# The FHI-aims All-Electron Framework for Molecular and Materials Simulations 

Volker Blum<br>Department of Mechanical Engineering \& Materials Science - Duke University, Durham, NC http://aims.pratt.duke.edu



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# The FHI-aims All-Electron Framework for Molecular and Materials Simulations 

## Volker Blum <br> Department of Mechanical Engineering \& Materials Science - Duke University, Durham, NC

 http://aims.pratt.duke.eduAll-Electron Theory for Large Systems: FHI-aims


Extending the Reach of DFT \& Many-Body Theory


HSE06, GaAs


Tunable Electronic Structure in Organic-Inorganic Hybrids


## Many Individuals Contributed to This Work - Thanks!



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Victor Yu Electronic Structure Infrastructure (NSF)

Björn Lange Jan Kloppenburg Tiago Botari


FHI-aims team and collaborators: Matthias Scheffler (Berlin), Xinguo Ren (Hefei), over 100 individuals with contributions to the project. Development continues in Berlin, Hefei, Munich, Helsinki, London, Duke, Argonne, and many more.

Dr. Raul Laasner Nuclear Spin States
\& NMR

Tong Zhu PV Materials
GW \& RPA


Garnett Liu
Perovskites
Excitonic Effects

## So, We Have a Theory to Model "Everything"

> Schrödinger (Dirac) Equation
> $\hat{\mathcal{H}} \Psi=E \Psi$

Perfect recipe for parameter-free modeling ...

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Perfect recipe for parameter-free modeling ... ... but how do we make it practical?
P.A.M.

Dirac

The approximation is accurate but
feasible systems are too small

Feasible systems large enough but approximations are not great

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## Schrödinger (Dirac) Equation <br> $\hat{\mathcal{H}} \Psi=E \Psi$

Perfect recipe for parameter-free modeling...

P.A.M. ... but how do we make it practical?

## The approximation is accurate but feasible systems are too small


$\frac{\text { Feasible systems large enough }}{\text { but }}$ approximations are not great
(How?) can we work towards having both?

## Current "Workhorse" Electronic Structure Theory

## Quantum chemistry \& many-body theory:

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E_{\text {tot }} \leq\langle\Psi| \mathrm{H}|\Psi\rangle \quad \text {... successive refinement of } \Psi
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E_{\mathrm{tot}}=E[n(r)]=T_{s}[n]+V[n]+V_{\mathrm{es}}[n]+E_{\mathrm{xc}}[n]
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"Perdew's ladder" to exact solution



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- Key practical approximation: $E_{x c}$ response / many-body terms: RPA, SOSEX, ...
hybrid functionals: non-local exchange meta-GGAs: $\quad \nabla^{2} n(r), \nabla^{2} \varphi(r)$

Generalized gradient approximations (GGAs): $|\nabla n(r)|$ Local-density approximation (LDA): "Perdew's ladder" to exact solution

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Quantum chemistry \& many-body theory:

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E_{\mathrm{tot}}=E[n(r)]=T_{\mathrm{s}}[n]+V[n]+V_{\mathrm{es}}[n]+E_{\mathrm{xc}}[n]
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- Key practical approximation: $E_{x c}$ response / many-body terms: RPA, SOSEX, ... + van der Waals hybrid functionals: non-local exchange Generalized gradient approximations (GGAs): $|\nabla n(r)|$ Local-density approximation (LDA): "Perdew's ladder" to exact solution


