

Bulk doping effects in hybrid organic/ inorganic systems from quantum mechanical first principles

Patrick Rinke¹, Yong Xu¹, Oliver T. Hofmann¹,
Niko Moll² and Matthias Scheffler¹

¹Fritz-Haber-Institut der Max-Planck-Gesellschaft, Berlin

²IBM Research - Zurich

May 2013



MAX-PLANCK-GESELLSCHAFT

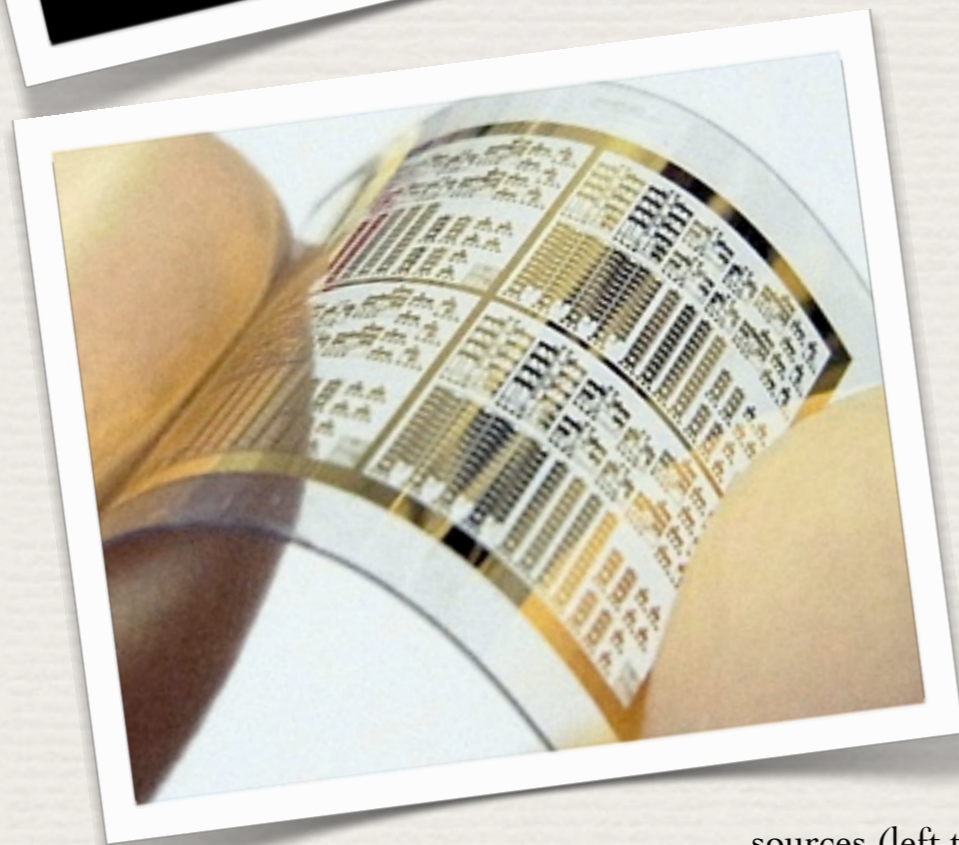
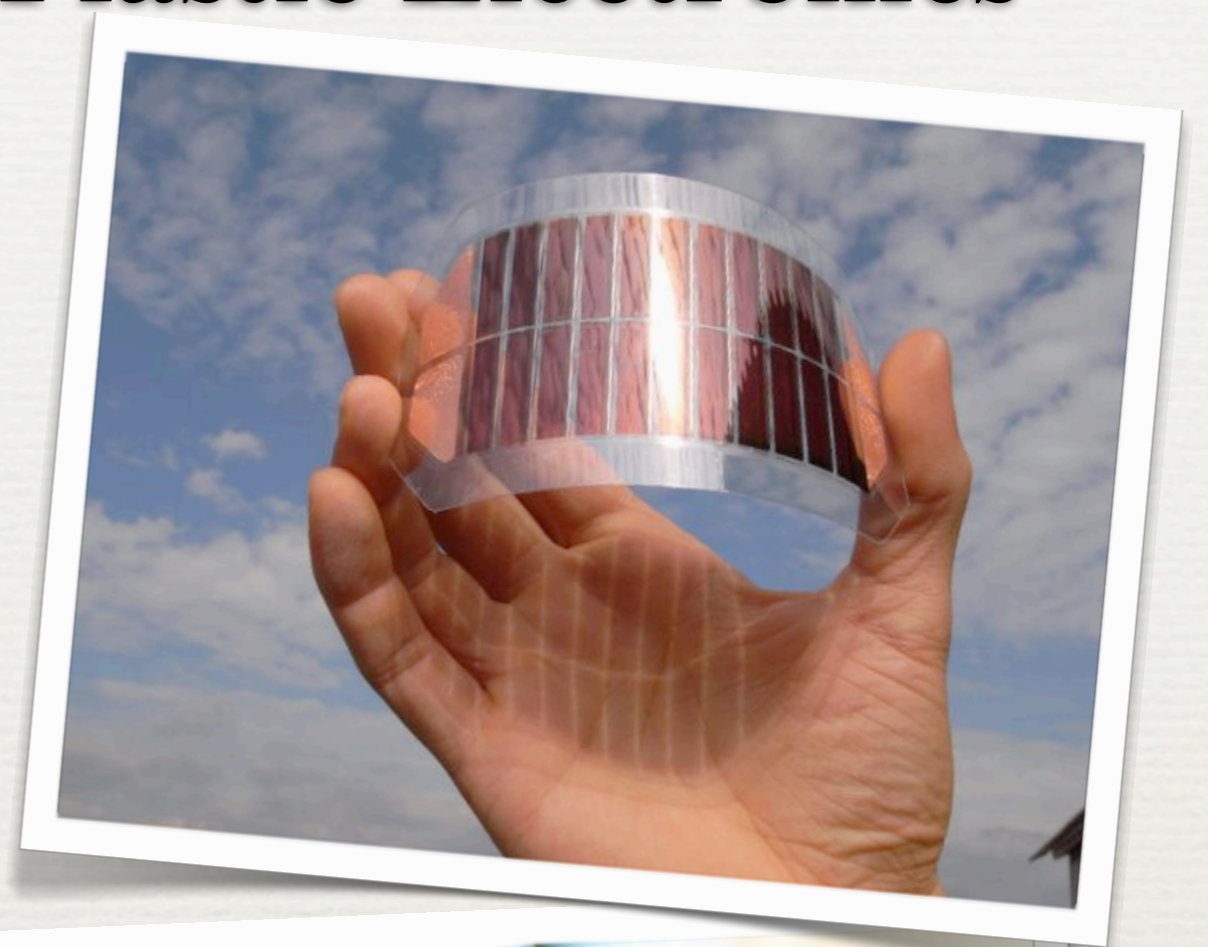


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Organic Electronics - Plastic Electronics



Organic Electronics - Plastic Electronics



Inorganic/organic interfaces are already present...

... and affect or determine device properties.

Can we understand and control them?



Hybrid Inorganic/Organic Interfaces (HIOs)

Combine the best of two worlds...

Hybrid electronics:

- make interface central

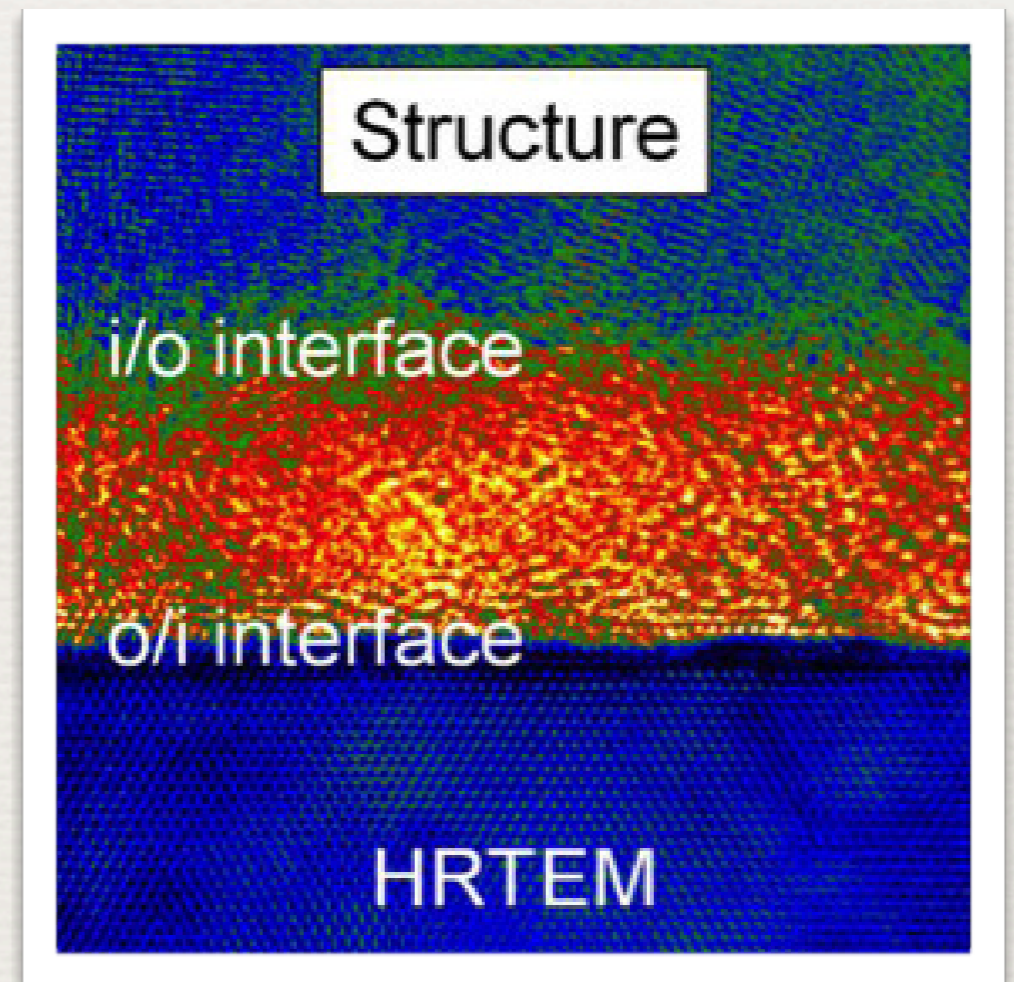
inorganic materials:

- stable crystal structures
- good growth control
- high charge carrier mobilities

organic materials:

- strong light matter coupling
- large chemical compound space

synergy or more!

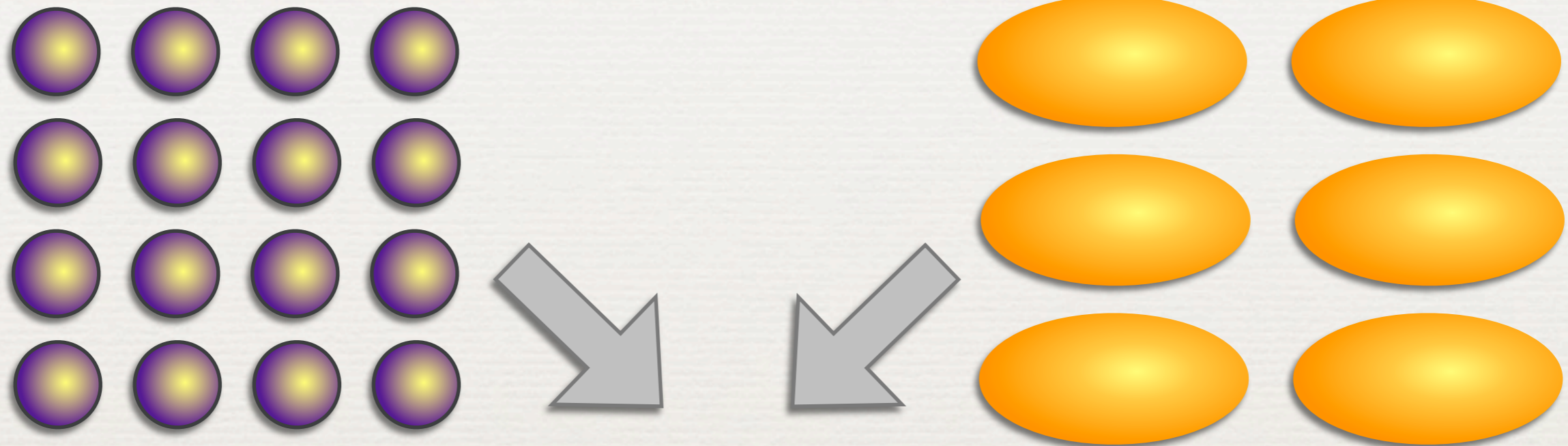


ZnO/p-sexiphenyl
(courtesy of S. Blumstengel)

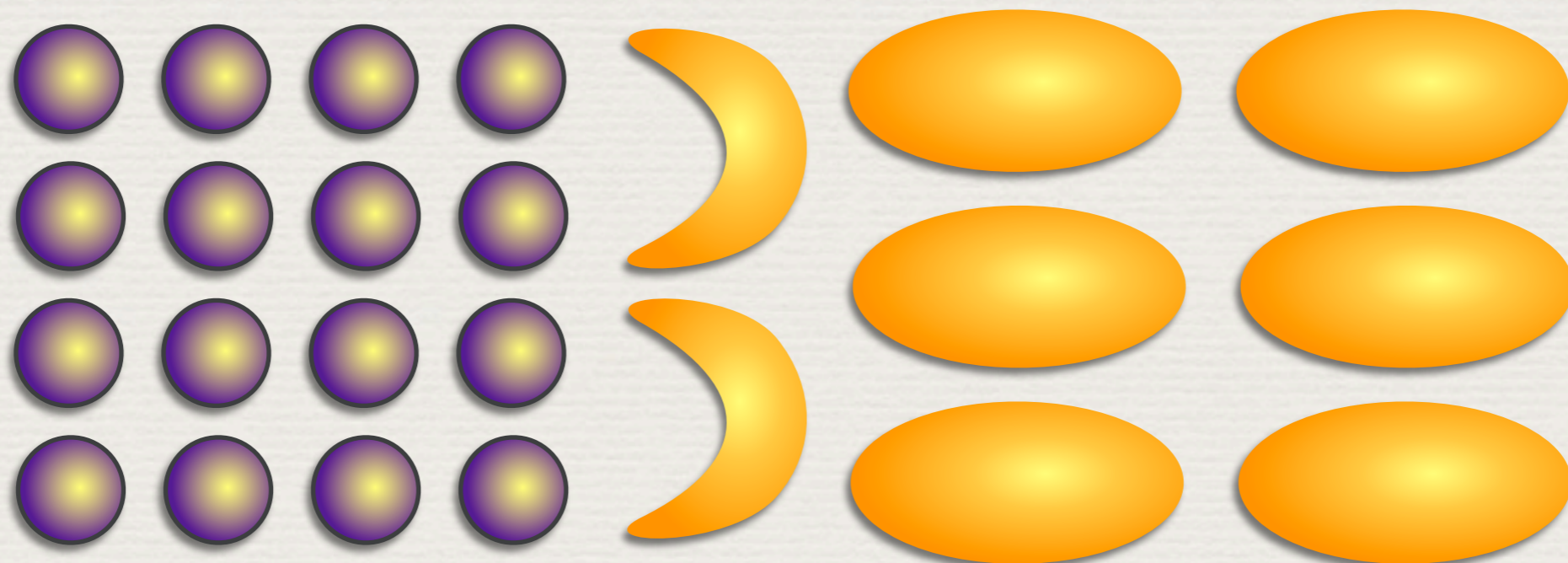
Potential for new physics at HIOS

solid

organic



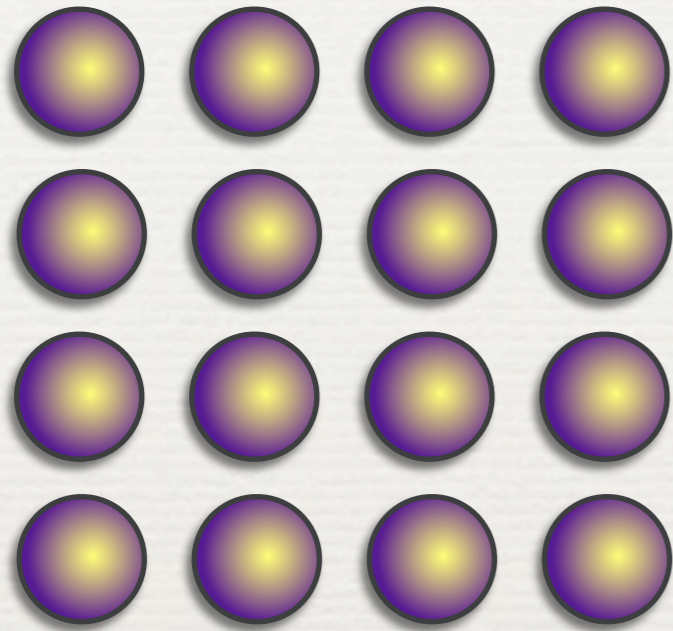
- **potential for new interface morphologies**



