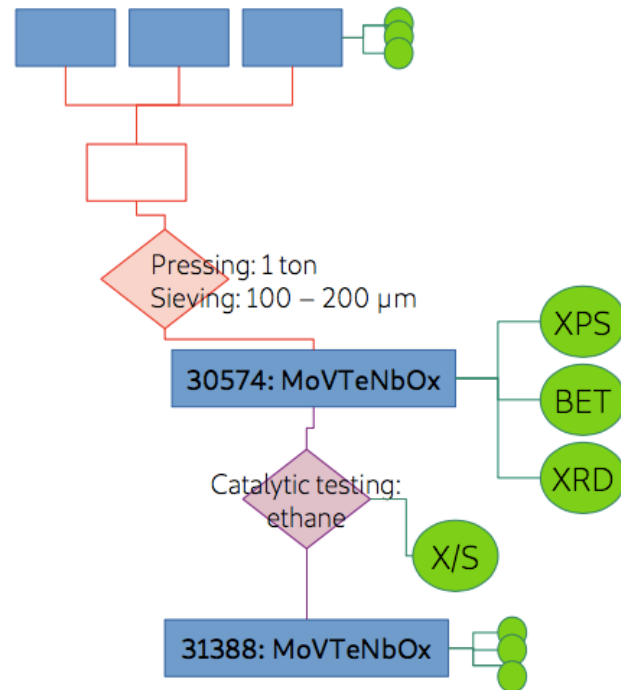
- Sample numbers as nodes
- Treatments as paths

- **Samples**

- **Provenance**: synthesis, pre-treatment, parent species
- **Characterisation**: "standard methods"
- **Performance**: catalytic testing
- **Tags**: "MoVTeNbOx" or "Clean Data"
- Spent samples get a new sample number

- **Treatments**

- Steps in synthesis, pre-treatment
- Processes with attached data: Catalytic testing, BESSY XPS, Conductivity, Microcalorimetry...



- **NoSQL database**

- JSON-based structures for structure and metadata
 - Human- and machine- readable
 - Non-rigid, extensible
- Allows “binary” and textfile attachments
 - CSV files have to be annotated
- MongoDB a good initial choice

- **October: Collect all available data for 30574**

- Estimate overall size of the archive
- First implementation of data structure

```
{
  "id": ,
  "sn": 30574,
  "provenance": {
    "parents": [29345],
    "treatment": [
      {"calcination": {"T": , "unit": "°C"}},
      {"pressing": {"pressure": 1, "unit": "ton"}},
      {"sieving": {"size": [100, 200], "unit": "µm"}},
    ],
    "data": [
      {"id": [...], "format": "Pico Technology logs"}
    ]
  },
  "XRD": {
    "a": {"value": 21.1733, "error": 0.0011, "unit": "Å"},
    "b": {"value": 26.6574, "error": 0.0017, "unit": "Å"},
    "c": {"value": 4.01639, "error": 0.00018, "unit": "Å"},
    "V": {"value": 2266.9, "error": 0.2, "unit": "Å³"},
    "crystal system": "orthorhombic",
    "space group": {"symbol": "Pba2", "number": 32}
    "data": [
      {"id": [...], "format": , "hash": }
    ]
  }
}
```

Conclusion

- Clean experimental data reveal the complexity of describing “the function” of a material
- The handbook and the resulting clean data will serve as input base for discussion in the community
- In addition to metadata storage, some compulsory procedures need to be established and control mechanisms need to be introduced
- Minimum requirements, conditions, descriptive metadata, and benchmark standards will be defined while ensuring the necessary flexibility and freedom of research