## In 2004,We Began a New Electronic Structure Code ...

| Group $\rightarrow$ <br> $\downarrow$ Period | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1 $H$ |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  | $\begin{gathered} 2 \\ \mathrm{He} \end{gathered}$ |
| 2 | 3 Li | 4 Be |  |  |  |  |  |  |  |  |  |  | 5 $B$ | 6 | 7 $N$ | 8 | F | 10 Ne |
| 3 | $\begin{aligned} & 11 \\ & \mathrm{Na} \end{aligned}$ | $\begin{aligned} & 12 \\ & \mathrm{Mg} \end{aligned}$ |  |  |  |  |  |  |  |  |  |  | 13 Al | $\begin{aligned} & 14 \\ & \mathrm{Si} \end{aligned}$ | $\begin{gathered} 15 \\ P \end{gathered}$ | $\begin{gathered} 16 \\ S \end{gathered}$ | 17 Cl | 18 Ar |
| 4 | $\begin{gathered} 19 \\ \mathrm{~K} \end{gathered}$ | $\begin{aligned} & 20 \\ & \mathrm{Ca} \end{aligned}$ | $\begin{aligned} & 21 \\ & \mathrm{Sc} \end{aligned}$ | $22$ | $\begin{aligned} & 23 \\ & \mathrm{~V} \end{aligned}$ | $24$ | $\begin{aligned} & 25 \\ & \mathrm{Mn} \end{aligned}$ | $\begin{aligned} & 26 \\ & \mathrm{Fe} \end{aligned}$ | $\begin{aligned} & 27 \\ & \text { Co } \end{aligned}$ | $\begin{aligned} & 28 \\ & \mathrm{Ni} \end{aligned}$ | $\begin{aligned} & 29 \\ & \mathrm{Cu} \end{aligned}$ | $\begin{aligned} & 30 \\ & \mathrm{Zn} \end{aligned}$ | $\begin{aligned} & 31 \\ & \mathrm{Ga} \end{aligned}$ | $\begin{aligned} & 32 \\ & \mathrm{Ge} \end{aligned}$ | $\begin{aligned} & 33 \\ & \text { As } \end{aligned}$ | $\begin{aligned} & 34 \\ & \mathrm{Se} \end{aligned}$ | 35 Br | 36 Kr |
| 5 | $\begin{aligned} & 37 \\ & \mathrm{Rb} \end{aligned}$ | $\begin{aligned} & 38 \\ & \mathrm{Sr} \end{aligned}$ | $\begin{gathered} 39 \\ Y \end{gathered}$ | $\begin{aligned} & 40 \\ & \mathrm{Zr} \end{aligned}$ | $\begin{aligned} & 41 \\ & \mathrm{Nb} \end{aligned}$ | $\begin{aligned} & 42 \\ & \mathrm{Mo} \end{aligned}$ | $\begin{aligned} & 43 \\ & \mathrm{Tc} \end{aligned}$ | $\begin{aligned} & 44 \\ & \mathrm{Ru} \end{aligned}$ | $\begin{aligned} & 45 \\ & \mathrm{Rh} \end{aligned}$ | $\begin{aligned} & 46 \\ & \mathrm{Pd} \end{aligned}$ | $\begin{aligned} & 47 \\ & \mathrm{Ag} \end{aligned}$ | $\begin{aligned} & 48 \\ & \mathrm{Cd} \end{aligned}$ | $\begin{aligned} & 49 \\ & \text { In } \end{aligned}$ | $\begin{aligned} & 50 \\ & 5 n \end{aligned}$ | $\begin{aligned} & 51 \\ & \mathrm{Sb} \end{aligned}$ | $\begin{aligned} & 52 \\ & \mathrm{Te} \end{aligned}$ | 53 1 | 54 $\times \mathrm{e}$ |
| 6 | $\begin{aligned} & 55 \\ & \mathrm{Cs} \end{aligned}$ | $\begin{aligned} & 56 \\ & \mathrm{Ba} \end{aligned}$ |  | $\begin{aligned} & 72 \\ & \mathrm{Hf} \end{aligned}$ | $\begin{aligned} & 73 \\ & \mathrm{Ta} \end{aligned}$ | $\begin{aligned} & \hline 74 \\ & \mathrm{w} \end{aligned}$ | $\begin{aligned} & 75 \\ & \mathrm{Re} \end{aligned}$ | $\begin{aligned} & 76 \\ & \text { Os } \end{aligned}$ | $\begin{aligned} & \hline 77 \\ & \text { Ir } \end{aligned}$ | $\begin{aligned} & \hline 78 \\ & \text { Pt } \end{aligned}$ | $\begin{aligned} & 79 \\ & \mathrm{Au} \end{aligned}$ | $\begin{aligned} & \hline 80 \\ & \mathrm{Hg} \end{aligned}$ | $\begin{gathered} 81 \\ \mathrm{TI} \end{gathered}$ | $\begin{aligned} & 82 \\ & \mathrm{~Pb} \end{aligned}$ | $\begin{aligned} & 83 \\ & \mathrm{Bi} \end{aligned}$ | $\begin{aligned} & 84 \\ & \mathrm{Po} \end{aligned}$ | 85 At | 86 Rn |
| 7 | $\begin{aligned} & 87 \\ & \mathrm{Fr} \end{aligned}$ | $\begin{aligned} & 88 \\ & \mathrm{Ra} \end{aligned}$ |  | $\begin{gathered} 104 \\ \mathrm{Rf} \end{gathered}$ | $\begin{gathered} 105 \\ \mathrm{Db} \end{gathered}$ | $\begin{gathered} 106 \\ \mathrm{Sg} \end{gathered}$ | $\begin{gathered} 107 \\ \mathrm{Bh} \end{gathered}$ | $\begin{gathered} 108 \\ \mathrm{Hs} \end{gathered}$ | $\begin{gathered} 109 \\ \mathrm{Mt} \end{gathered}$ | $\begin{gathered} 110 \\ \text { Ds } \end{gathered}$ | $\begin{gathered} \hline 111 \\ \mathrm{Rg} \end{gathered}$ | $\begin{gathered} 112 \\ \mathrm{Cn} \end{gathered}$ | $\begin{aligned} & 113 \\ & \text { Uut } \end{aligned}$ | $\begin{gathered} 114 \\ \mathrm{FI} \end{gathered}$ | $\begin{array}{\|l} \hline 115 \\ \text { Uup } \\ \hline \end{array}$ | $\begin{gathered} 116 \\ \mathrm{Lv} \end{gathered}$ | $\begin{aligned} & 117 \\ & \text { Uus } \end{aligned}$ | $\begin{aligned} & 118 \\ & \text { Uuo } \end{aligned}$ |
|  |  | a |  | $\begin{aligned} & 57 \\ & \text { La } \end{aligned}$ | $\begin{aligned} & 58 \\ & \mathrm{Ce} \end{aligned}$ | $\begin{aligned} & 59 \\ & \mathrm{Pr} \end{aligned}$ | $\begin{aligned} & 60 \\ & \mathrm{Nd} \end{aligned}$ | $\begin{aligned} & \hline 61 \\ & \text { Pm } \end{aligned}$ | $\begin{aligned} & 62 \\ & \mathrm{Sm} \end{aligned}$ | $\begin{aligned} & 63 \\ & \mathrm{Eu} \end{aligned}$ | $\begin{aligned} & 64 \\ & \text { Gd } \end{aligned}$ | $\begin{aligned} & 65 \\ & \mathrm{~Tb} \end{aligned}$ | $\begin{aligned} & 66 \\ & \text { Dy } \end{aligned}$ | $\begin{aligned} & 67 \\ & \mathrm{Ho} \end{aligned}$ | $\begin{aligned} & 68 \\ & \mathrm{Er} \end{aligned}$ | $\begin{aligned} & 69 \\ & \mathrm{Tm} \end{aligned}$ | 70 Yb | 71 Lu |
|  |  | Acti |  | $\begin{aligned} & 89 \\ & \text { AC } \end{aligned}$ | $\begin{aligned} & 90 \\ & \text { Th } \end{aligned}$ | $\begin{aligned} & \hline 91 \\ & \mathrm{~Pa} \end{aligned}$ | $\begin{gathered} 92 \\ U \end{gathered}$ | $\begin{aligned} & 93 \\ & \mathrm{~Np} \end{aligned}$ | $\begin{aligned} & 94 \\ & \mathrm{Pu} \end{aligned}$ | $\begin{gathered} 95 \\ \text { Am } \end{gathered}$ | $\begin{aligned} & 96 \\ & \mathrm{Cm} \end{aligned}$ | $\begin{aligned} & \hline 97 \\ & \text { Bk } \end{aligned}$ | $\begin{aligned} & 98 \\ & \text { Cf } \end{aligned}$ | $\begin{aligned} & 99 \\ & \text { Es } \end{aligned}$ | $\begin{gathered} 100 \\ \mathrm{Fm} \end{gathered}$ | $\begin{aligned} & 101 \\ & \mathrm{Md} \end{aligned}$ | 102 No | 103 Lr |

Universality: Cover full space of materials and chemistry

## In 2004,We Began a New Electronic Structure Code ...

| Group $\rightarrow$ $\downarrow$ Period | 1 | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 | 14 | 15 | 16 | 17 | 18 |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1 $H$ |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  | 2 He |
| 2 | 3 Li | 4 Be |  |  |  |  |  |  |  |  |  |  | 5 | 6 | 7 N | 8 | F | 10 Ne |
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| 4 | $\begin{gathered} 19 \\ K \end{gathered}$ | $\begin{aligned} & 20 \\ & \mathrm{Ca} \end{aligned}$ | $\begin{aligned} & 21 \\ & \mathrm{Sc} \end{aligned}$ | $\begin{gathered} 22 \\ \mathrm{Ti} \end{gathered}$ | $\begin{aligned} & 23 \\ & \mathrm{~V} \end{aligned}$ | $\begin{aligned} & 24 \\ & \mathrm{Cr} \end{aligned}$ | $\begin{aligned} & 25 \\ & \mathrm{Mn} \end{aligned}$ | $\begin{aligned} & 26 \\ & \mathrm{Fe} \end{aligned}$ | $\begin{aligned} & 27 \\ & \text { Co } \end{aligned}$ | $\begin{aligned} & 28 \\ & \mathrm{Ni} \end{aligned}$ | $\begin{aligned} & 29 \\ & \mathrm{Cu} \end{aligned}$ | $\begin{aligned} & 30 \\ & \mathrm{Zn} \end{aligned}$ | $31$ | $\begin{aligned} & 32 \\ & \mathrm{Ge} \end{aligned}$ | $\begin{aligned} & 33 \\ & \text { As } \end{aligned}$ | $\begin{aligned} & 34 \\ & \mathrm{Se} \end{aligned}$ | $\begin{aligned} & 35 \\ & \mathrm{Br} \end{aligned}$ | 36 Kr |
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Universality: Cover full space of materials and chemistry "Materials and molecules" - periodic (k-space) and non-periodic