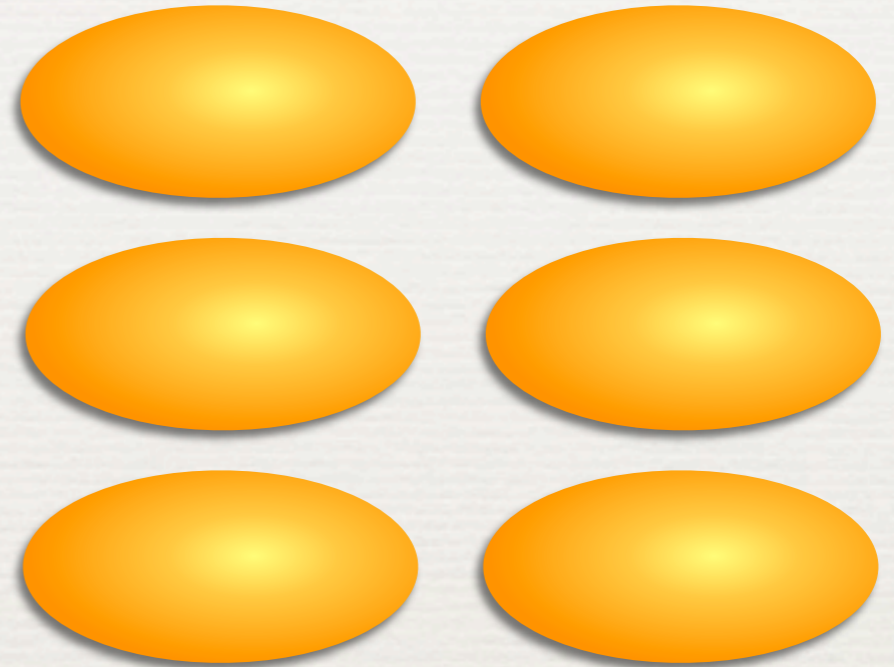
HIOS

HIOS in reality - defects and disorder

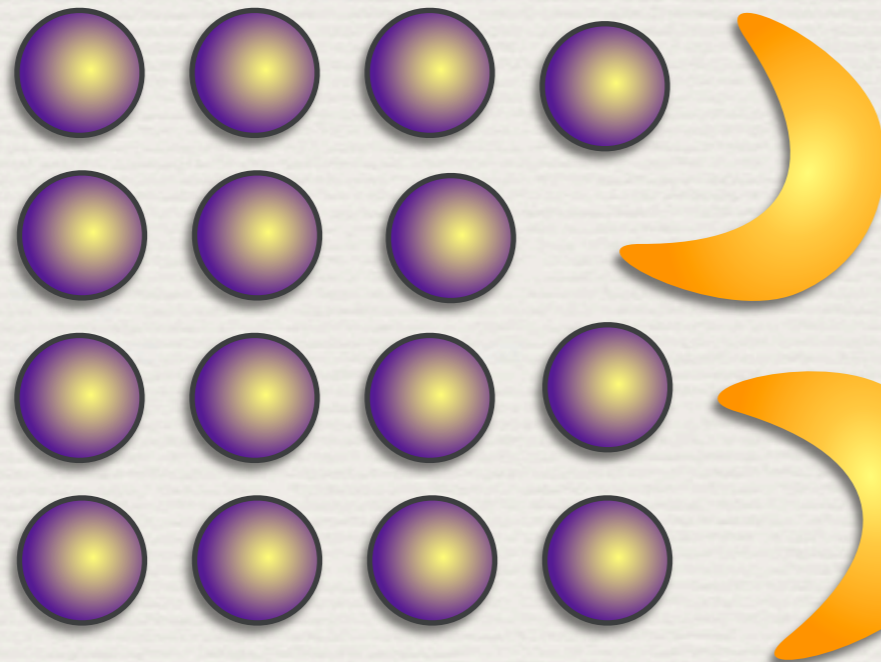
solid



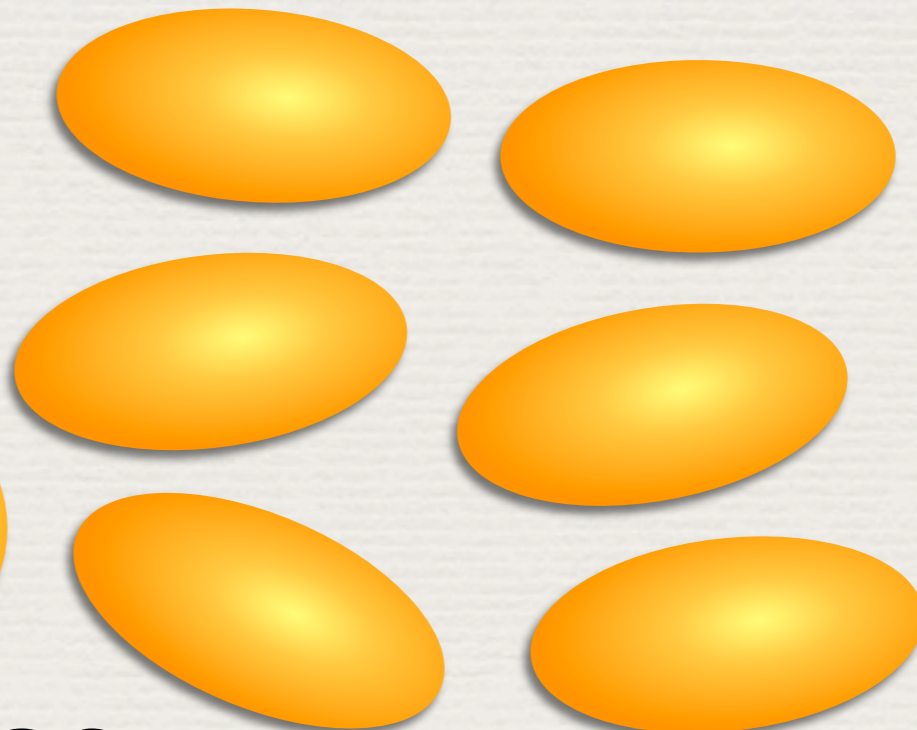
organic



defects...



disorder...



HIOS

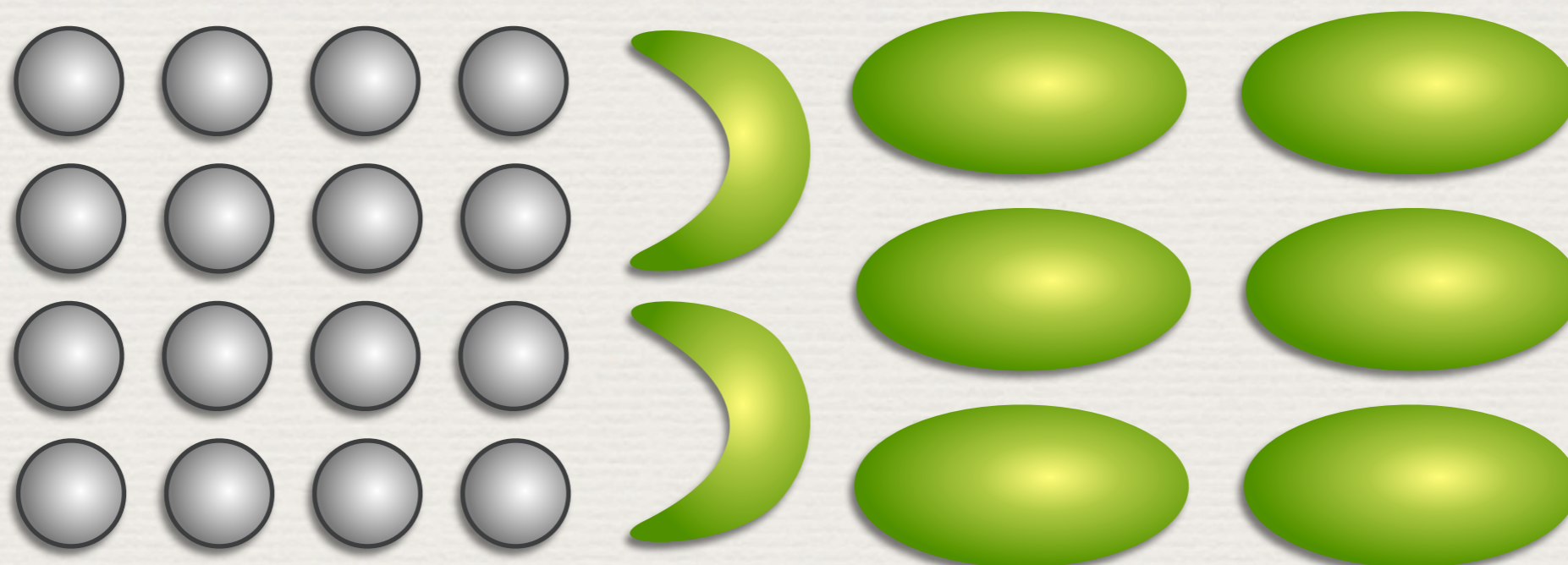
Optimizing HIOS - the right combination



- **What is the right combination?**

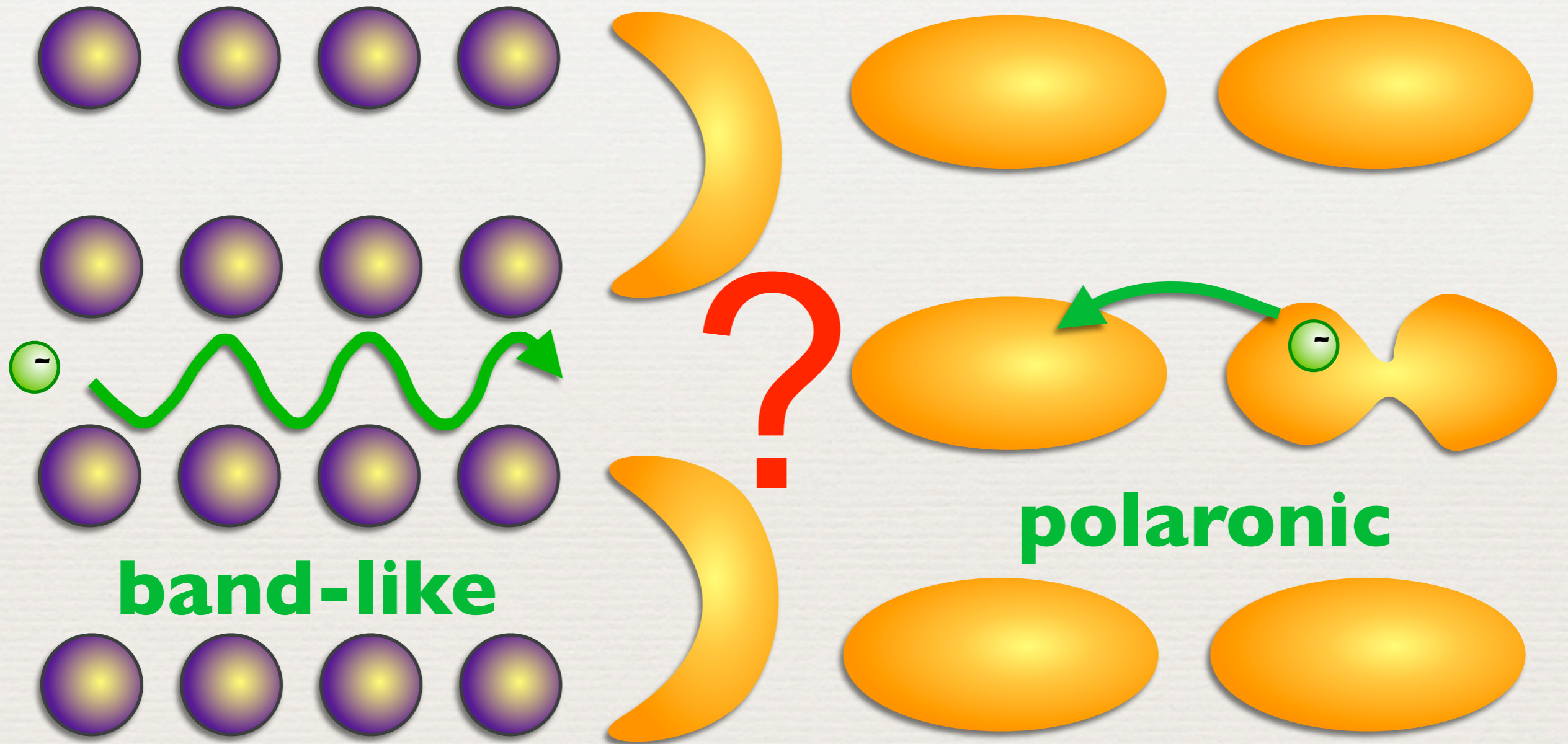


- **Vast chemical compound space**



HIOS

Fundamental questions at HIOS interfaces



- What is the nature of charge carriers?
- Do new quasiparticles emerge at interface?