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| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 1 $H$ |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| 2 | 3 Li | $\begin{gathered} 4 \\ \mathrm{Be} \end{gathered}$ |  |  |  |  |  |  |  |  |  |  | 5 | 6 | 7 N | 8 0 | F | 10 Ne |
| 3 | $\begin{aligned} & 11 \\ & \mathrm{Na} \end{aligned}$ | $\begin{aligned} & 12 \\ & \mathrm{Mg} \end{aligned}$ |  |  |  |  |  |  |  |  |  |  | $\begin{aligned} & \hline 13 \\ & \mathrm{Al} \end{aligned}$ | $\begin{aligned} & 14 \\ & \mathrm{Si} \end{aligned}$ | $\begin{gathered} \hline 15 \\ P \end{gathered}$ | $\begin{gathered} 16 \\ 5 \end{gathered}$ | 17 Cl | 18 Ar |
| 4 | $\begin{gathered} 19 \\ K \end{gathered}$ | $\begin{aligned} & 20 \\ & \mathrm{Ca} \end{aligned}$ | $\begin{aligned} & 21 \\ & \mathrm{Sc} \end{aligned}$ | $\begin{gathered} 22 \\ \mathrm{Ti} \end{gathered}$ | $\begin{aligned} & 23 \\ & \mathrm{~V} \end{aligned}$ | $\begin{aligned} & 24 \\ & \mathrm{Cr} \end{aligned}$ | $\begin{aligned} & 25 \\ & \mathrm{Mn} \end{aligned}$ | $\begin{aligned} & 26 \\ & \mathrm{Fe} \end{aligned}$ | $\begin{aligned} & 27 \\ & \text { Co } \end{aligned}$ | $\begin{aligned} & 28 \\ & \mathrm{Ni} \end{aligned}$ | $\begin{aligned} & 29 \\ & \mathrm{Cu} \end{aligned}$ | $\begin{aligned} & 30 \\ & \mathrm{Zn} \end{aligned}$ | $\begin{aligned} & 31 \\ & \mathrm{Ga} \end{aligned}$ | $\begin{aligned} & 32 \\ & \mathrm{Ge} \end{aligned}$ | $\begin{aligned} & 33 \\ & \text { As } \end{aligned}$ | $\begin{aligned} & 34 \\ & \mathrm{Se} \end{aligned}$ | 35 Br | 36 Kr |
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| 7 | $\begin{aligned} & \hline 87 \\ & \mathrm{Fr} \end{aligned}$ | $\begin{aligned} & \hline 88 \\ & \text { Ra } \end{aligned}$ |  | $\begin{gathered} \hline 104 \\ \mathrm{Rf} \end{gathered}$ | $\begin{gathered} 105 \\ \mathrm{Db} \end{gathered}$ | $\begin{gathered} \hline 106 \\ \mathrm{Sg} \end{gathered}$ | $\begin{gathered} \hline 107 \\ \text { Bh } \\ \hline \end{gathered}$ | $\begin{gathered} \hline 108 \\ \mathrm{Hs} \end{gathered}$ | $\begin{gathered} \hline 109 \\ \mathrm{Mt} \end{gathered}$ | $\begin{gathered} 110 \\ \text { Ds } \end{gathered}$ | $\begin{gathered} \hline 111 \\ \mathrm{Rg} \end{gathered}$ | $\begin{gathered} 112 \\ C n \end{gathered}$ | $\begin{aligned} & 113 \\ & \text { Uut } \end{aligned}$ | $\begin{array}{\|c} \hline 114 \\ \mathrm{FI} \end{array}$ | $\begin{aligned} & 115 \\ & \text { Uup } \end{aligned}$ | $\begin{gathered} 116 \\ \mathrm{Lv} \end{gathered}$ | $\begin{aligned} & \hline 117 \\ & \text { Uus } \end{aligned}$ | $\begin{aligned} & \hline 118 \\ & \text { Uuo } \end{aligned}$ |
| Lanthanides |  |  |  | $\begin{aligned} & 57 \\ & \mathrm{La} \end{aligned}$ | $\begin{aligned} & 58 \\ & \mathrm{Ce} \end{aligned}$ | $\begin{aligned} & \hline 59 \\ & \mathrm{Pr} \\ & \hline \end{aligned}$ | $\begin{aligned} & 60 \\ & \mathrm{Nd} \end{aligned}$ | $\begin{array}{\|c\|} \hline 61 \\ \mathrm{Pm} \\ \hline \end{array}$ | $\begin{aligned} & 62 \\ & \mathrm{Sm} \end{aligned}$ | $\begin{aligned} & 63 \\ & \mathrm{Eu} \end{aligned}$ | $\begin{aligned} & 64 \\ & \mathrm{Gd} \end{aligned}$ | $\begin{aligned} & 65 \\ & \mathrm{~Tb} \end{aligned}$ | $\begin{aligned} & 66 \\ & \text { Dy } \\ & \hline \end{aligned}$ | $\begin{aligned} & 67 \\ & \mathrm{Ho} \end{aligned}$ | $\begin{aligned} & 68 \\ & \mathrm{Er} \end{aligned}$ | $\begin{array}{\|c\|} \hline 69 \\ \mathrm{Tm} \\ \hline \end{array}$ | 70 Yb | 71 <br> Lu |
| Actinides |  |  |  | $\begin{aligned} & 89 \\ & \mathrm{Ac} \end{aligned}$ | $\begin{aligned} & 90 \\ & \text { Th } \end{aligned}$ | $\begin{aligned} & 91 \\ & \mathrm{~Pa} \end{aligned}$ | $\begin{gathered} 92 \\ u \end{gathered}$ | $\begin{aligned} & 93 \\ & \mathrm{~Np} \end{aligned}$ | $\begin{aligned} & 94 \\ & \mathrm{Pu} \end{aligned}$ | $\begin{gathered} 95 \\ \mathrm{Am} \end{gathered}$ | $\begin{aligned} & 96 \\ & \mathrm{Cm} \end{aligned}$ | $\begin{aligned} & 97 \\ & \mathrm{Bk} \end{aligned}$ | $\begin{aligned} & 98 \\ & \mathrm{Cf} \end{aligned}$ | $\begin{aligned} & 99 \\ & \text { Es } \end{aligned}$ | $\begin{gathered} 100 \\ \mathrm{Fm} \end{gathered}$ | $\begin{aligned} & 101 \\ & \mathrm{Md} \end{aligned}$ | $\begin{gathered} 102 \\ \text { No } \end{gathered}$ | $\begin{gathered} 103 \\ \text { Lr } \end{gathered}$ |

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| Group $\rightarrow$ <br> $\downarrow$ Period <br> 1 |  | 2 | 3 | 4 | 5 | 6 | 7 | 8 | 9 | 10 | 11 | 12 | 13 | 14 | 15 | 16 | 17 | $\begin{gathered} 18 \\ \hline 2 \\ \mathrm{He} \end{gathered}$ |
| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
|  | 1 $H$ |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| 2 | 3 Li | $\begin{gathered} 4 \\ \mathrm{Be} \end{gathered}$ |  |  |  |  |  |  |  |  |  |  | 5 | 6 | 7 N | 8 0 | F | 10 Ne |
| 3 | $\begin{aligned} & 11 \\ & \mathrm{Na} \end{aligned}$ | $\begin{aligned} & 12 \\ & \mathrm{Mg} \end{aligned}$ |  |  |  |  |  |  |  |  |  |  | $\begin{aligned} & \hline 13 \\ & \mathrm{Al} \end{aligned}$ | $\begin{aligned} & 14 \\ & \mathrm{Si} \end{aligned}$ | $\begin{gathered} \hline 15 \\ P \end{gathered}$ | $\begin{gathered} 16 \\ 5 \end{gathered}$ | 17 Cl | 18 Ar |
| 4 | $\begin{gathered} 19 \\ K \end{gathered}$ | $\begin{aligned} & 20 \\ & \mathrm{Ca} \end{aligned}$ | $\begin{aligned} & 21 \\ & \mathrm{Sc} \end{aligned}$ | $\begin{gathered} 22 \\ \mathrm{Ti} \end{gathered}$ | $\begin{aligned} & 23 \\ & \mathrm{~V} \end{aligned}$ | $\begin{aligned} & 24 \\ & \mathrm{Cr} \end{aligned}$ | $\begin{aligned} & 25 \\ & \mathrm{Mn} \end{aligned}$ | $\begin{aligned} & 26 \\ & \mathrm{Fe} \end{aligned}$ | $\begin{aligned} & 27 \\ & \text { Co } \end{aligned}$ | $\begin{aligned} & 28 \\ & \mathrm{Ni} \end{aligned}$ | $\begin{aligned} & 29 \\ & \mathrm{Cu} \end{aligned}$ | $\begin{aligned} & 30 \\ & \mathrm{Zn} \end{aligned}$ | $\begin{aligned} & 31 \\ & \mathrm{Ga} \end{aligned}$ | $\begin{aligned} & 32 \\ & \mathrm{Ge} \end{aligned}$ | $\begin{aligned} & 33 \\ & \text { As } \end{aligned}$ | $\begin{aligned} & 34 \\ & \mathrm{Se} \end{aligned}$ | 35 Br | 36 Kr |
| 5 | $\begin{aligned} & 37 \\ & \mathrm{Rb} \end{aligned}$ | $\begin{aligned} & 38 \\ & \mathrm{Sr} \end{aligned}$ | $\begin{gathered} 39 \\ Y \\ \hline \end{gathered}$ | $\begin{aligned} & \hline 40 \\ & \mathrm{Zr} \\ & \hline \end{aligned}$ | $\begin{aligned} & 41 \\ & \mathrm{Nb} \end{aligned}$ | $\begin{aligned} & 42 \\ & \mathrm{Mo} \end{aligned}$ | $\begin{aligned} & \hline 43 \\ & \mathrm{Tc} \end{aligned}$ | $\begin{aligned} & 44 \\ & \mathrm{Ru} \\ & \hline \end{aligned}$ | $\begin{aligned} & 45 \\ & \mathrm{Rh} \end{aligned}$ | $\begin{aligned} & 46 \\ & \mathrm{Pd} \\ & \hline \end{aligned}$ | $\begin{aligned} & 47 \\ & \mathrm{Ag} \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 48 \\ & \mathrm{Cd} \\ & \hline \end{aligned}$ | $\begin{aligned} & 49 \\ & \text { In } \end{aligned}$ | $\begin{aligned} & 50 \\ & \mathrm{Sn} \\ & \hline \end{aligned}$ | $\begin{aligned} & \hline 51 \\ & \mathrm{Sb} \end{aligned}$ | $\begin{aligned} & 52 \\ & \mathrm{Te} \end{aligned}$ | $\begin{gathered} 53 \\ 1 \\ \hline \end{gathered}$ | $\begin{aligned} & 54 \\ & \mathrm{Xe} \end{aligned}$ |
| 6 | $\begin{aligned} & 55 \\ & \mathrm{Cs} \end{aligned}$ | $\begin{aligned} & 56 \\ & \mathrm{Ba} \end{aligned}$ |  | $\begin{aligned} & 72 \\ & \mathrm{Hf} \end{aligned}$ | $\begin{aligned} & 73 \\ & \mathrm{Ta} \end{aligned}$ | $\begin{aligned} & 74 \\ & \mathrm{w} \end{aligned}$ | $\begin{aligned} & 75 \\ & \mathrm{Re} \end{aligned}$ | $\begin{aligned} & 76 \\ & \text { Os } \end{aligned}$ | $\begin{aligned} & 77 \\ & \text { Ir } \end{aligned}$ | $\begin{aligned} & 78 \\ & \text { Pt } \end{aligned}$ | $\begin{aligned} & 79 \\ & \mathrm{Au} \end{aligned}$ | $\begin{aligned} & 80 \\ & \mathrm{Hg} \end{aligned}$ | $\begin{aligned} & 81 \\ & \mathrm{TI} \end{aligned}$ | $\begin{aligned} & 82 \\ & \mathrm{~Pb} \end{aligned}$ | $\begin{aligned} & 83 \\ & \mathrm{Bi} \end{aligned}$ | $\begin{aligned} & 84 \\ & \text { Po } \end{aligned}$ | $\begin{aligned} & 85 \\ & \text { At } \end{aligned}$ | 86 Rn |
| 7 | $\begin{aligned} & \hline 87 \\ & \mathrm{Fr} \end{aligned}$ | $\begin{aligned} & \hline 88 \\ & \text { Ra } \end{aligned}$ |  | $\begin{gathered} \hline 104 \\ \mathrm{Rf} \end{gathered}$ | $\begin{gathered} 105 \\ \mathrm{Db} \end{gathered}$ | $\begin{gathered} \hline 106 \\ \mathrm{Sg} \end{gathered}$ | $\begin{gathered} \hline 107 \\ \text { Bh } \\ \hline \end{gathered}$ | $\begin{gathered} \hline 108 \\ \mathrm{Hs} \end{gathered}$ | $\begin{gathered} \hline 109 \\ \mathrm{Mt} \end{gathered}$ | $\begin{gathered} 110 \\ \text { Ds } \end{gathered}$ | $\begin{gathered} \hline 111 \\ \mathrm{Rg} \end{gathered}$ | $\begin{gathered} 112 \\ C n \end{gathered}$ | $\begin{aligned} & 113 \\ & \text { Uut } \end{aligned}$ | $\begin{array}{\|c} \hline 114 \\ \mathrm{FI} \end{array}$ | $\begin{aligned} & 115 \\ & \text { Uup } \end{aligned}$ | $\begin{gathered} 116 \\ \mathrm{Lv} \end{gathered}$ | $\begin{aligned} & \hline 117 \\ & \text { Uus } \end{aligned}$ | $\begin{aligned} & \hline 118 \\ & \text { Uuo } \end{aligned}$ |
| Lanthanides |  |  |  | $\begin{aligned} & 57 \\ & \mathrm{La} \end{aligned}$ | $\begin{aligned} & 58 \\ & \mathrm{Ce} \end{aligned}$ | $\begin{aligned} & \hline 59 \\ & \mathrm{Pr} \\ & \hline \end{aligned}$ | $\begin{aligned} & 60 \\ & \mathrm{Nd} \end{aligned}$ | $\begin{array}{\|c\|} \hline 61 \\ \mathrm{Pm} \\ \hline \end{array}$ | $\begin{aligned} & 62 \\ & \mathrm{Sm} \end{aligned}$ | $\begin{aligned} & 63 \\ & \mathrm{Eu} \end{aligned}$ | $\begin{aligned} & 64 \\ & \mathrm{Gd} \end{aligned}$ | $\begin{aligned} & 65 \\ & \mathrm{~Tb} \end{aligned}$ | $\begin{aligned} & 66 \\ & \text { Dy } \\ & \hline \end{aligned}$ | $\begin{aligned} & 67 \\ & \mathrm{Ho} \end{aligned}$ | $\begin{aligned} & 68 \\ & \mathrm{Er} \end{aligned}$ | $\begin{array}{\|c\|} \hline 69 \\ \mathrm{Tm} \\ \hline \end{array}$ | 70 Yb | 71 <br> Lu |
| Actinides |  |  |  | $\begin{aligned} & 89 \\ & \mathrm{Ac} \end{aligned}$ | $\begin{aligned} & 90 \\ & \text { Th } \end{aligned}$ | $\begin{aligned} & 91 \\ & \mathrm{~Pa} \end{aligned}$ | $\begin{gathered} 92 \\ u \end{gathered}$ | $\begin{aligned} & 93 \\ & \mathrm{~Np} \end{aligned}$ | $\begin{aligned} & 94 \\ & \mathrm{Pu} \end{aligned}$ | $\begin{gathered} 95 \\ \mathrm{Am} \end{gathered}$ | $\begin{aligned} & 96 \\ & \mathrm{Cm} \end{aligned}$ | $\begin{aligned} & 97 \\ & \mathrm{Bk} \end{aligned}$ | $\begin{aligned} & 98 \\ & \mathrm{Cf} \end{aligned}$ | $\begin{aligned} & 99 \\ & \text { Es } \end{aligned}$ | $\begin{gathered} 100 \\ \mathrm{Fm} \end{gathered}$ | $\begin{aligned} & 101 \\ & \mathrm{Md} \end{aligned}$ | $\begin{gathered} 102 \\ \text { No } \end{gathered}$ | $\begin{gathered} 103 \\ \text { Lr } \end{gathered}$ |