Atomistic understanding of molecules@surfaces

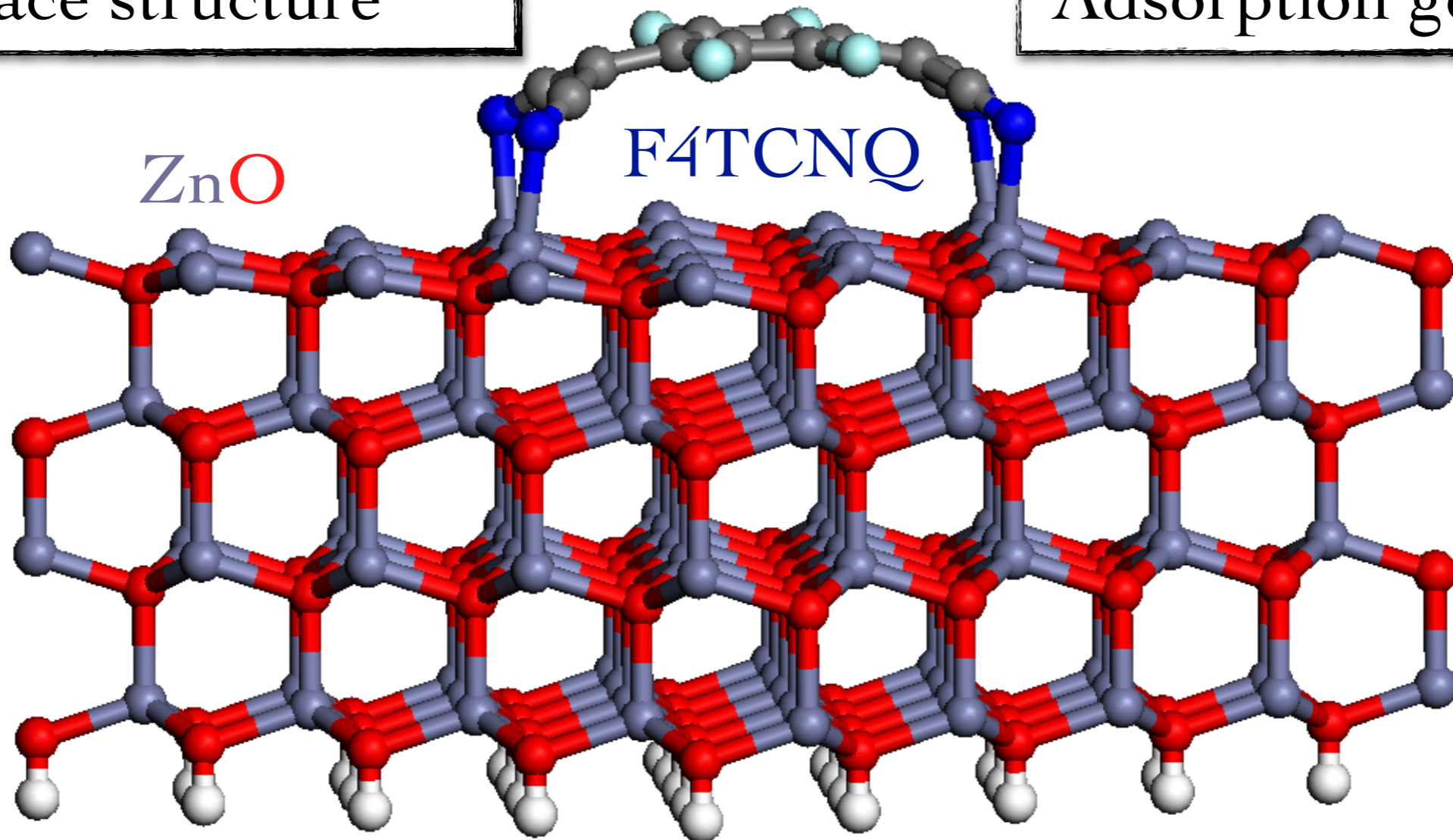
Charge transfer

Level alignment

Van der Waals interaction

Surface structure

Adsorption geometry



Atomistic understanding of molecules@surfaces

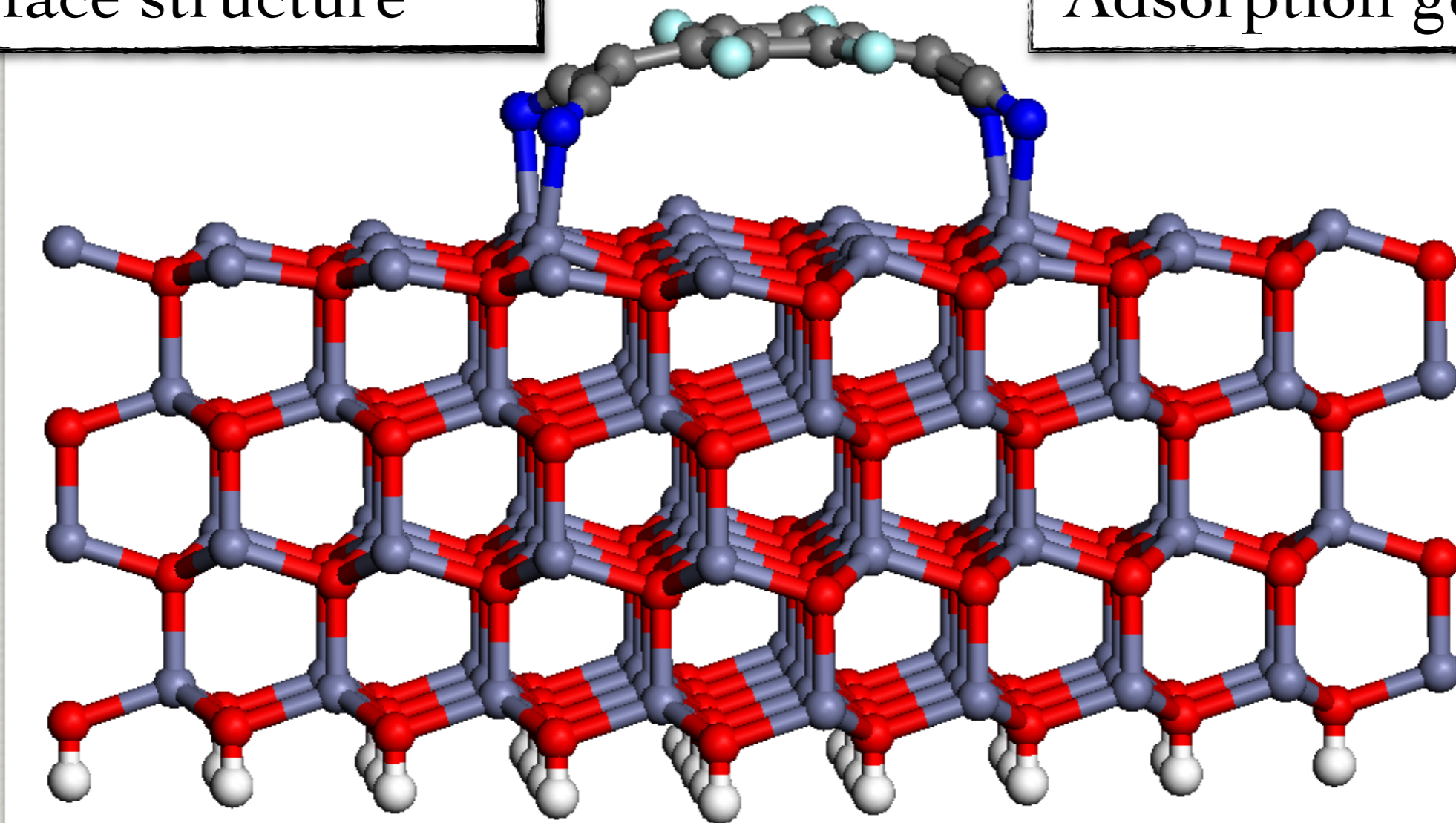
Charge transfer

Level alignment

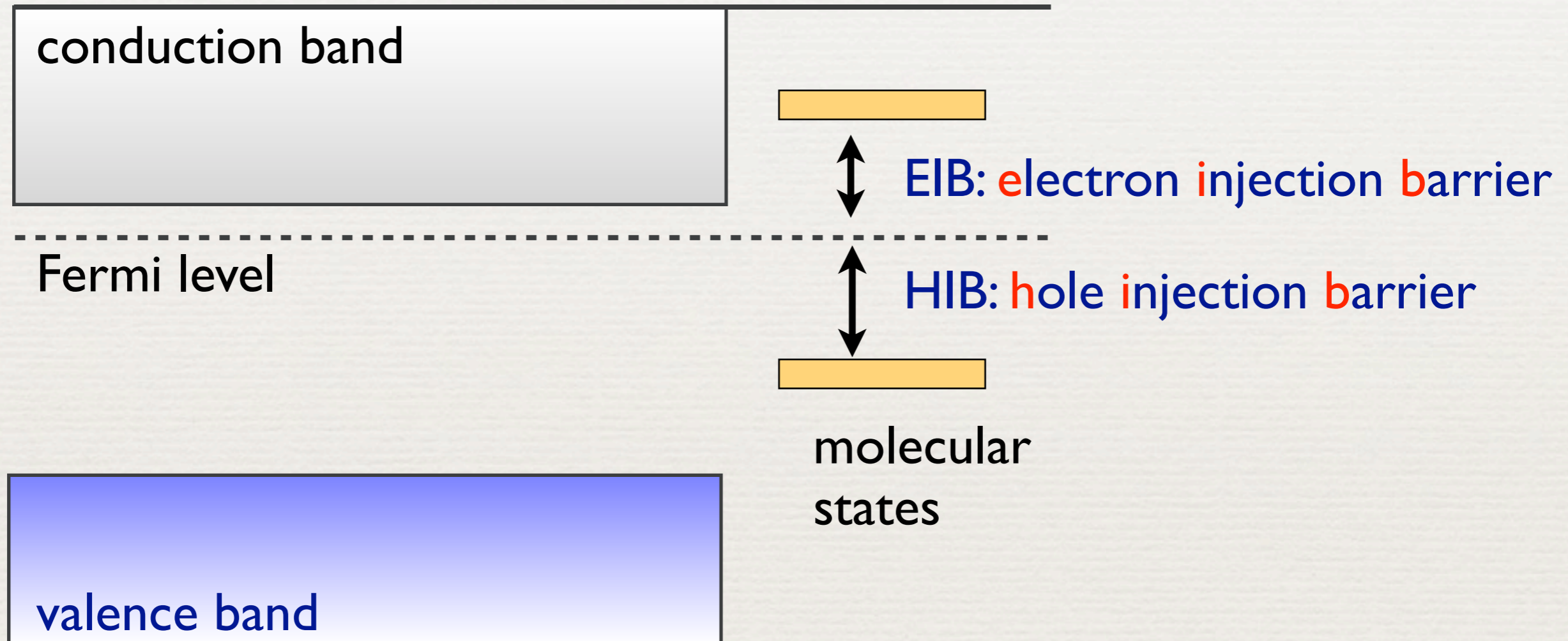
Van der Waals interaction

Surface structure

Adsorption geometry



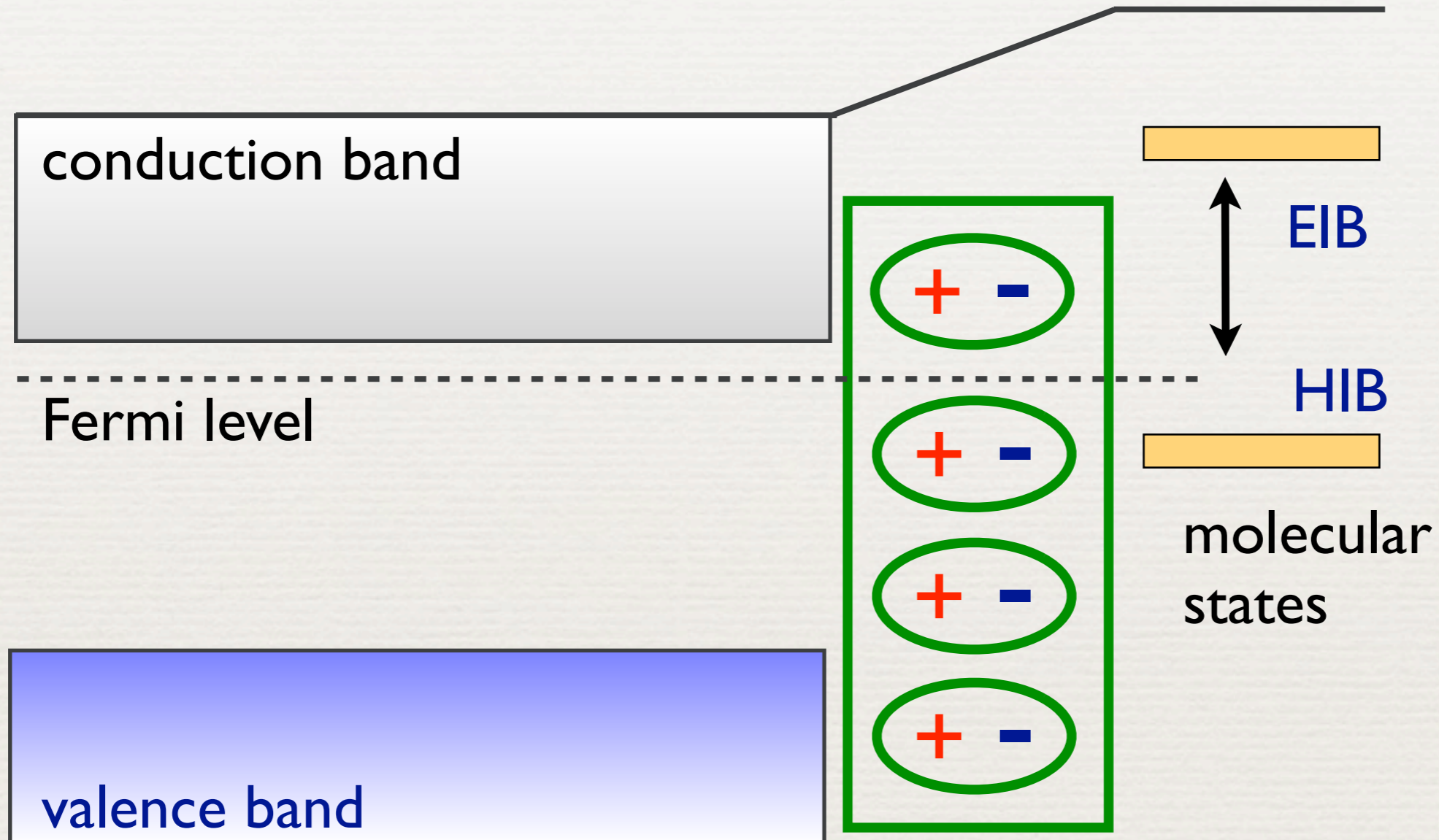
Level alignment at interface



injection limited current:

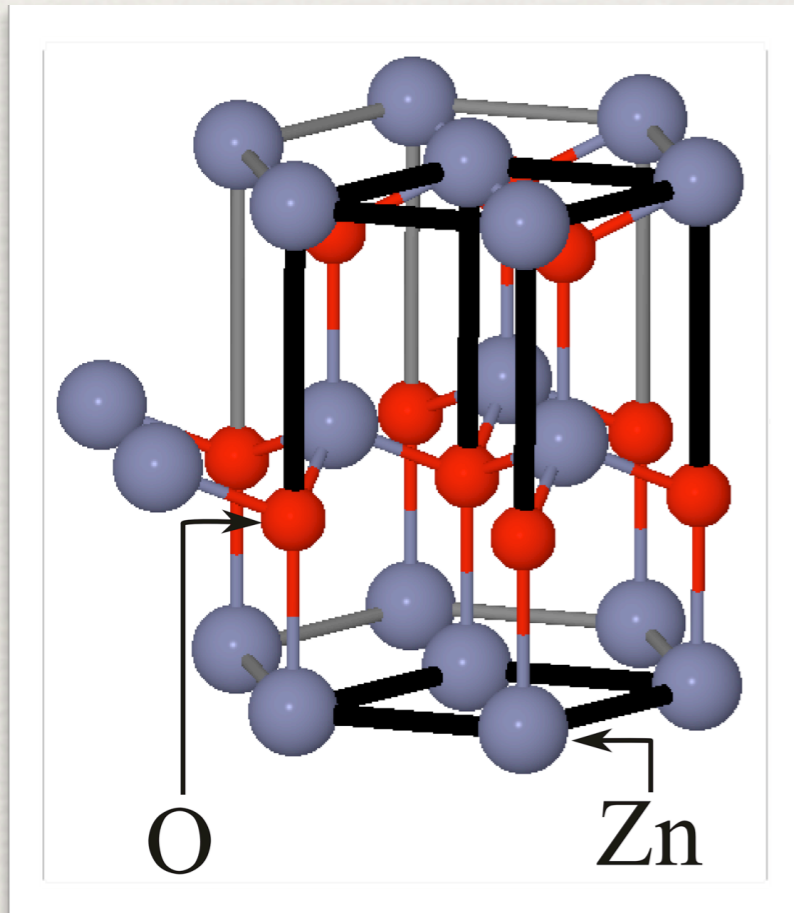
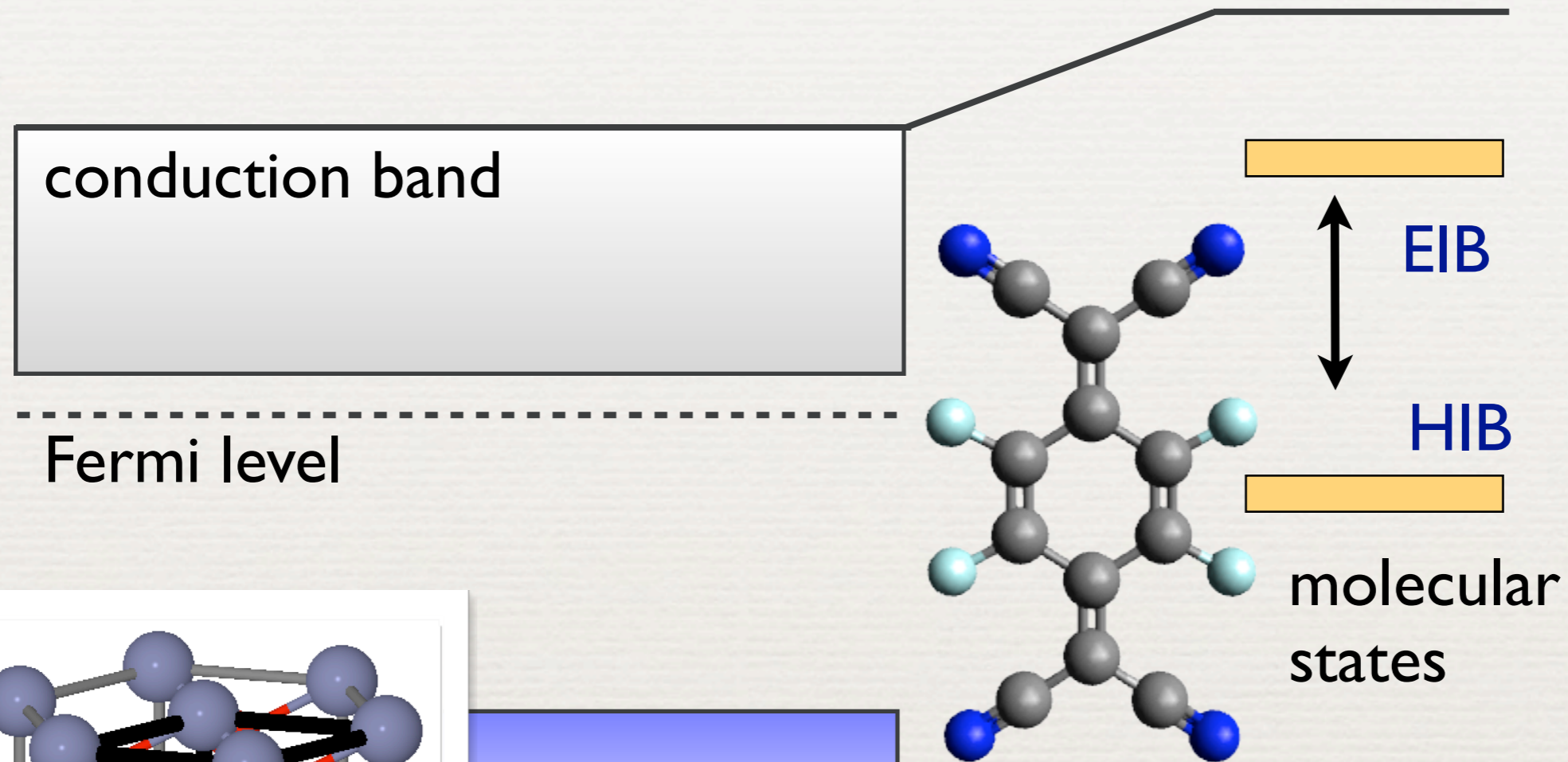
$$j \propto AT^2 \exp\left(-\frac{\text{charge injection barrier}}{k_B T}\right)$$

Modifying level alignment at interface



interface layer to adjust level alignment

Modifying level alignment at interface

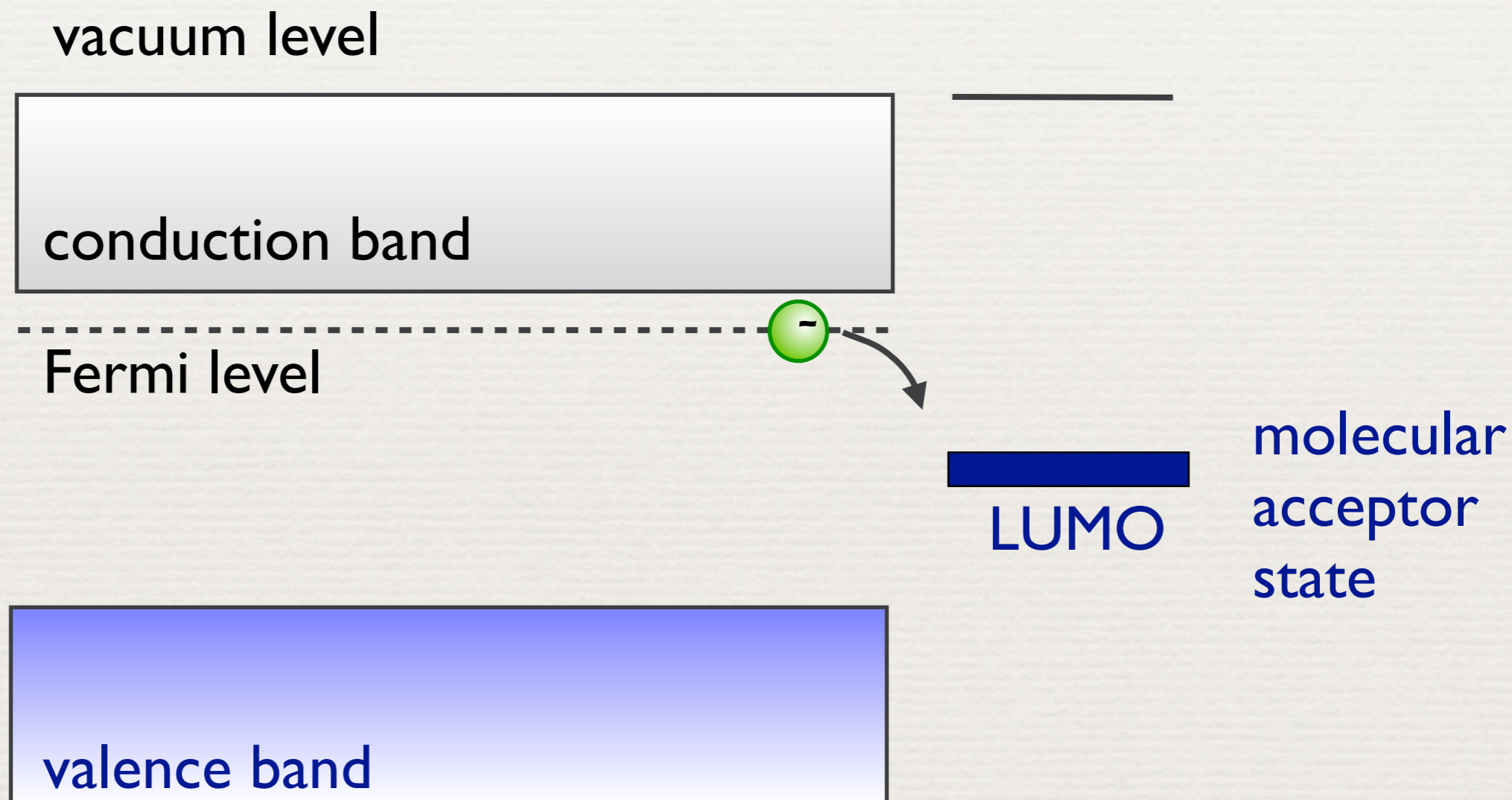


ZnO

**e.g. acceptor F4TCNQ
(tetrafluoro-tetracyano
quinodimethan)**

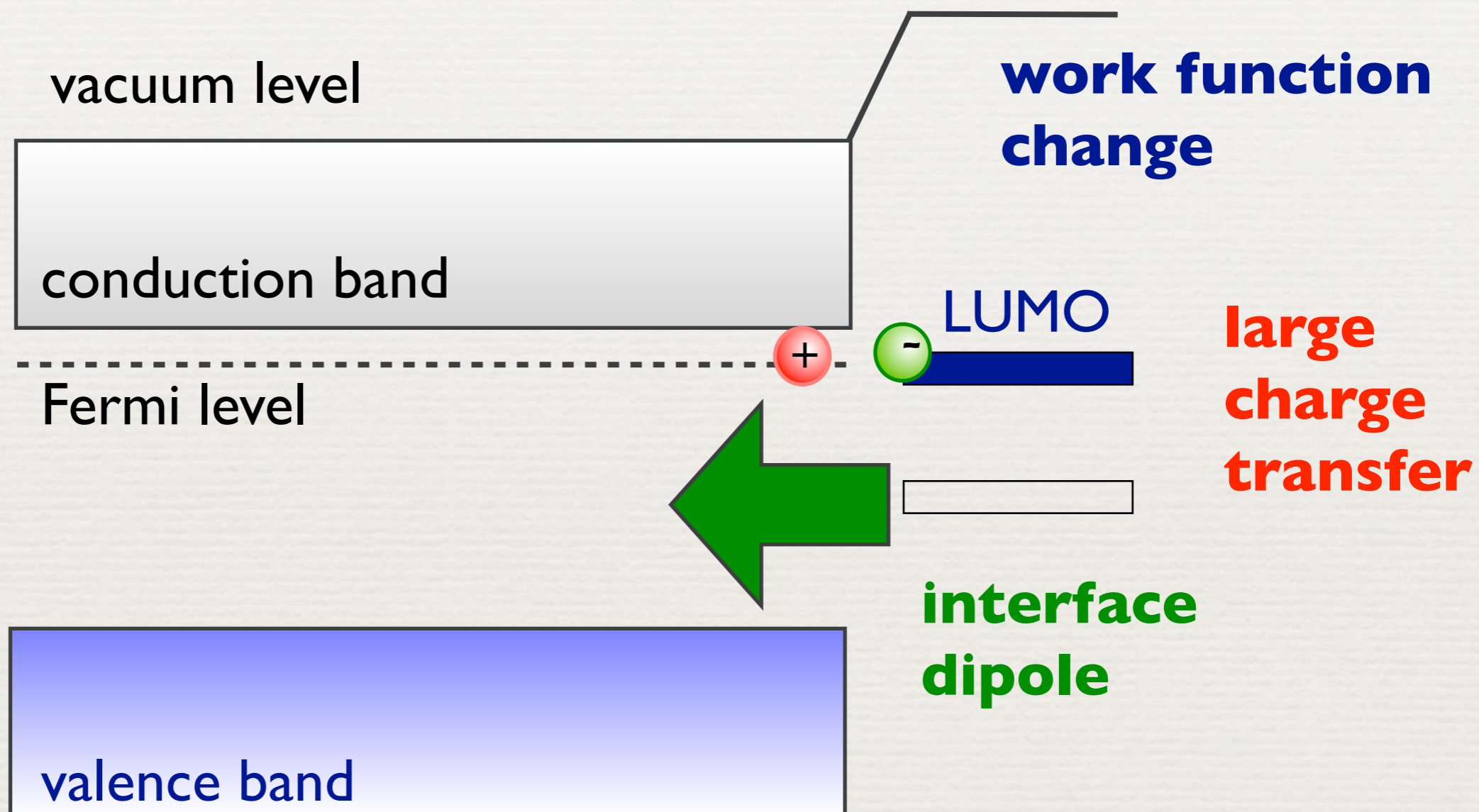
Charge transfer induced work function increase