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| :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: | :---: |
| 1 | 1 $H$ |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |  |
| 2 | 3 Li | 4 Be |  |  |  |  |  |  |  |  |  |  | 5 | 6 | 7 $N$ | 8 | F | 10 Ne |
| 3 | $\begin{aligned} & 11 \\ & \mathrm{Na} \end{aligned}$ | $\begin{aligned} & 12 \\ & \mathrm{Mg} \end{aligned}$ |  |  |  |  |  |  |  |  |  |  | $\begin{aligned} & 13 \\ & \mathrm{Al} \end{aligned}$ | $\begin{aligned} & 14 \\ & \mathrm{Si} \end{aligned}$ | $\begin{gathered} 15 \\ P \end{gathered}$ | $\begin{gathered} 16 \\ S \end{gathered}$ | $\begin{aligned} & \hline 17 \\ & \mathrm{Cl} \end{aligned}$ | 18 Ar |
| 4 | $\begin{gathered} 19 \\ \mathrm{~K} \end{gathered}$ | $\begin{aligned} & 20 \\ & \mathrm{Ca} \end{aligned}$ | $\begin{aligned} & 21 \\ & \mathrm{Sc} \end{aligned}$ | $22$ | $\begin{aligned} & 23 \\ & \mathrm{~V} \end{aligned}$ | $\begin{aligned} & 24 \\ & \mathrm{Cr} \end{aligned}$ | $\begin{aligned} & 25 \\ & \mathrm{Mn} \end{aligned}$ | $\begin{aligned} & 26 \\ & \mathrm{Fe} \end{aligned}$ | $\begin{aligned} & 27 \\ & \text { Co } \end{aligned}$ | $\begin{aligned} & 28 \\ & \mathrm{Ni} \end{aligned}$ | $\begin{aligned} & 29 \\ & \mathrm{Cu} \end{aligned}$ | $\begin{aligned} & 30 \\ & \mathrm{Zn} \end{aligned}$ | $\begin{aligned} & 31 \\ & \mathrm{Ga} \end{aligned}$ | $\begin{aligned} & 32 \\ & \mathrm{Ge} \end{aligned}$ | $\begin{aligned} & 33 \\ & \text { As } \end{aligned}$ | $\begin{aligned} & 34 \\ & \mathrm{Se} \end{aligned}$ | $\begin{aligned} & 35 \\ & \mathrm{Br} \end{aligned}$ | $\begin{aligned} & 36 \\ & \mathrm{Kr} \end{aligned}$ |
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## Efficient (I,000s of atoms), but do not sacrifice accuracy!

## Central Decision: How to Discretize the Problem?

$$
\left[-\frac{\nabla^{2}}{2}+v_{\mathrm{ext}}(\boldsymbol{r})+v_{\mathrm{es}}(\boldsymbol{r})+v_{\mathrm{xc}}(\boldsymbol{r})\right] \psi_{k}(\boldsymbol{r})=\epsilon_{k} \psi_{k}(\boldsymbol{r})
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"Basis set expansion":
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Kohn, Sham 1965
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Generalized eigenvalue problem:

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$$
\begin{gathered}
h_{i j}=\left\langle\varphi_{i}\right| \hat{h}_{\mathrm{KS}}\left|\varphi_{j}\right\rangle \\
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Many good options:

- Plane waves (VASP, abinit, ...)

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$$

- Many others:"Augmented plane waves", wavelets, finite elements, numeric atom-centered functions, ...
Our choice!

$$
\varphi_{i[l m]}(\boldsymbol{r})=\frac{u_{i}(r)}{r} \cdot Y_{l m}(\Omega)
$$

## Numeric Atom-Centered Basis Functions: Some Advantages

$$
\varphi_{i[l m]}(\boldsymbol{r})=\frac{u_{i}(r)}{r} \cdot Y_{l m}(\Omega)
$$



Many popular implementations:
DMol ${ }^{3}$ (Delley), FPLO (Eschrig et al.) PLATO (Horsfield et al.), ADF, BDF, PAOs (Siesta, Conquest, OpenMX², Fireball, ...)