High doping \Rightarrow flat bands \Rightarrow metallic regime



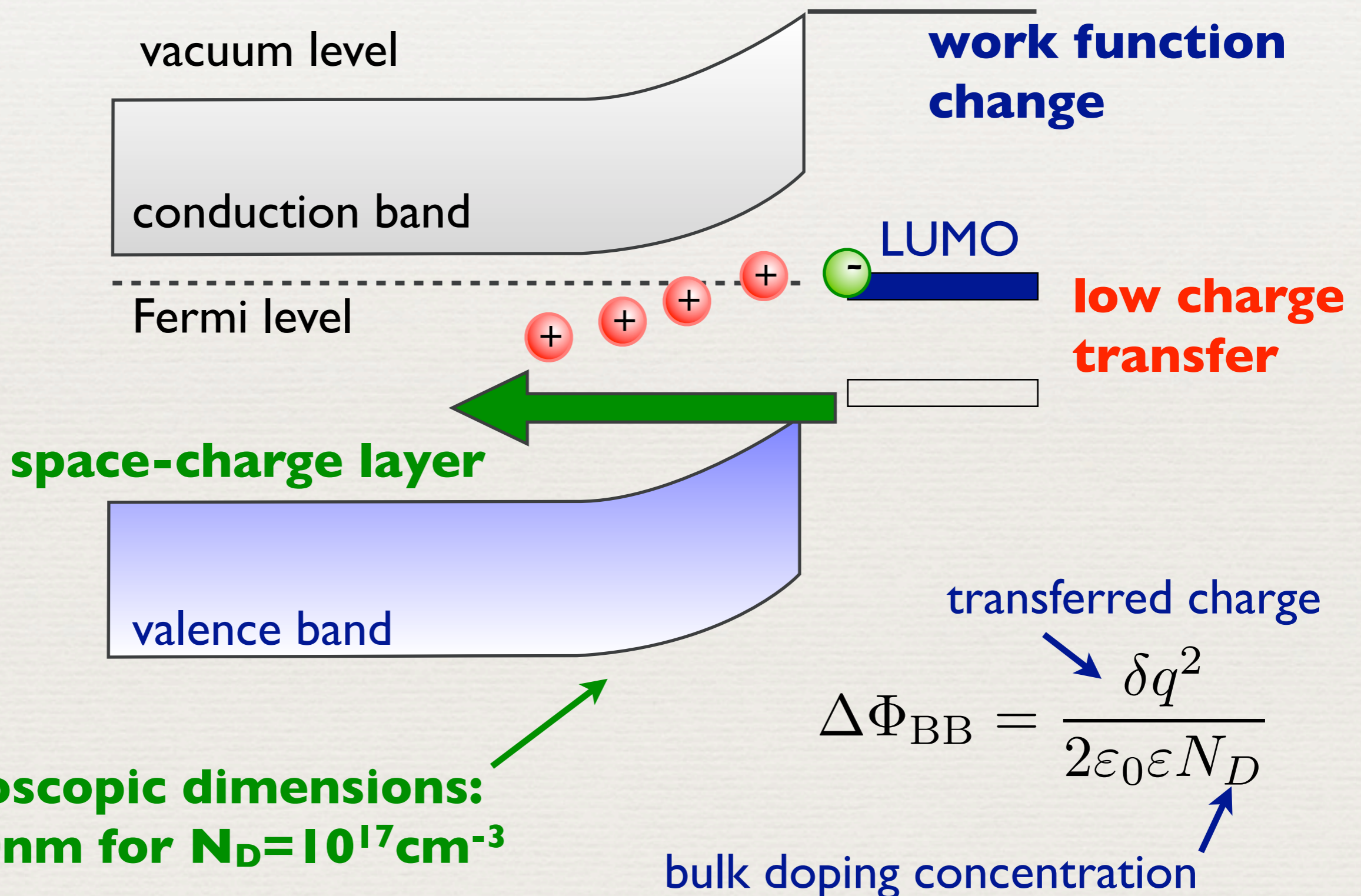
Charge transfer induced work function increase

High doping \Rightarrow flat bands \Rightarrow metallic regime



Formation of space-charge layer

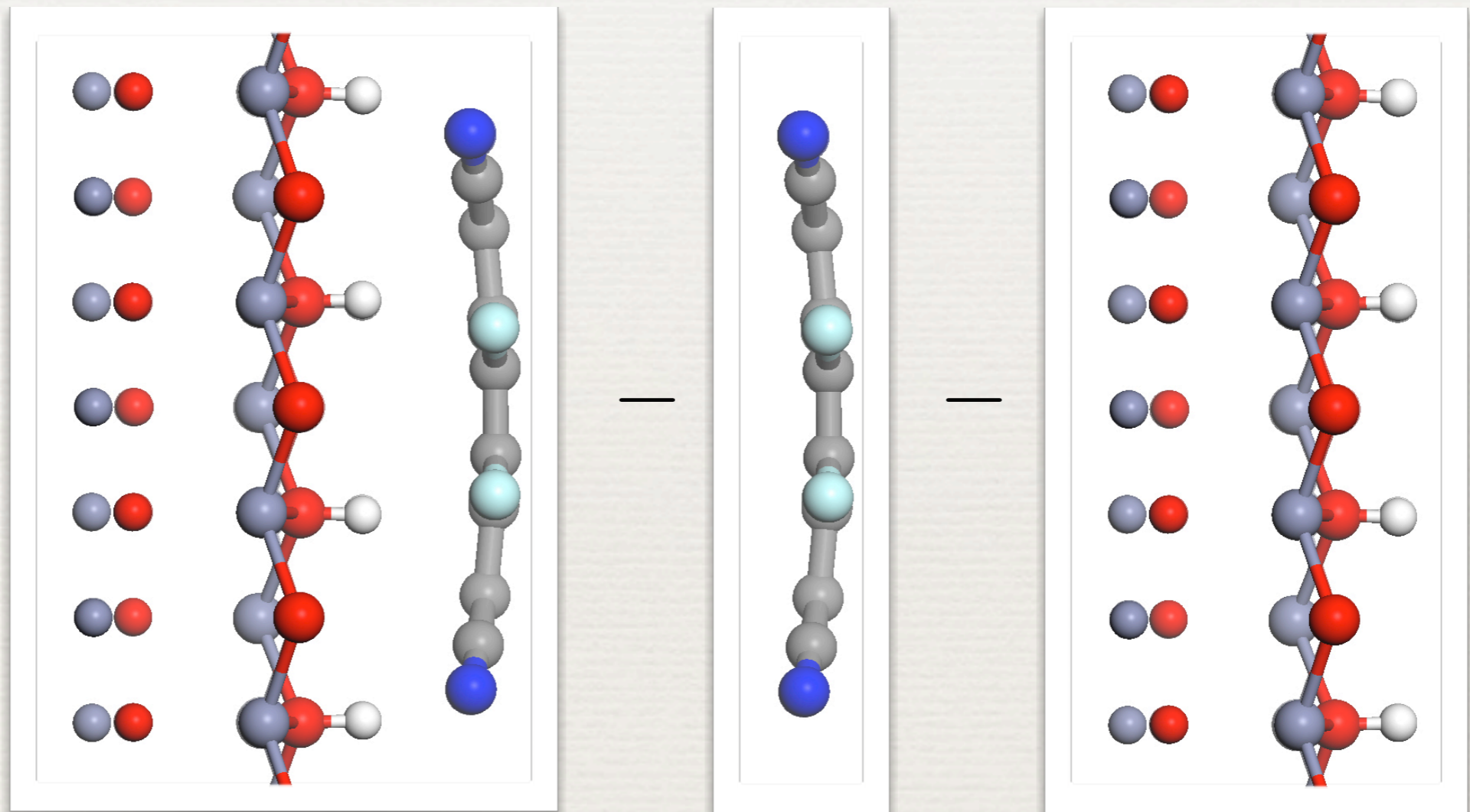
low doping \Rightarrow band bending



Quantum mechanical treatment

adsorption energy:

$$E^{ads} =$$



$$E^{ads} = E^{mol@surf} - E^{mol} - E^{surf}$$

Density-Functional Theory (DFT)

Our Code: FHI-aims [1]

All-electron **DFT and beyond** code developed in the Fritz-Haber-Institut

- Accurate numerical atomic orbital basis sets:

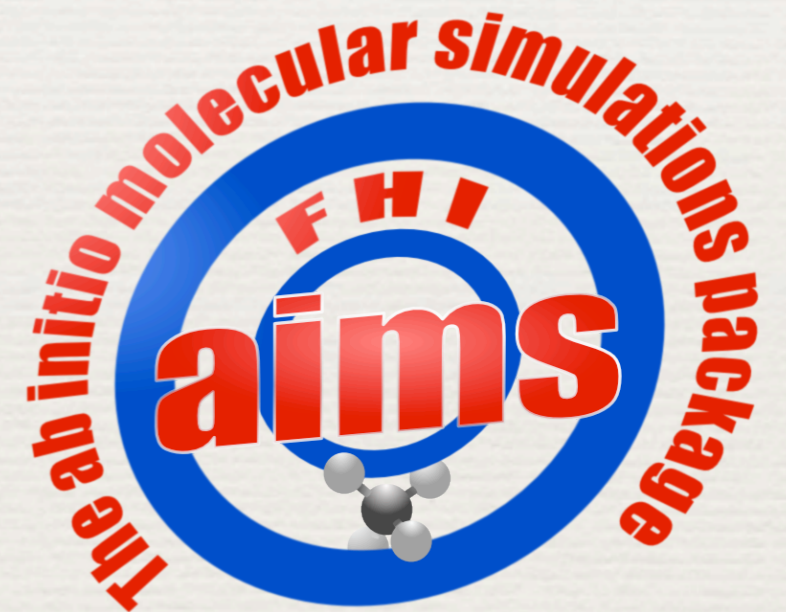
$$\varphi_i(\vec{r}) = \frac{u_i(r)}{r} Y_{lm}(\Omega)$$

- Massively scalable, massively parallel
- DFT functionals: LDA, **PBE**, hybrids (HSE), ...
- pairwise van der Waals (vdW) scheme [2]:

$$E_{\text{vdW}} = - \sum_{B>A} f_{\text{damp}} \frac{C_6^{AB}[n]}{R_{AB}^6}$$

density-dependent C_6 coefficient

distance between atoms A and B



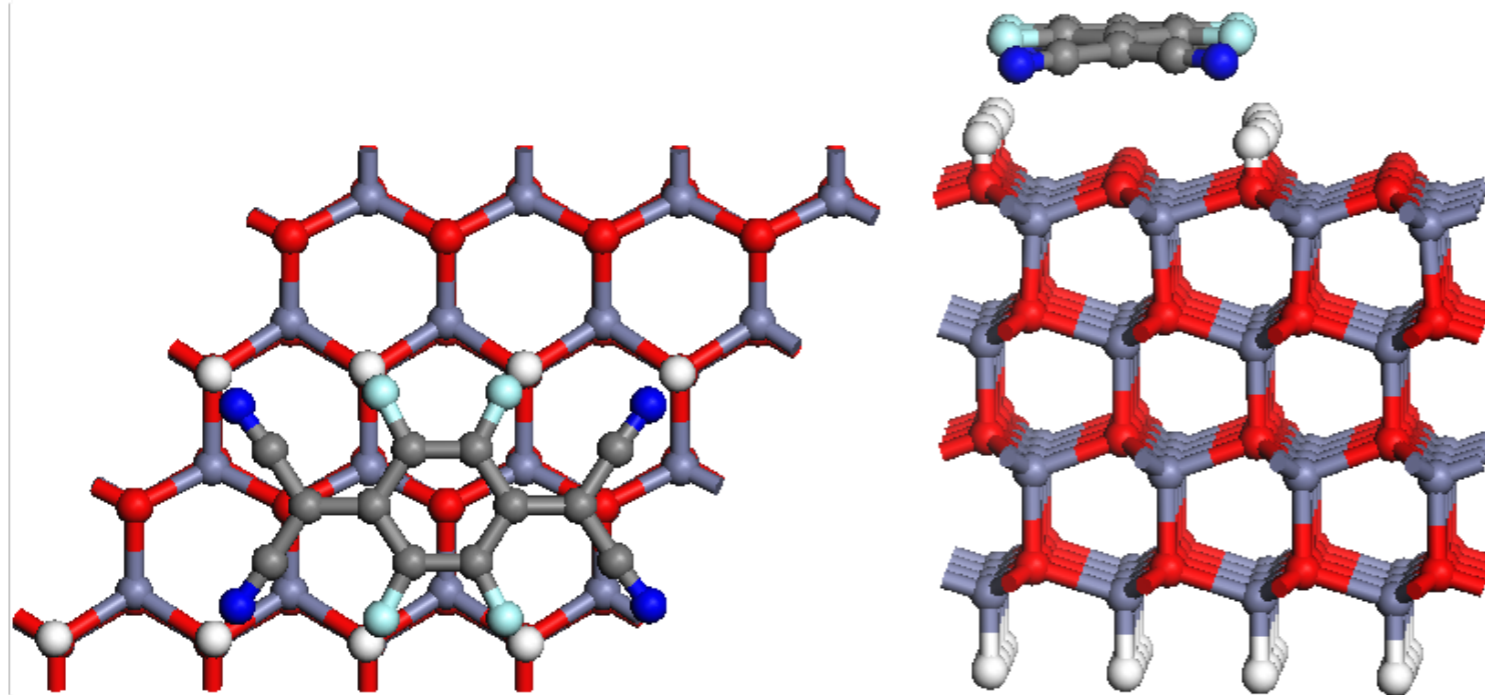
[1] V. Blum et al. *Comp. Phys. Comm.* **180**, 2175 (2009)

[2] A. Tkatchenko and M. Scheffler *Phys. Rev. Lett.* **102**, 073005 (2009)

Adsorption geometries and energies

$$E^{ads} = E^{mol@surf} - E^{mol} - E^{surf}$$

F4TCNQ on hydrogen terminated ZnO(000-1)