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$$
\varphi_{i[l m]}(\boldsymbol{r})=\frac{u_{i}(r)}{r} \cdot Y_{l m}(\Omega)
$$



- Flexible shape
- "Naturally" all-electron
- Strictly localizable - $O(N)$ computational scaling for most expensive steps
- Rather compact basis sets for converged results

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## Accuracy in Community Wide Benchmark - "Delta Test"

S. Cottenier and coworkers (Ghent University): https://molmod.ugent.be/deltacodesdft
$E(V)$ for 7 I elemental solids - Reference: Full-Potential LAPW (Wien2k).

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## Similar High Precision Achievable for Band Structures

## Root-mean-square deviation between calculated band structures: <br> FHI-aims (tier 2) vs.Wien2k, scalar relativistic, valence bands, DFT-PBE

Dr.William Huhn (Duke Univ.)




0

## All-Electron Accuracy for Large Systems: FHI-aims

- Accurate, efficient quantum mechanics for molecules, materials
- Numerical Foundation: Numeric atom-centered basis functions Seamless from "light" to basis-converged results, from light to heavy elements
- Non-periodic and periodic structure models, same framework
- Density Functional Theory (semilocal, global and rangeseparated hybrid functionals, van der Waals corrections)
- Beyond DFT (non-periodic): RPA, GW, MP2,TD-LDA, ...
- Scalable (I,000s of atoms on I(00),000s of CPU cores)
- Properties (charged and neutral excitations, IR spectra, transport, dielectric functions, dynamics \& transition states, ...)
~ 100 contributors to date, over 100 licensing groups. Active development in Berlin, Duke, Munich, Hefei, Helsinki, London, Argonne, ...
New stable release " 160328 "
Blum, Gehrke, Hanke, Havu, Havu, Ren, Reuter, Scheffler, Computer Physics Communications I80, 2I75(2009) Ren, Rinke, Blum, Wieferink, Tkatchenko, Sanfilippo, Reuter, Scheffler, New J. Phys. I4, 053020 (2012)
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## Scalability: 3-Layer Graphene on 3C-SiC(000I)



Structure: Nemec et al., PRL I I I, 065502 (20I3).
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## Parallel Eigenvalue Solvers - the Problem



IBM BlueGene (MPG, Garching) 16384 CPU cores

## Parallel Eigenvalue Solvers - the Problem


$\alpha$-helical Ala। 100 (1000 atoms), high accuracy

IBM BlueGene (MPG, Garching) 16384 CPU cores

## Parallel Eigenvalue Solvers - the Problem



## Bottleneck:Tridiagonalization

"Conventional" reduction:


## Key Step in ELPA:Two-Step MPI-Parallel Tridiagonalization

Two-stage reduction algorithm, ELPA:


Larger fraction of efficient matrix-matrix operations Efficient compute kernels for added backtransform step (4)

Note: Hard to beat dense linear algebra for small to midsized problems, many EVs

Auckenthaler, Blum, Bungartz, Huckle, Johanni, Krämer, Lang, Lederer, Willems, Parallel Computing 37, 783 (20I I)
A. Marek, V. Blum, R. Johanni, V. Havu, B. Lang,T.Auckenthaler, A. Heinecke, H.-J. Bungartz, H. Lederer, The Journal of Physics: Condensed Matter 26, 21320 I (2014).

## ELPA, Two-Step Solver




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4) Pole Expansion and Selective Inversion (PEXSI)

Reduced-scaling approach focused on density matrix, at most $O\left(N^{2}\right)$ for 3D semilocal DFT. Lin, Lu, Ying, Car, E, Commun. Math. Sci. 7, 755 (2009); Lin, Chen, Yang, He, J. Phys. Condens. Matter 25, 29550 (2013).
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## ELSI: Solving or Circumventing the Eigenvalue Problem

... in one infrastructure
\& for "any" code?
... and many other "stakeholders" from the community

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Work in Progress:
"ELSI" - Electronic Structure Infrastructure (NSF-SI2)


HPC platform optimization (distrib. SMP, GPU, manycore)
http://elsi-interchange.org
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## Beyond DFT-LDA/GGA:Two-Electron Integrals

## Common bottleneck:Two-electron interactions, e.g.:

$$
\int d^{3} r d^{3} r^{\prime} \phi_{i}(r) \phi_{j}\left(r^{\prime}\right) \frac{1}{\left|r-r^{\prime}\right|} \phi_{k}\left(r^{\prime}\right) \phi_{l}(r)
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$$

## Early Solution:"Resolution of the identity"

[Boys and Shavitt (1950s), Whitten (1974), Dunlap et al. (I979),Vahtras et al. (1993), many others]

1. Expand pair products in smaller auxiliary basis set $\left\{P_{\mu}\right\}$
$\varphi_{i}(\boldsymbol{r}) \varphi_{j}(\boldsymbol{r})=\sum_{\mu} C_{i j}^{\mu} P_{\mu}(\boldsymbol{r})$
2. Use "Coulomb metric" $V$ to cancel linear error terms

$$
\begin{aligned}
& C_{i j}^{\mu}=\sum_{\nu}(i j \mid \nu) V_{\nu \mu}^{-1} \\
& (i j \mid \nu)=\int d^{3} r d^{3} r^{\prime} \frac{\varphi_{i}(\boldsymbol{r}) \varphi_{j}(\boldsymbol{r}) P_{\nu}\left(\boldsymbol{r}^{\prime}\right)}{\left|\boldsymbol{r}-\boldsymbol{r}^{\prime}\right|}
\end{aligned}
$$

$\rightarrow$ The rest is matrix algebra - works with NAO basis sets! Hartree-Fock, hybrid functionals, MP2, GW, RPA, LR-TDLDA, ...

Ren, Rinke, Blum, Wieferink, Tkatchenko, Sanfilippo, Reuter, Scheffler, New J. Phys. I4, 053020 (2012)

## Localized "Resolution of Identity" (RI) for Two-Electron Terms

$$
\begin{aligned}
& \qquad(i j \mid k l)=\int d^{3} r d^{3} r^{\prime} \frac{\varphi_{i}(\boldsymbol{r}) \varphi_{j}\left(\boldsymbol{r}^{\prime}\right) \varphi_{k}(\boldsymbol{r}) \varphi_{l}\left(\boldsymbol{r}^{\prime}\right)}{\left|\boldsymbol{r}-\boldsymbol{r}^{\prime}\right|} \\
& \varphi_{i}(\boldsymbol{r}) \varphi_{j}(\boldsymbol{r})=\sum_{\mu} C_{i j}^{\mu} P_{\mu}(\boldsymbol{r}) \\
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& \text { Problem: Full RI-V delocalizes } C \\
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& \begin{array}{l}
\text { Scross entire system } \\
\text { Solution: For each } C_{i j^{\prime}}, \text { restrict } \mu \\
\hline \text { only to atoms } I \text { and } J \text { at which } i \text { and } \\
j \text { are centered! }
\end{array}
\end{aligned}
$$

Ihrig,Wieferink, Zhang, Ropo, Ren, Rinke, Scheffler, Blum, New J. Phys. I7, 093020 (2015)

## Hybrid Functionals: Scalability, Large Systems

Levchenko, Ren, Wieferink, Rinke, Johanni, Blum, Scheffler, Comp. Phys. Commun. I92, 60-69 (2015).
$\mathrm{O}(\mathrm{N})$ scaling implementation, localized resolution of identity.