$$E^{ads} = 1.84 \text{ eV}$$

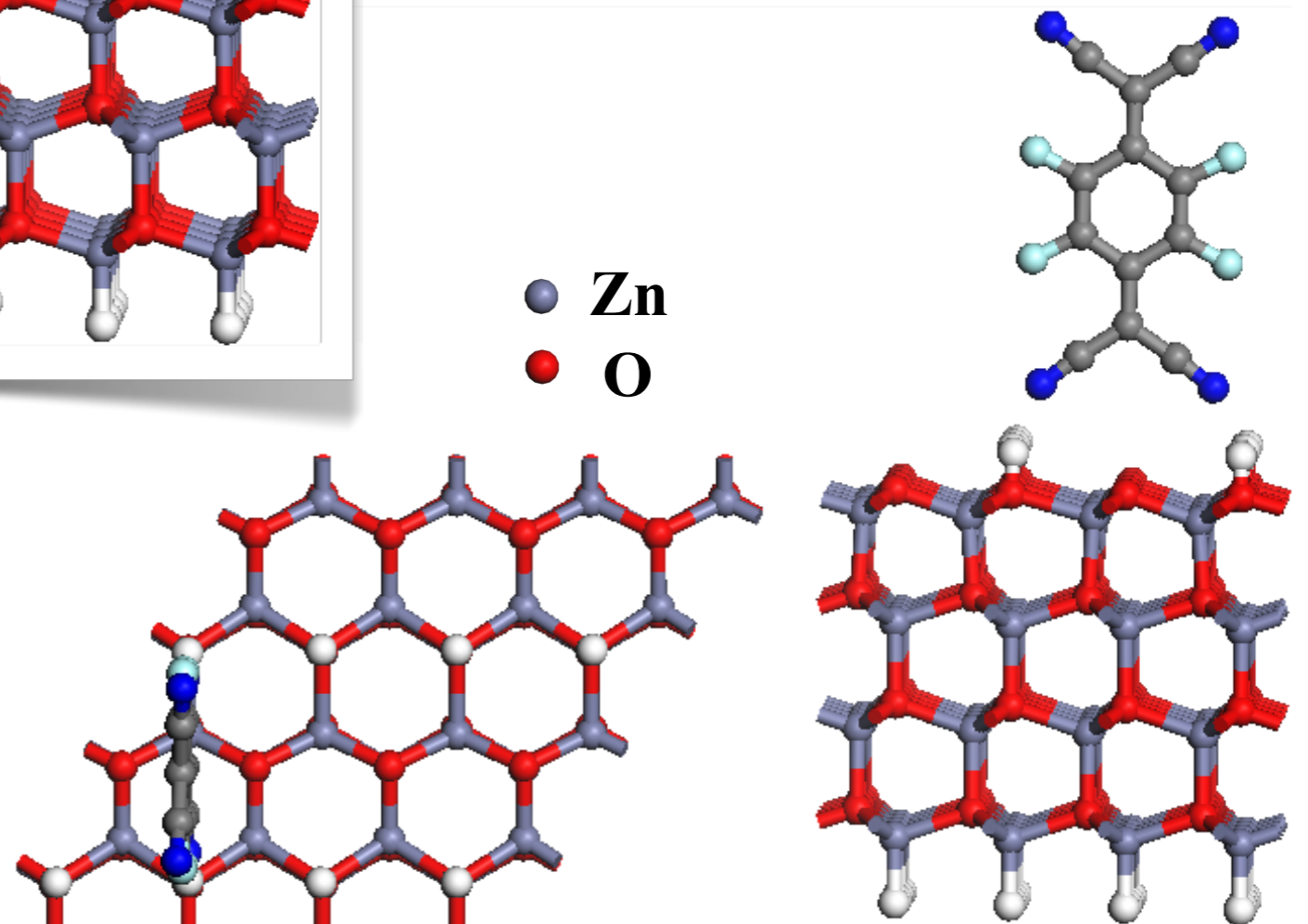
flat-lying

all energies: HSE*+vdW^{scr}

● Zn
● O

upright

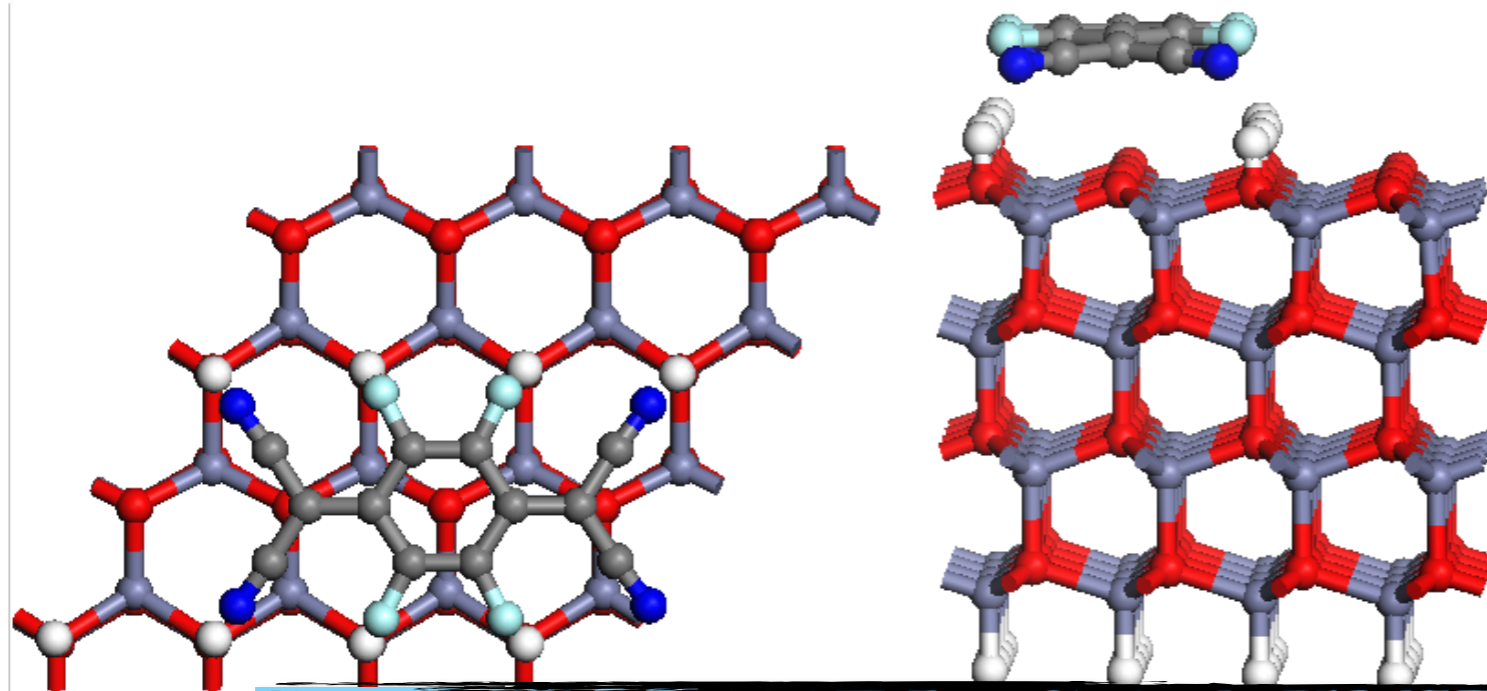
$$E^{ads} = 0.40 \text{ eV}$$



Adsorption geometries and energies

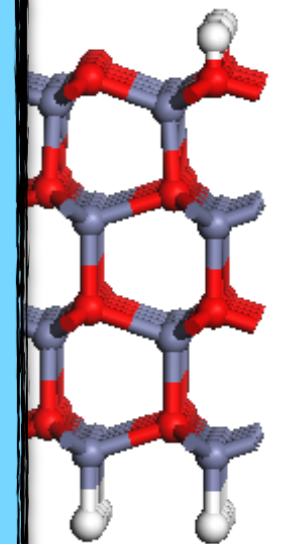
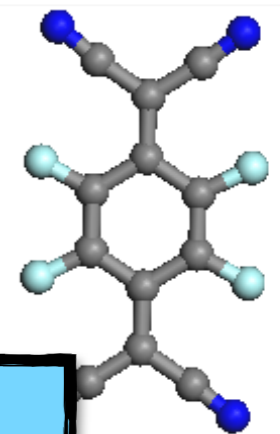
$$E^{ads} = E^{mol@surf} - E^{mol} - E^{surf}$$

F4TCNQ on hydrogen terminated ZnO(000-1)



upright

$$E^{ads} = 0.40 \text{ eV}$$



Structural features:

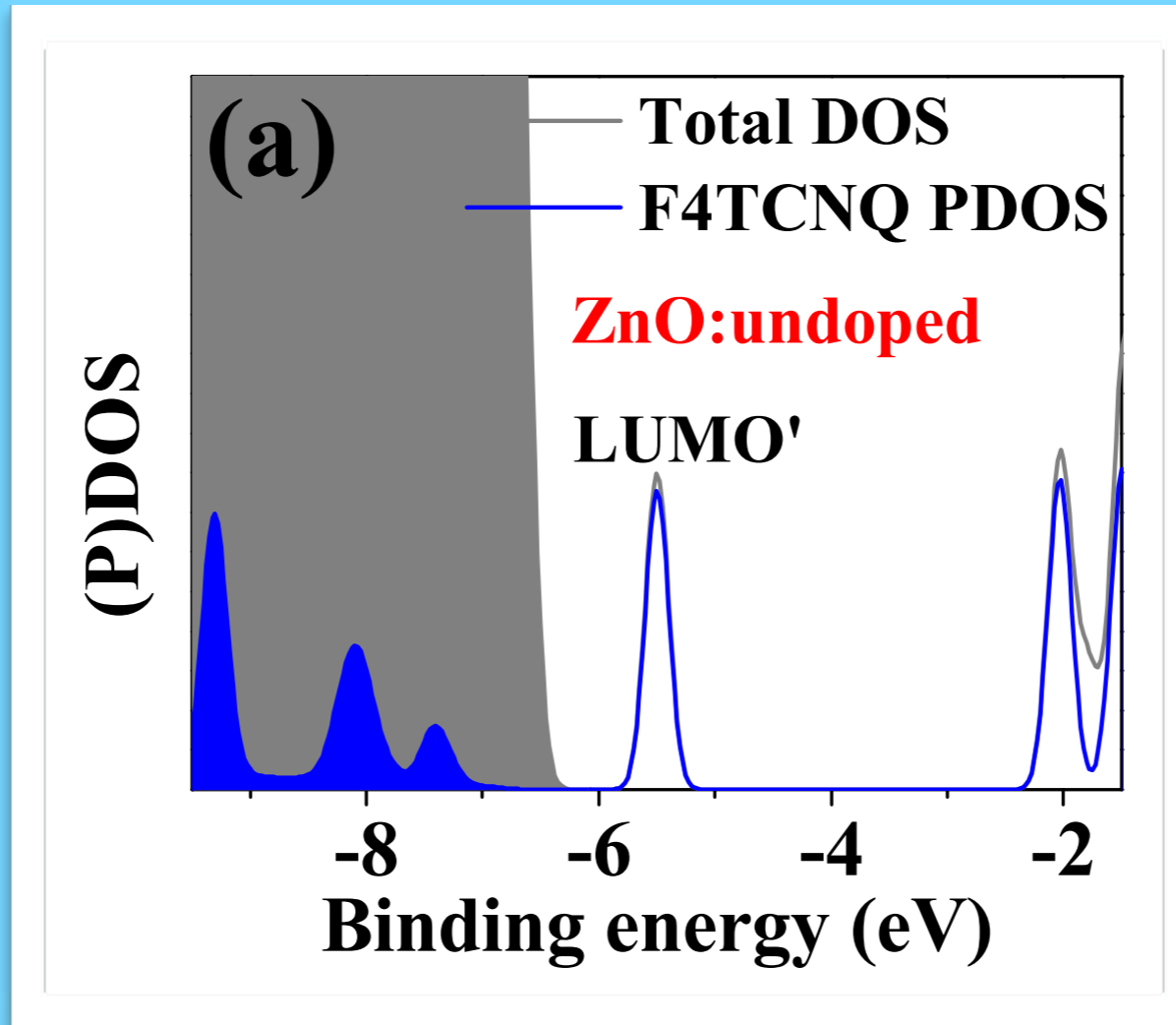
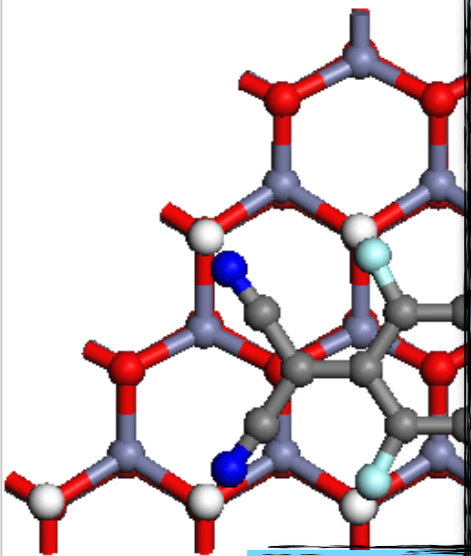
- weak binding (hydrogen bonding, vdW)
- CN groups bind to hydrogens
- F4TCNQ does not bend appreciably (unlike on metals)

all ene

Adsorption geometries and energies

$$E_{ads} = E^{mol@surf} - E^{mol} - E^{surf}$$

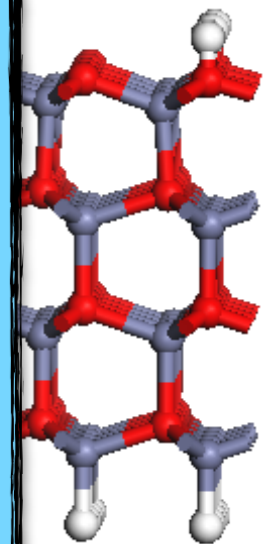
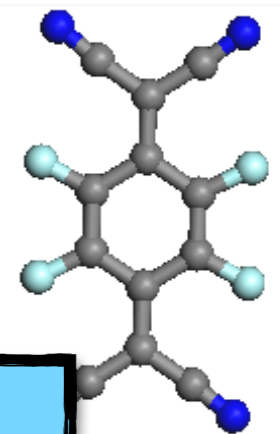
F4TCNQ



0-1)

upright

= 0.40 eV



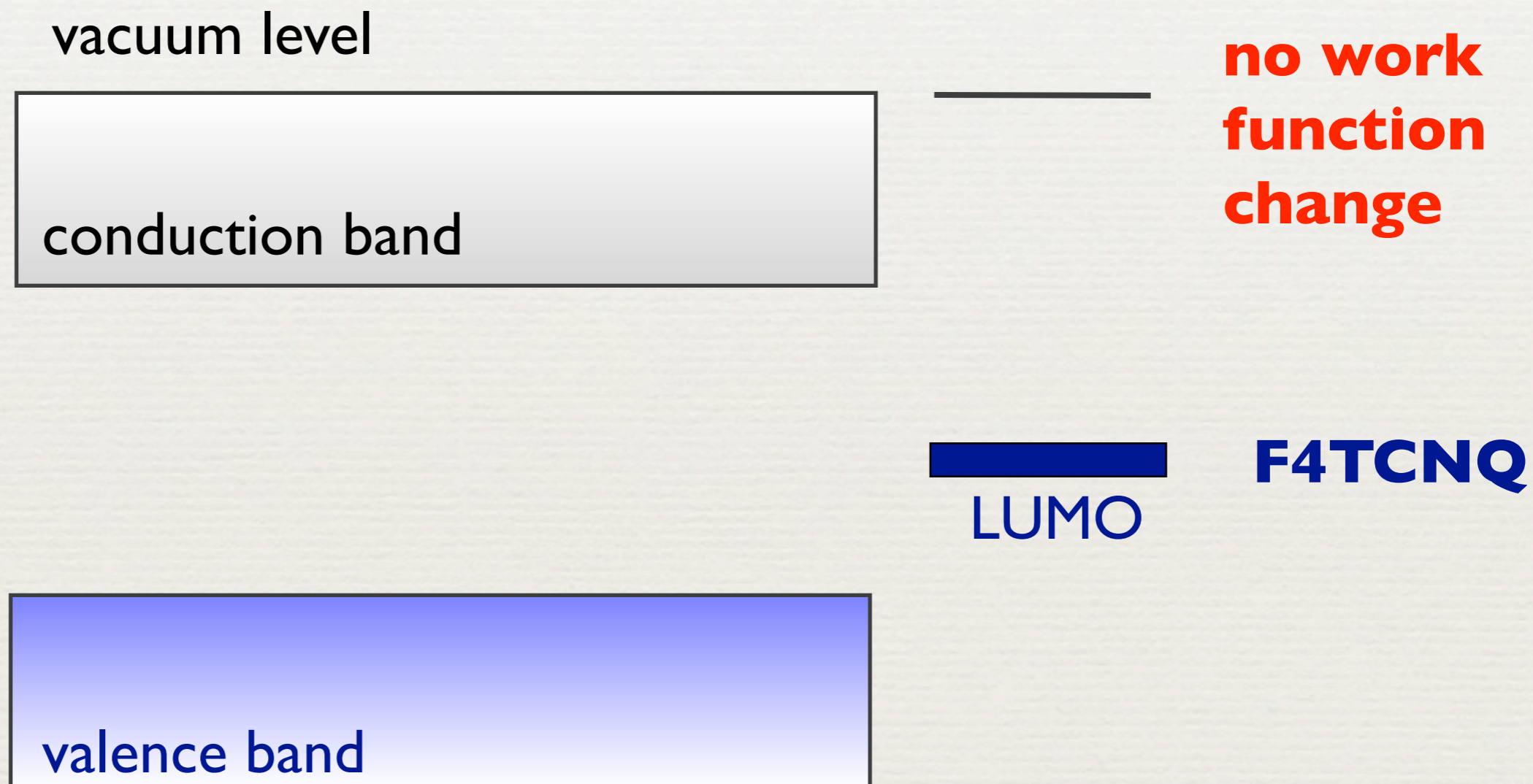
Str

-
- CN groups bind to hydrogens
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all ene

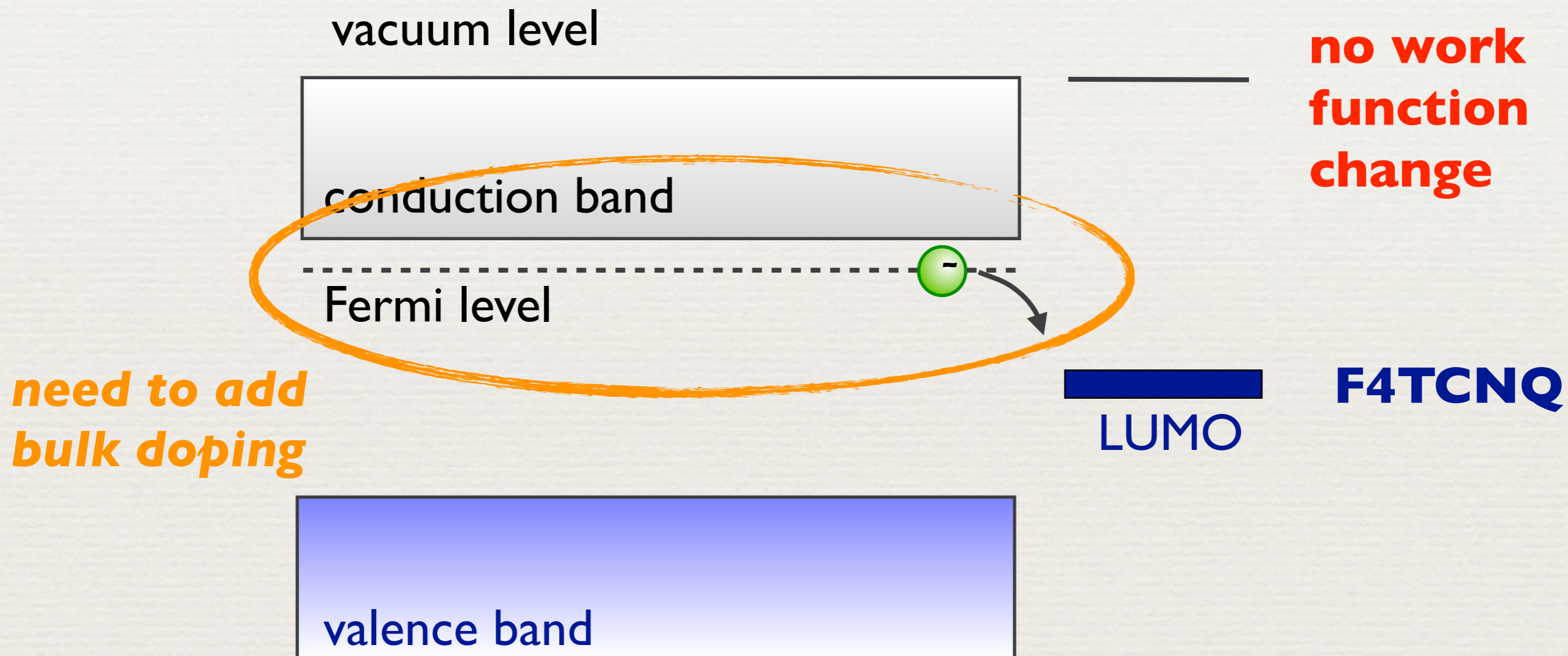
F4TCNQ@ZnO in DFT

Our normal DFT calculations: undoped



F4TCNQ@ZnO in DFT

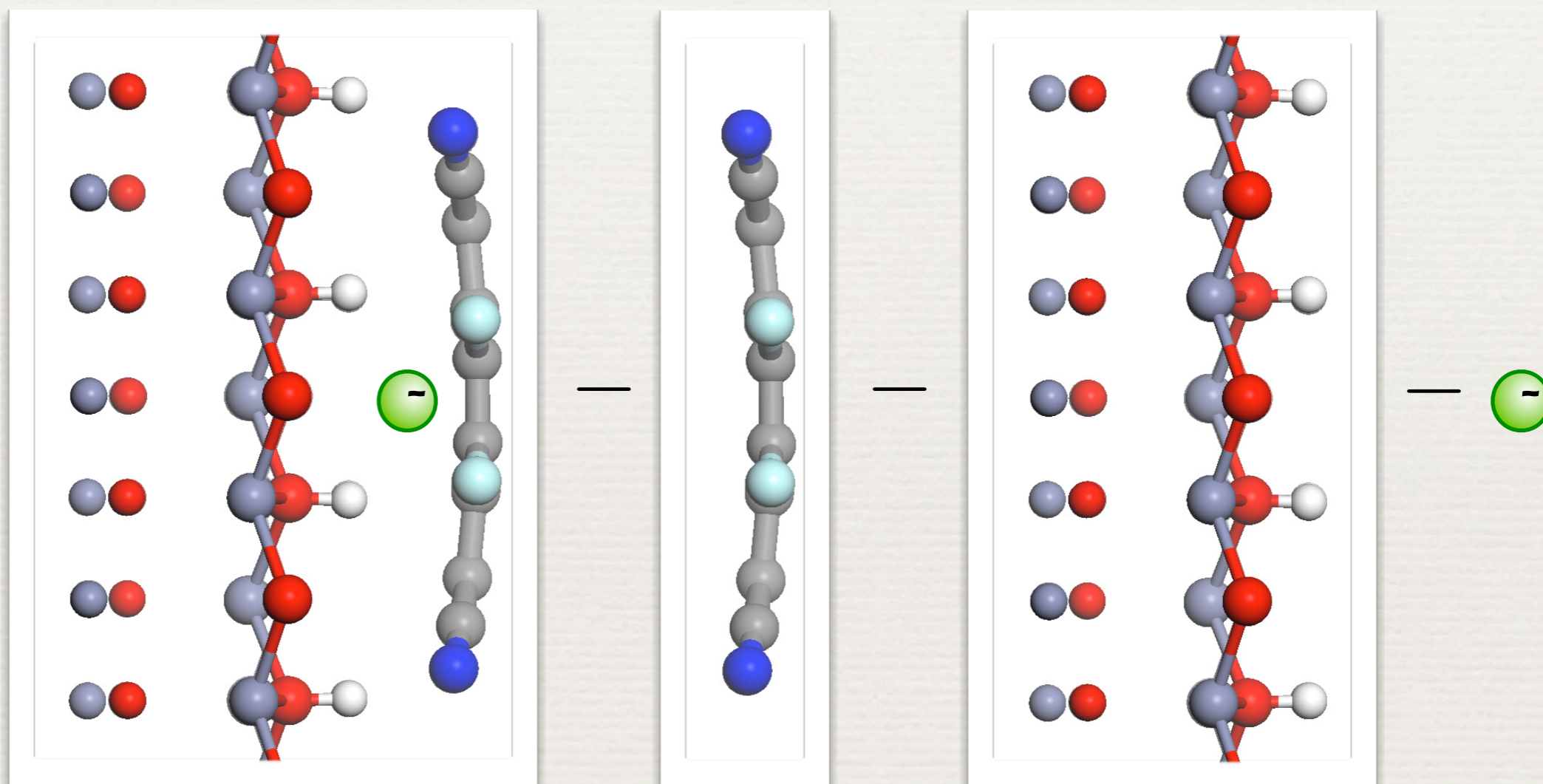
Our normal DFT calculations: undoped



Step 1: add electrons to DFT calculation

adsorption energy:

E^{ads} =



$$E^{ads}(q) = E^{mol@surf}(q) - E^{mol} - E^{surf} + q\Delta\epsilon_F$$

energy of electron reservoir



Adding electrons to surface calculations

adsorption energy:

$$E^{ads} = E^{mol@surf} - E^{mol} - E^{surf} + q\mu_e$$

chemical potential of electrons



Adding electrons to surface calculations

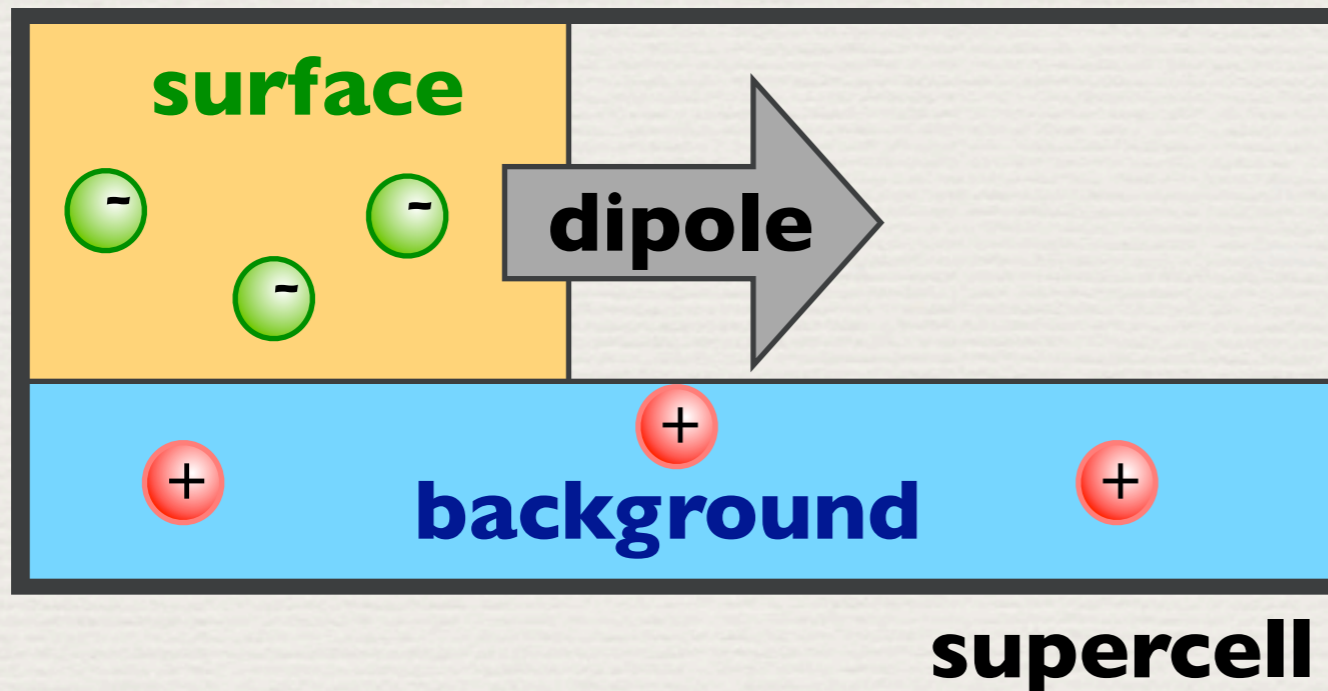
adsorption energy:

$$E^{ads} = E^{mol@surf} - E^{mol} - E^{surf} + q\mu_e$$

Adding electrons to supercells:

chemical potential of electrons

- requires compensating charge background



Adding electrons to surface calculations

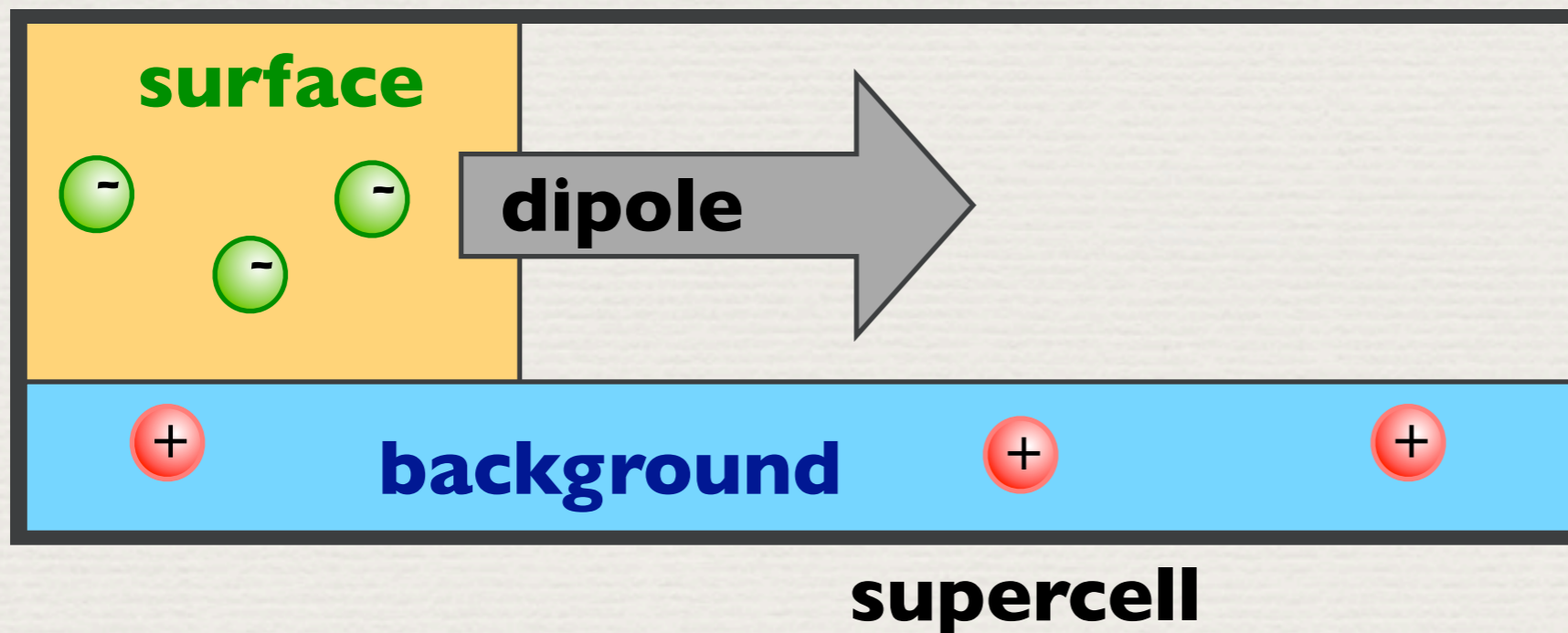
adsorption energy:

$$E^{ads} = E^{mol@surf} - E^{mol} - E^{surf} + q\mu_e$$

Adding electrons to supercells:

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Adding electrons to surface calculations

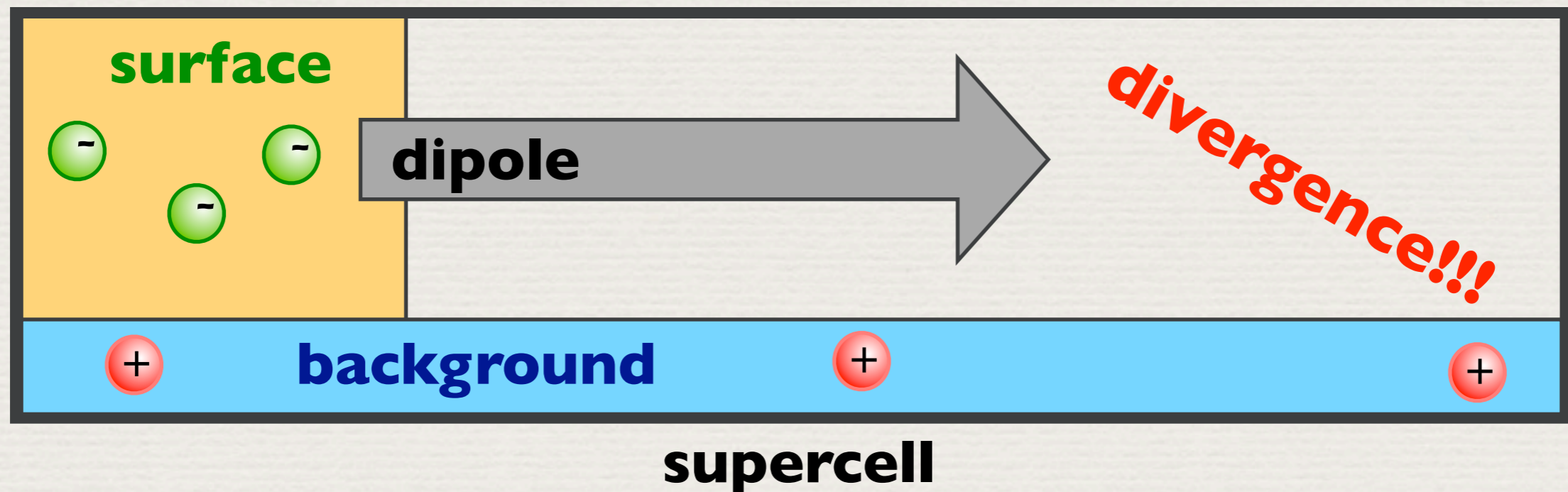
adsorption energy:

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Adding electrons to supercells:

chemical potential of electrons

- requires compensating charge background



Adding electrons to surface calculations

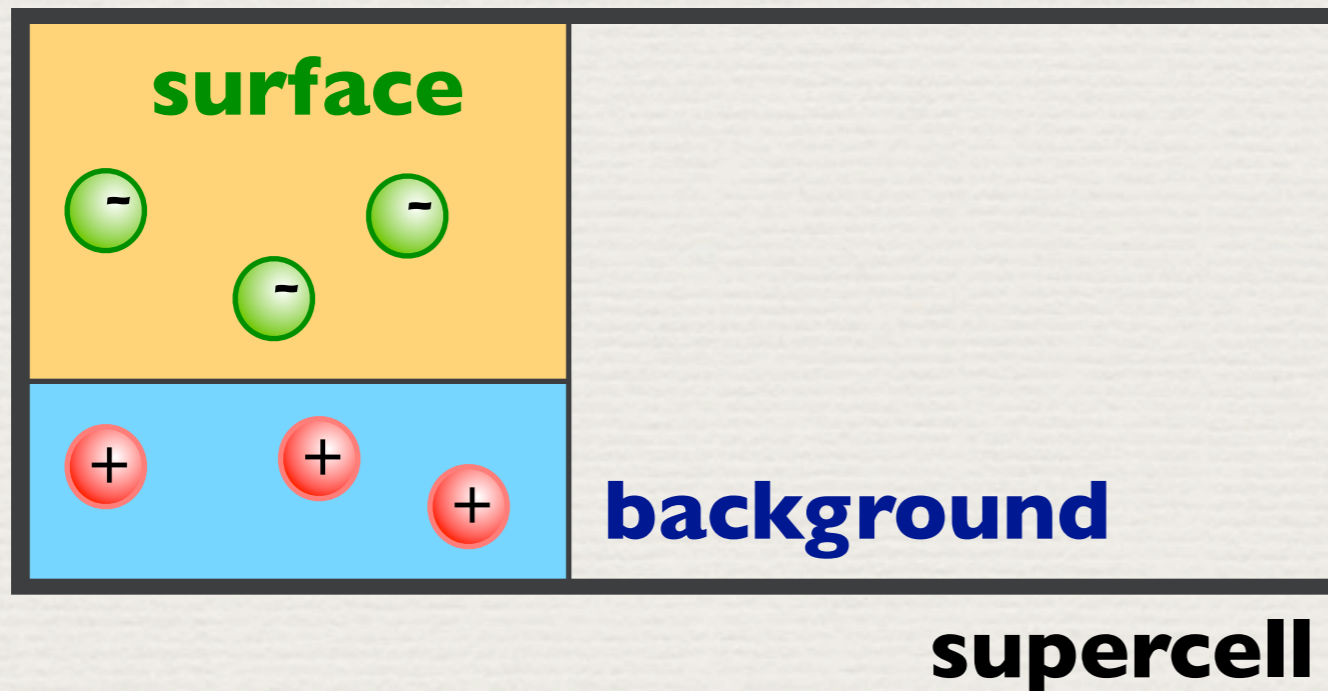
adsorption energy:

$$E^{ads} = E^{mol@surf} - E^{mol} - E^{surf} + q\mu_e$$

Adding electrons to supercells:

chemical potential of electrons

- we confine charge background (virtual crystal approximation)

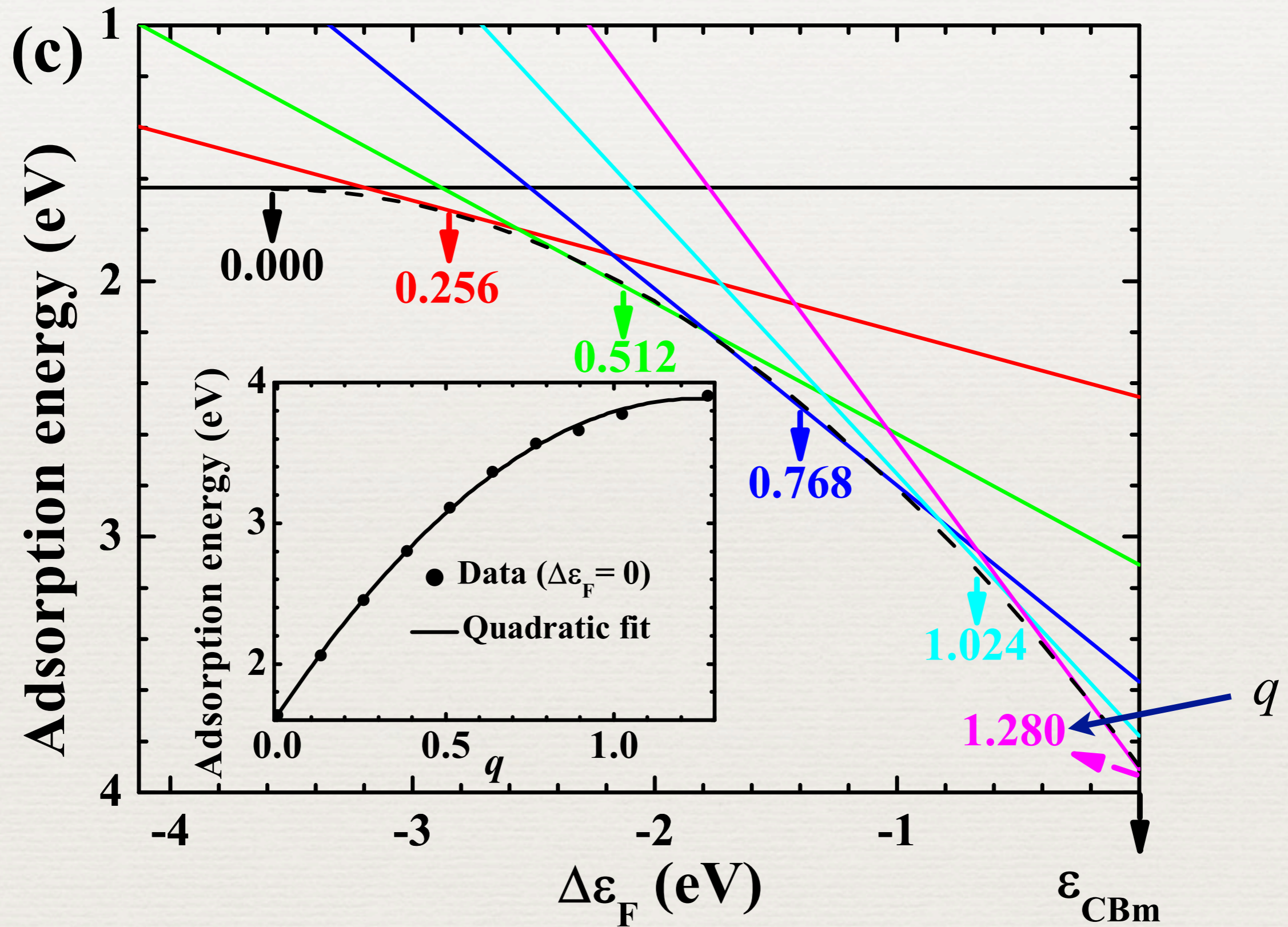


In our all-electron code:

- we change nuclear charge:

$$Z \rightarrow Z + \delta \quad (\delta \sim 10^{-2})$$

Adsorption energy as function of Fermi energy



Step 2: contribution from space charge region

Step 2a:

bulk doping concentration

- translate Fermi energy into doping dependence (standard semiconductor text book expressions)

$$\Delta\epsilon_F \rightarrow N_D$$

Step 2b:

- introduce space charge region (add band bending contribution)

surface area

supercell length

$$\Delta E_q^{\text{ads}}(N_D) = E_{\text{ads}}^{\text{DFT}}(q) - \underbrace{\frac{e^2}{6\epsilon\epsilon_0 N_D A^2} |q|^3}_{\text{macroscopic band bending}} + \underbrace{\frac{e^2 d}{6\epsilon\epsilon_0 A} q^2}_{\text{spurious band bending in calculation}}$$

**macroscopic
band bending**

**spurious band
bending
in calculation**

Step 2: contribution from space charge region

Step 2a:

bulk doping concentration

- translate Fermi energy into doping dependence (standard semiconductor text book expressions)

$$\Delta\epsilon_F \rightarrow N_D$$

Step 2b:

- introduce space charge region (add band bending contribution)

surface area

supercell length

$$\Delta E_q^{\text{ads}}(N_D) = E_{\text{ads}}^{\text{DFT}}(q) - \underbrace{\frac{e^2}{6\epsilon\epsilon_0 N_D A^2} |q|^3}_{\text{surface area}} + \underbrace{\frac{e^2 d}{6\epsilon\epsilon_0 A} q^2}_{\text{supercell length}}$$

Step 2c:

- maximize $\Delta E_q^{\text{ads}}(N_D)$ for optimal charge transfer Q

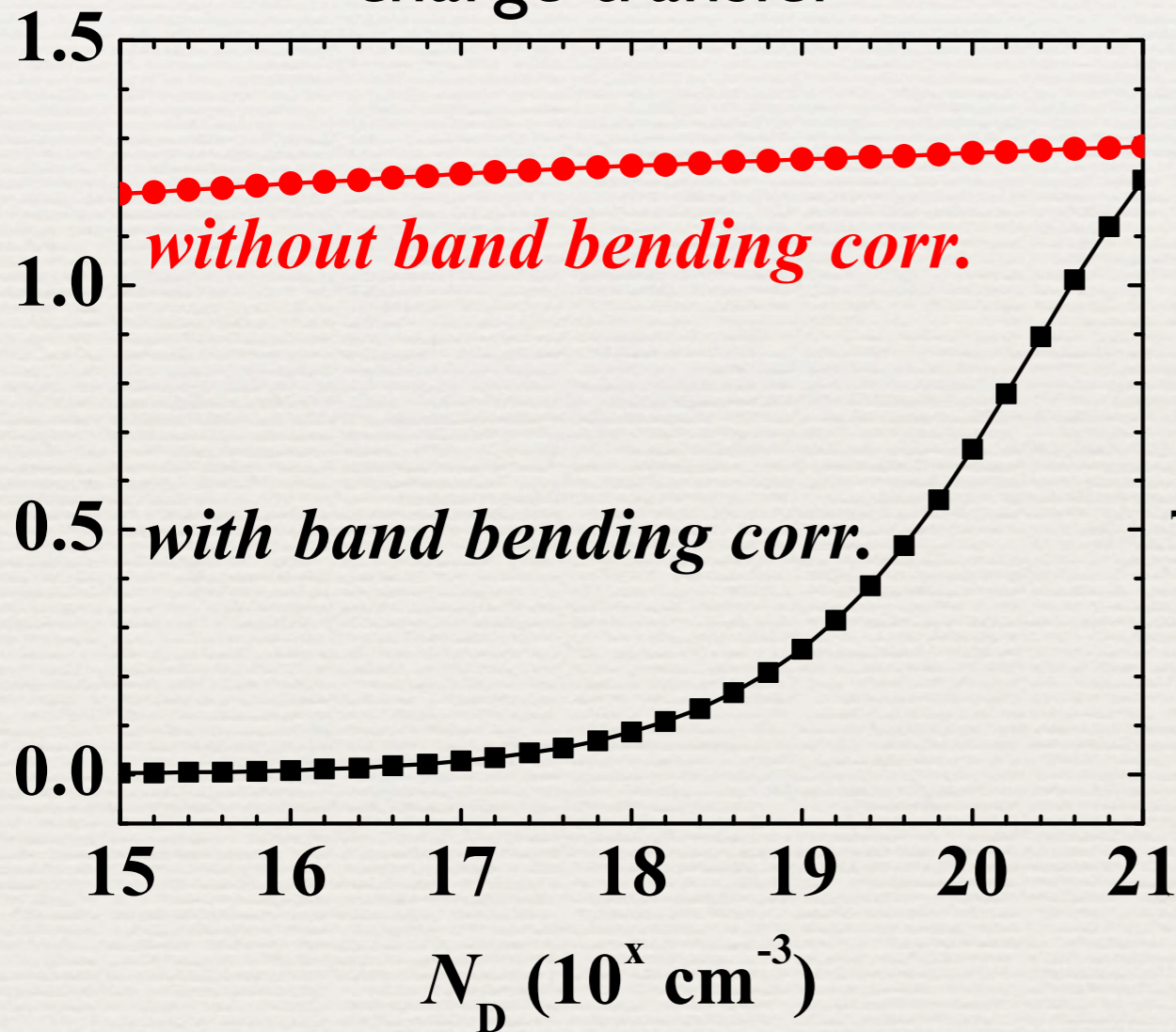
Introducing bulk doping into 1st principles

$$E^{ads}(N_D) = \underbrace{E^{mol@surf} - E^{mol} - E^{surf}}_{\text{microscopic: DFT}} + \underbrace{\Delta_{BB}(Q, N_D)}_{\text{macroscopic model}}$$

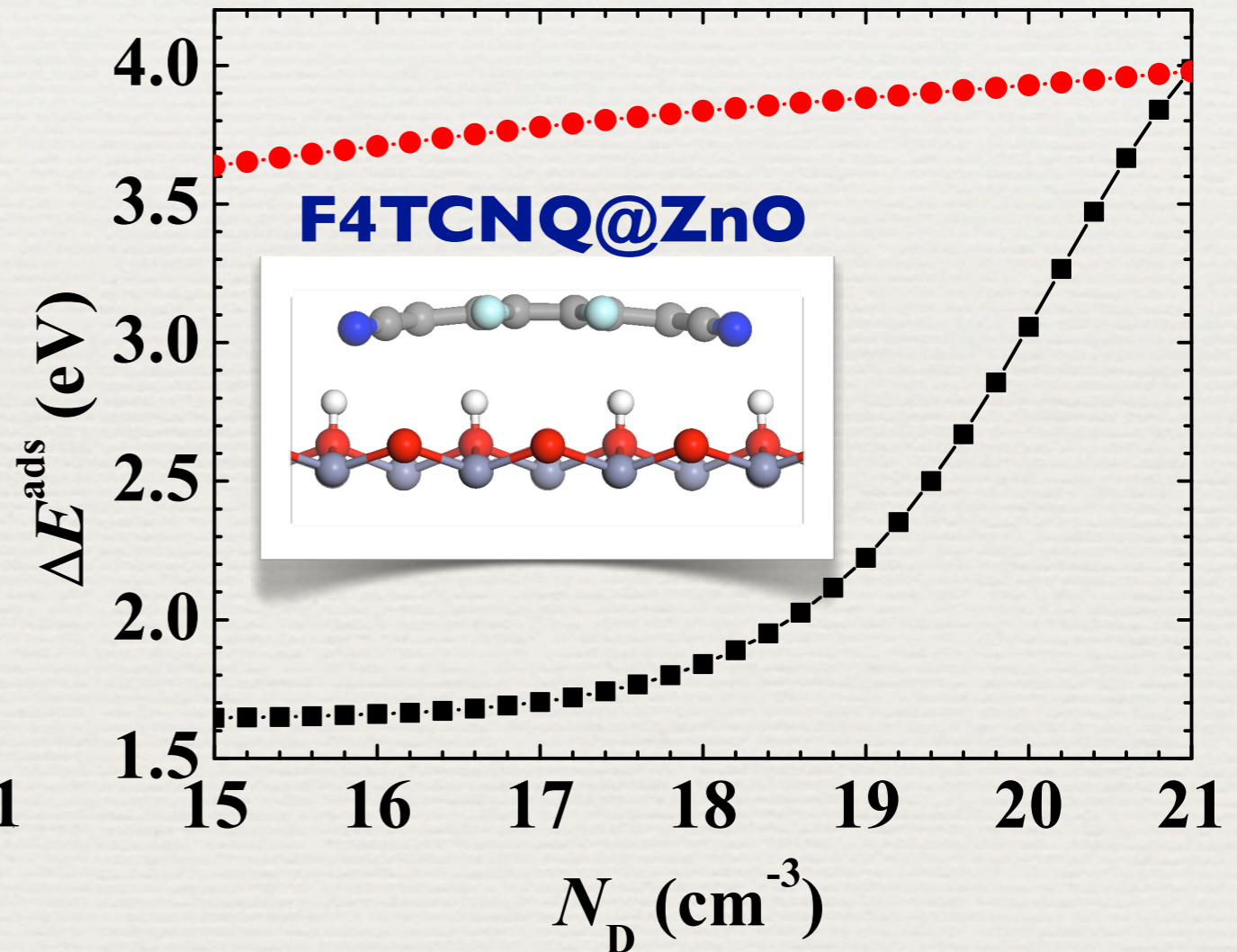
microscopic: DFT

macroscopic model

charge transfer



adsorption energy



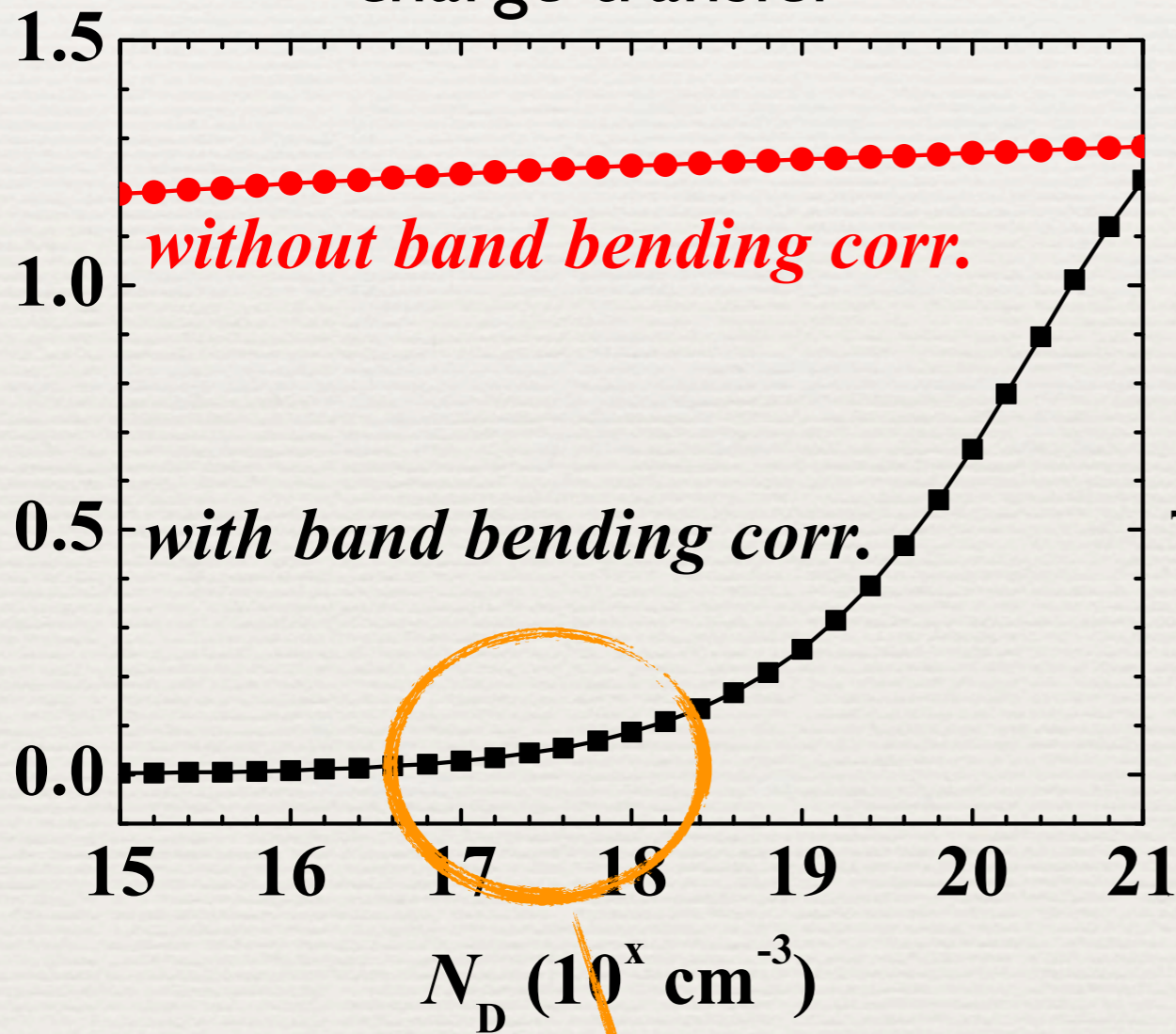
Introducing bulk doping into 1st principles

$$E^{ads}(N_D) = \underbrace{E^{mol@surf} - E^{mol} - E^{surf}}_{\text{microscopic: DFT}} + \underbrace{\Delta_{BB}(Q, N_D)}_{\text{macroscopic model}}$$