Note: Localized RI also works for MP2, RPA, GW, etc. (but not $O(N)$ )


Zincblende GaAs

Computational Scaling of Periodic GaAs, HSE06 Hybrid Density-Functional Theory
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## Scaling Limitations for Many-Body Theory: GW

GoW0:Widely used to obtain accurate quasiparticle energies (molecules and materials)

Central: Self-Energy

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However, $G_{0}$, $W_{0}$ not easy to converge -
known by different terminologies in different communities:

- "Slow-converging sums over states"
- "Slow convergence with basis set size"
- "Slow convergence of the electron-electron cusp" (Quantum Chemistry)


## Result: Notoriously Tedious Convergence (Any Basis Set!)

## RPA@PBE Total Energy Convergence of Free Atoms:



Igor Ying Zhang, Xinguo Ren, Patrick Rinke,Volker Blum, and Matthias Scheffler, New Journal of Physics I5, I23033 (20I3).

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RPA@PBE Total Energy Convergence of Free Atoms:


New NAO-VCC-nZ basis sets (H-Ar):
More systematic convergence of unoccupied state sums* *not a problem in occupied-state based DFT

Igor Ying Zhang, Xinguo Ren, Patrick Rinke,Volker Blum, and Matthias Scheffler, New Journal of Physics I5, I23033 (20I3).

## Basis Set Extrapolation for $G_{0} W_{0}$ ?

Example: Naphthalene, Go Wo@PBE HOMO / LUMO


Tong Zhu

## Basis Set Extrapolation for $G_{0} W_{0}$ ?

Example: Naphthalene,GoWo@PBE HOMO / LUMO


NAO-VCC-nZ (NAO) aug-CC-pVnZ (AUG) CC-pVnZ (CC)

## Can We Extrapolate Smaller (Cheaper) Basis Sets?



2Z-3Z Extrapolation (T. Zhu): $\underline{5 Z}$ quality, but at the price of $3 Z$. Works for valence and low-lying conduction levels, light-element molecules. However, not for core states, unbound states; restricted to H-Ar.

## Summary

$$
\hat{\mathcal{H}} \Psi=E \Psi
$$

P.A.M. Dirac

High-accuracy platform for predictive molecular, materials simulations Scalable to large systems, advanced electronic structure approximations

Ongoing frontier:Accurate, affordable many-body perturbation theory towards excited states for real materials


Crystalline tunable organic-inorganic hybrid materials - predictive approach to truly "new" materials, close integration with experiment


ALCF
National Science Foundation

## Constructing a Basis Set Library for DFT

Goal: Element-dependent, transferable basis sets
from fast qualitative to meV-converged total energy accuracy (ground-state DFT)

## Can't we have the computer pick good basis sets for us?

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$$
\mathrm{E}^{(\mathrm{n})}=\mathrm{E}\left[\{u\}^{(\mathrm{n}-\mathrm{I})} \oplus \mathrm{u}_{\text {trial }}\right]
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until $E^{(n-I)}-E^{(n)}<$ threshold

## Iterative Selection of NAO Basis Functions

```
"Pool" of trial basis functions:
    2+ ionic u(r)
Hydrogen-like u(r) for z=0.I-20
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> Optimization target: Non-selfconsistent symmetric dimers, averaged for different d

Pick basis functions one by one, up to complete total energy convergence

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## Result: Hierarchical Basis Set Library for All Elements

|  | H | C | O | Au | Systematic hierarchy of basis (sub)sets, iterative |
| :---: | :---: | :---: | :---: | :---: | :---: |
| minimal | $1 s$ | $[\mathrm{He}]+2 s 2 p$ | $[\mathrm{He}]+2 s 2 p$ | $[\mathrm{Xe}]+6 s 5 d 4 f$ |  |
| Tier 1 | $\mathrm{H}(2 s, 2.1)$ | $\mathrm{H}(2 p, 1.7)$ | H(2p,1.8) | $\mathrm{Au}^{2+}(6 \mathrm{p})$ | automated construction <br> based on dimers |
|  | $\mathrm{H}(2 p, 3.5)$ | $\mathrm{H}(3 d, 6.0)$ | $\mathrm{H}(3 d, 7.6)$ | $\mathrm{H}(4 f, 7.4)$ |  |
|  |  | $\mathrm{H}(2 s, 4.9)$ | H(3s,6.4) | $\mathrm{Au}^{2+}(6 \mathrm{~s})$ \} | "First tier (level)" |
|  |  |  |  | $\mathrm{H}(5 g, 10)$ | 'First tier (level) |
|  |  |  |  | H(6h, 12.8) |  |
|  |  |  |  | $\mathrm{H}(3 d, 2.5) \quad$ ) |  |
| Tier 2 | $\mathrm{H}(1 s, 0.85)$ | $\mathrm{H}(4 f, 9.8)$ | $\mathrm{H}(4 f, 11.6)$ | $\mathrm{H}(5 f, 14.8)$ | "Second tier" |
|  | $\mathrm{H}(2 p, 3.7)$ | $\mathrm{H}(3 p, 5.2)$ | H(3p,6.2) | H(4d,3.9) |  |
|  | $\mathrm{H}(2 s, 1.2)$ | $\mathrm{H}(3 s, 4.3)$ | H(3d,5.6) | $\mathrm{H}(3 p, 3.3)$ |  |
|  | $\mathrm{H}(3 d, 7.0)$ | H( $5 g, 14.4)$ | $\mathrm{H}(5 g, 17.6)$ | $\mathrm{H}(1 s, 0.45)$ |  |
|  |  | $\mathrm{H}(3 d, 6.2)$ | $\mathrm{H}(1 s, 0.75)$ | $\mathrm{H}(5 \mathrm{~g}, 16.4)$ |  |
|  |  |  |  | H(6h,13.6) |  |
| Tier 3 | $\mathrm{H}(4 f, 11.2)$ | H(2p,5.6) | $\mathrm{O}^{2+}(2 p)$ | $\mathrm{H}(4 f, 5.2)^{*} 7$ |  |
|  | $\mathrm{H}(3 p, 4.8)$ | $\mathrm{H}(2 s, 1.4)$ | $\mathrm{H}(4 f, 10.8)$ | $\mathrm{H}(4 d, 5.0)$ |  |
|  | ... | ... | ... | ... | "Third tier" |

## Accuracy: $\left(\mathrm{H}_{2} \mathrm{O}\right)_{2}$ Hydrogen Bond Energy



## Using Numeric Atom-Centered Basis Functions: Pieces

- Numerical Integration

$$
h_{i j}=\int d^{3} r \varphi_{i}(\boldsymbol{r}) \hat{h}_{\mathrm{KS}} \varphi_{\cdot j} .(\boldsymbol{r})
$$

- Electron density update
- All-electron electrostatics

$$
\begin{aligned}
n(\boldsymbol{r}) & =\sum_{k} f_{k}\left|\psi_{k}(\boldsymbol{r})\right|^{2} \\
v_{\mathrm{es}}(\boldsymbol{r}) & =\int d^{3} r^{\prime} \frac{n\left(\boldsymbol{r}^{\prime}\right)}{\left|\boldsymbol{r}-\boldsymbol{r}^{\prime}\right|}
\end{aligned}
$$

- Eigenvalue solver

$$
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## Using Numeric Atom-Centered Basis Functions: Pieces

- Numerical Integration
- Electron density update
- All-electron electrostatics
- Eigenvalue solver
- Relativity?

$$
h_{i j}=\int d^{3} r \varphi_{i}(\boldsymbol{r}) \hat{h}_{\mathrm{KS}} \varphi_{\cdot j} .(\boldsymbol{r})
$$

$$
n(\boldsymbol{r})=\sum_{k} f_{k}\left|\psi_{k}(\boldsymbol{r})\right|^{2}
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$$
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$$

$$
\underline{\underline{h}} \underline{c}_{k}=\epsilon_{k} \underline{\underline{s}} \underline{c}_{k}
$$

needed for heavy elements

- Periodic systems? need suitable basis, electrostatics
- Coulomb operator?

$$
(i j \mid k l)=\int d^{3} r d^{3} r^{\prime} \frac{\varphi_{i}(\boldsymbol{r}) \varphi_{j}\left(\boldsymbol{r}^{\prime}\right) \varphi_{k}(\boldsymbol{r}) \varphi_{l}\left(\boldsymbol{r}^{\prime}\right)}{\left|\boldsymbol{r}-\boldsymbol{r}^{\prime}\right|}
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## Numeric Atom-Centered Basis Functions: Integration

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- Discretize to integration grid: $\quad \int d^{3} r f(\boldsymbol{r}) \rightarrow \sum_{\boldsymbol{r}} w(\boldsymbol{r}) f(\boldsymbol{r})$
... but even-spaced integration grids are out:
$f(r)$ strongly peaked near all nuclei!


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but even-spaced integration grids are out: $f(r)$ strongly peaked near all nuclei!
- Overlapping atom-centered integration grids:
- Radial shells (e.g., H, light: 24;Au, tight: I47)
- Specific angular point distribution ("Lebedev") exact up to given integration order $l$ ( $50,110,194,302, \ldots$. points per shell)


Pioneered by
Becke JCP 88, 2547 (I988), Delley, JCP 92, 508 (I990), MANY others!

## All-Electron Integrals: Rather Benign for NAOs

$$
\int d^{3} r \phi_{1 s}(\boldsymbol{r}) \hat{H} \phi_{1 s}(\boldsymbol{r})=\int d r[f(r)] \times \text { angular integral. }
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$f(r)$ for
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$f(r)$ for
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$f(r)$ for contracted Gaussian radial function:

## Overlapping Atom-Centered Grids:"Partitioning of Unity"

$$
h_{i j}=\int d^{3} r \varphi_{i}(\boldsymbol{r}) \hat{h}_{\mathrm{KS}} \varphi_{j}(\boldsymbol{r})
$$

Becke, 1988

- Rewrite to atom-centered integrands:

$$
\begin{aligned}
& \int d^{3} r f(\boldsymbol{r})=\sum_{\text {atoms }} \int d^{3} r p_{\text {atom }}(\boldsymbol{r}) f(\boldsymbol{r}) \\
& \text { exact: } \quad \sum_{\text {atoms }} p_{\text {atom }}(\boldsymbol{r})=1 \\
& \text { through } \quad p_{\text {atom }}(\boldsymbol{r})=\frac{g_{\text {atom }}(\boldsymbol{r})}{\sum_{\text {atom }^{\prime}} g_{\text {atom }^{\prime}(\boldsymbol{r})}}
\end{aligned}
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$$

through $\quad p_{\text {atom }}(\boldsymbol{r})=\frac{g_{\text {atom }}(\boldsymbol{r})}{\sum_{\text {atom }^{\prime}} g_{\text {atom }^{\prime}}(\boldsymbol{r})}$

- e.g.: $g_{\text {atom }}=\frac{\rho_{\text {atom }}(r)}{r^{2}}$ (Delley 1990)
many alternatives:
Becke 1988, Stratmann 1996, Koepernik I999, ...



## Integration in Practice: Large Systems, Small Errors!



Fully extended Polyalanine peptide molecule Ala ${ }_{20}$, DFT-PBE (203 atoms)


## Hartree Potential (Electrostatics): Overlapping Multipoles

$$
v_{\mathrm{es}}(\boldsymbol{r})=\int d^{3} r^{\prime} \frac{n\left(\boldsymbol{r}^{\prime}\right)}{\left|\boldsymbol{r}-\boldsymbol{r}^{\prime}\right|}
$$

- Partitioning of Unity: (same trick as used for integrals)

$$
n(\boldsymbol{r})=\sum_{\text {atoms }} p_{\text {atom }}(\boldsymbol{r}) n(\boldsymbol{r}) \quad \begin{gathered}
\frac{\text { Delley }}{J C P 92,} \\
508(1990)
\end{gathered}
$$

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508(1990)
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- Multipole expansion: $\quad n_{\text {atom }, l m}(r)=\int_{s=\left|r^{\prime}-R_{\text {atom }}\right|} p_{\text {atom }}\left(\boldsymbol{r}^{\prime}\right) n\left(\boldsymbol{r}^{\prime}\right) Y_{l m}(\Omega)$


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- Classical electrostatics:

$$
v_{\mathrm{es}}(\boldsymbol{r})=\sum_{\text {atoms }} \sum_{l m}^{l_{\text {max }}} v_{\text {atom }, l m}\left(\left|\boldsymbol{r}-\boldsymbol{R}_{\text {atom }}\right|\right) Y_{l m}\left(\Omega_{\text {atom }}\right)
$$

## Electrostatics: Multipole expansion

$$
v_{\mathrm{es}}(\boldsymbol{r})=\sum_{\text {atoms }} \sum_{l m}^{l_{\mathrm{max}}} v_{\text {atom }, l m}\left(\left|\boldsymbol{r}-\boldsymbol{R}_{\text {atom }}\right|\right) Y_{l m}\left(\Omega_{\text {atom }}\right)
$$

Polyalanine Ala ${ }_{20}$, DFT-PBE (203 atoms)
$\alpha$-helical vs. extended:Total energy convergence with $l_{\max }$