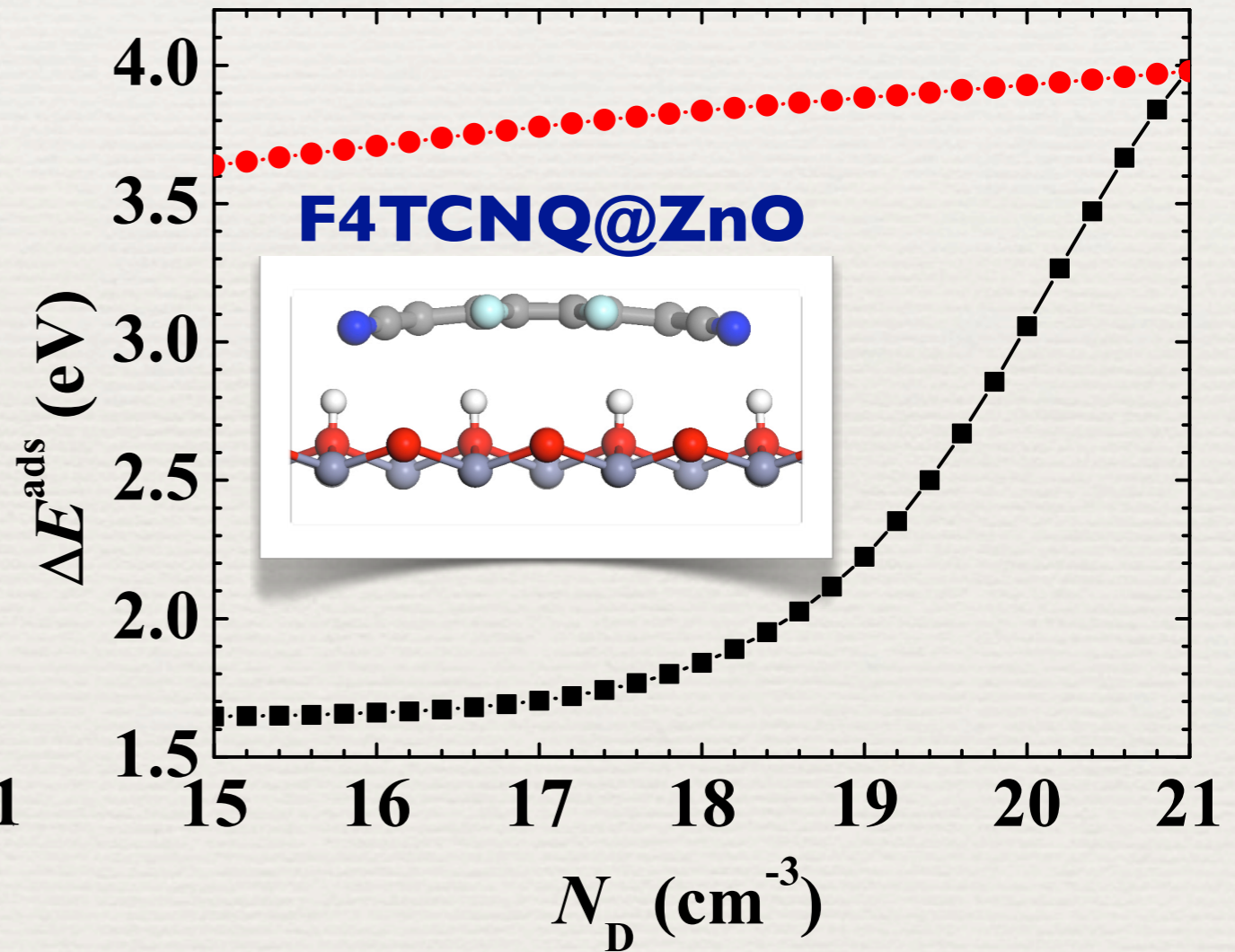
microscopic: DFT

macroscopic model

charge transfer



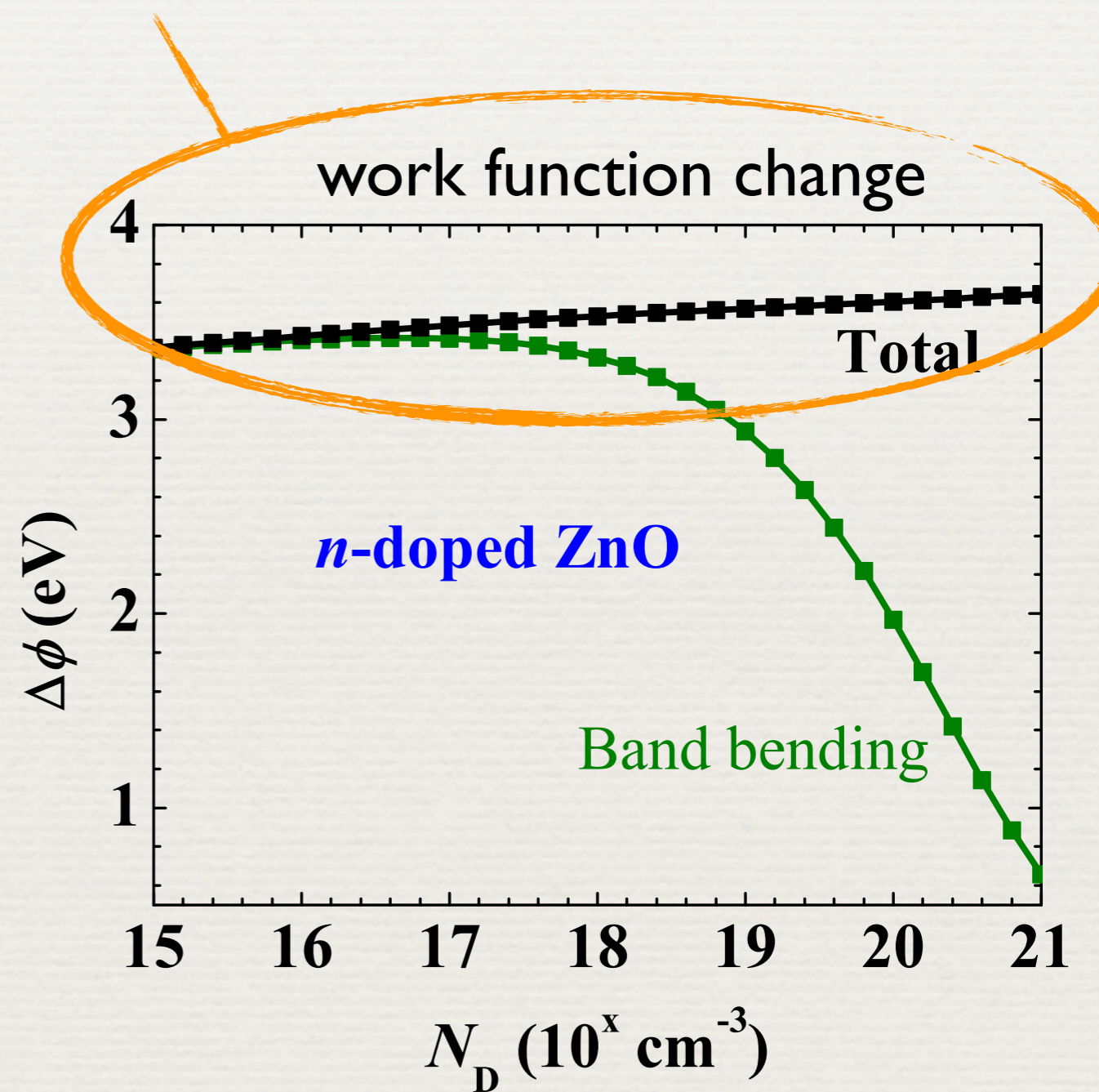
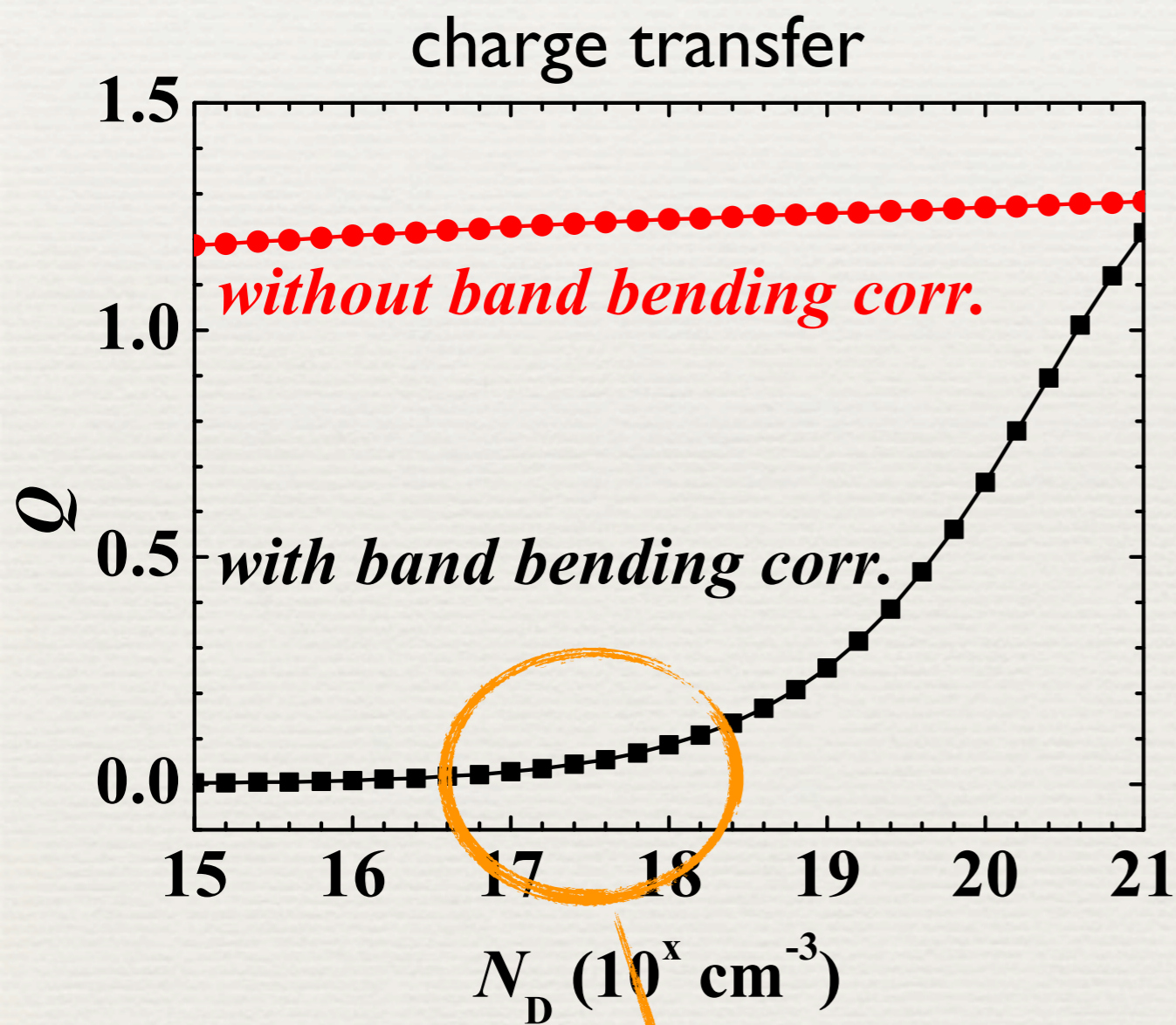
adsorption energy



• **small charge transfer**

Adsorption energy and charge transfer

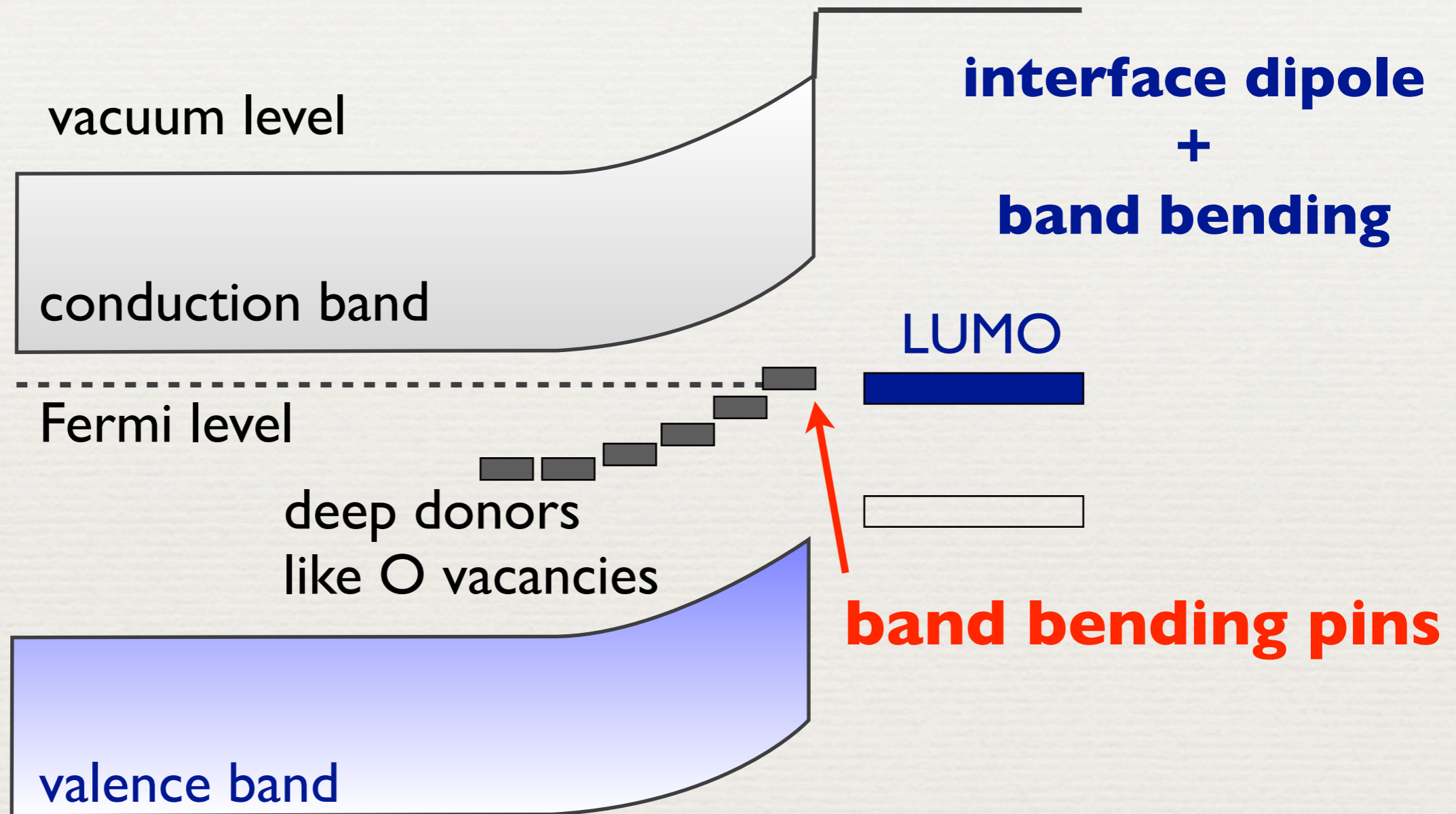
• large work function increase



• small charge transfer

Formation of space-charge layer

low doping + deep donors \Rightarrow band bending pins

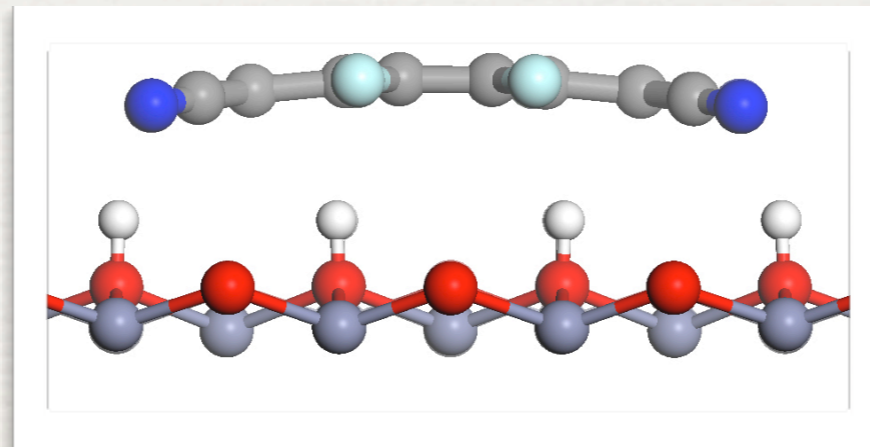


Work function increase in photoemission

photoemission (UPS/XPS) in collaboration with:

Humboldt University Berlin

R. Schlesinger
J. Frisch
J. Niederhausen
S. Blumstengel
F. Henneberger
N. Koch



**Synchrotron -
BESSY II**

S. Winkler
A. Vollmer

Experiment

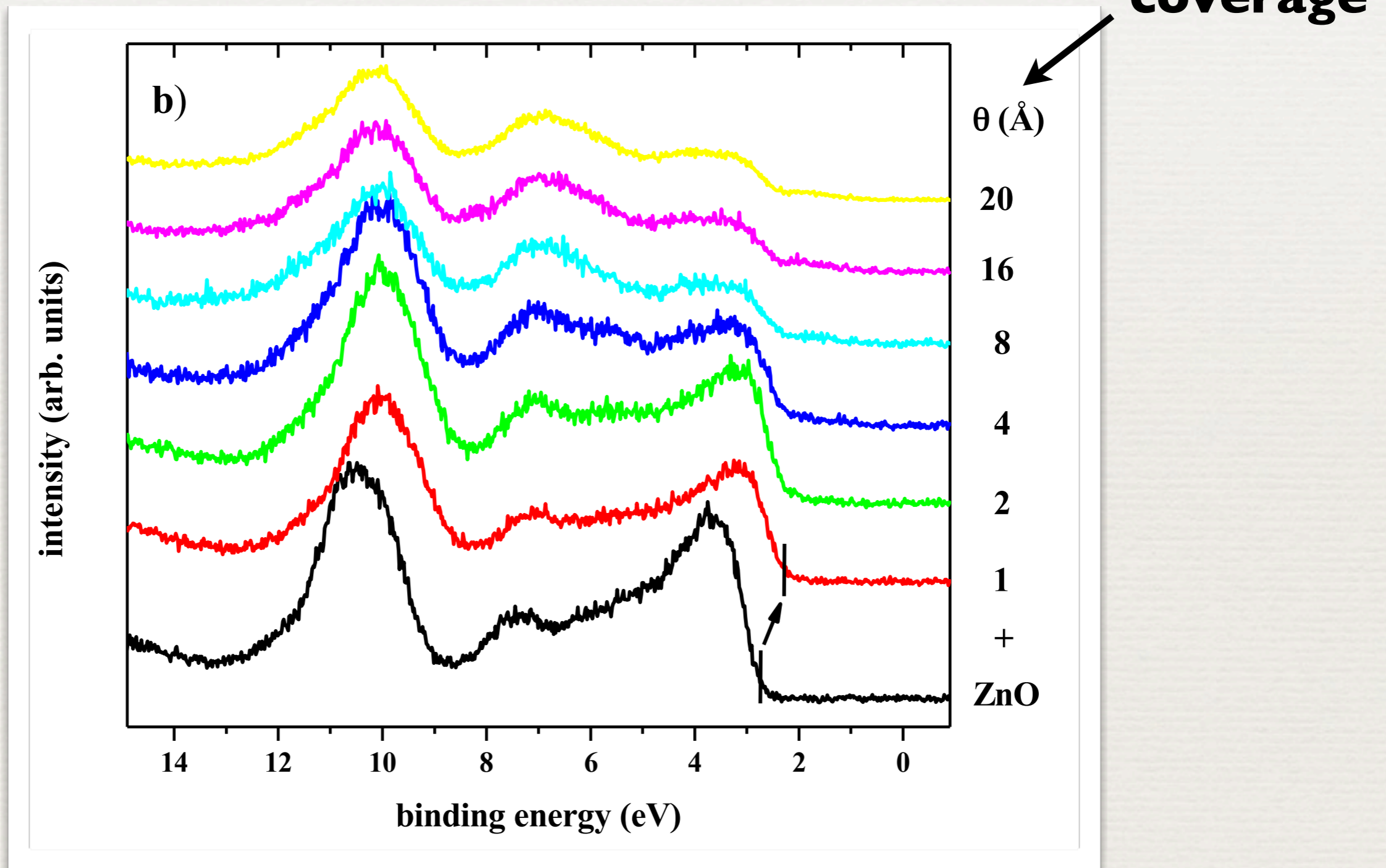
- work function increase: 1.4 eV
- minimal charge transfer



**Collaborative
Research Center
“Hybrid Inorganic/
Organic Systems”**

F4TCNQ@ZnO - photoemission

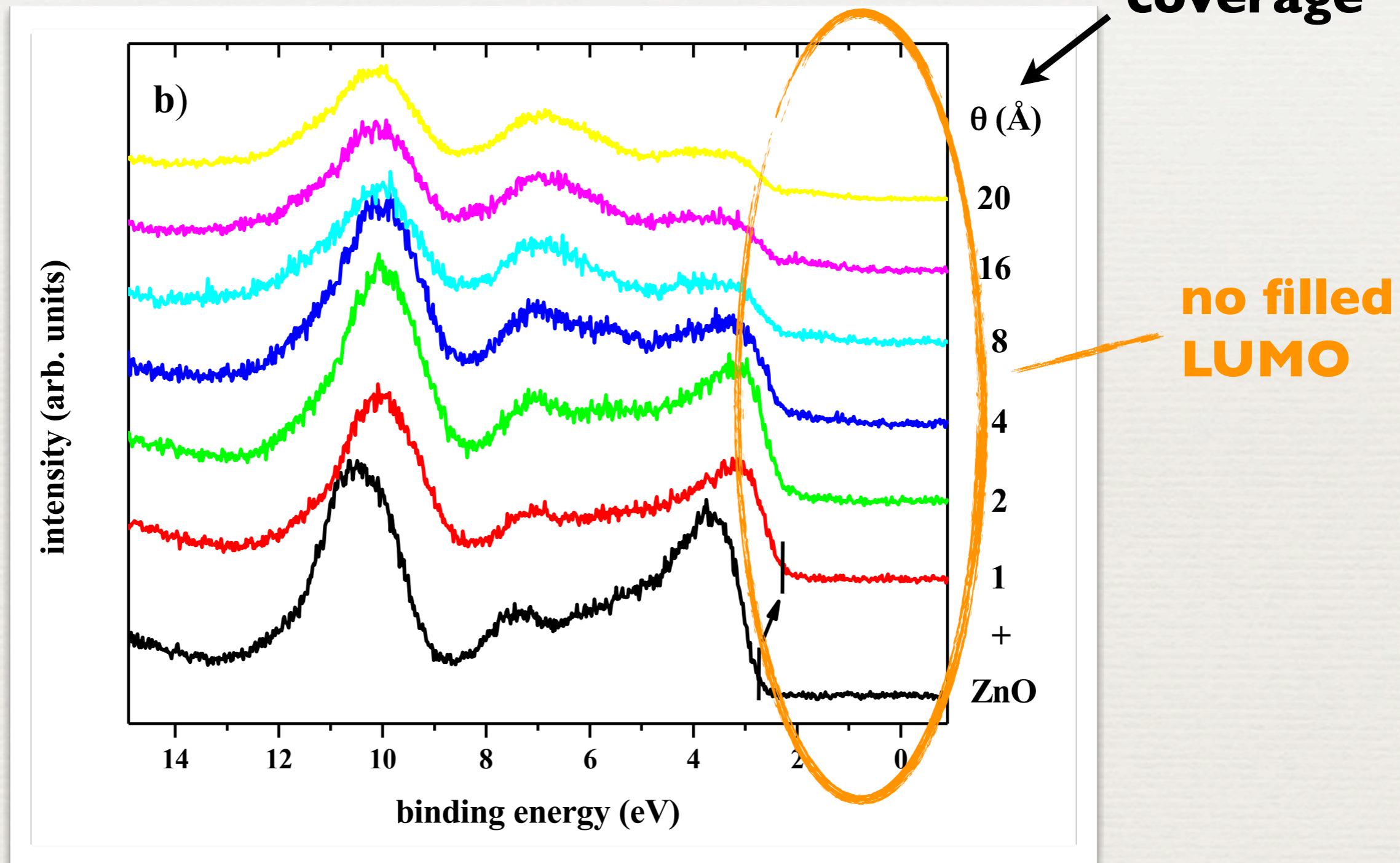
F4TCNQ@ZnO(00-10)



similar spectra for ZnO(0010)

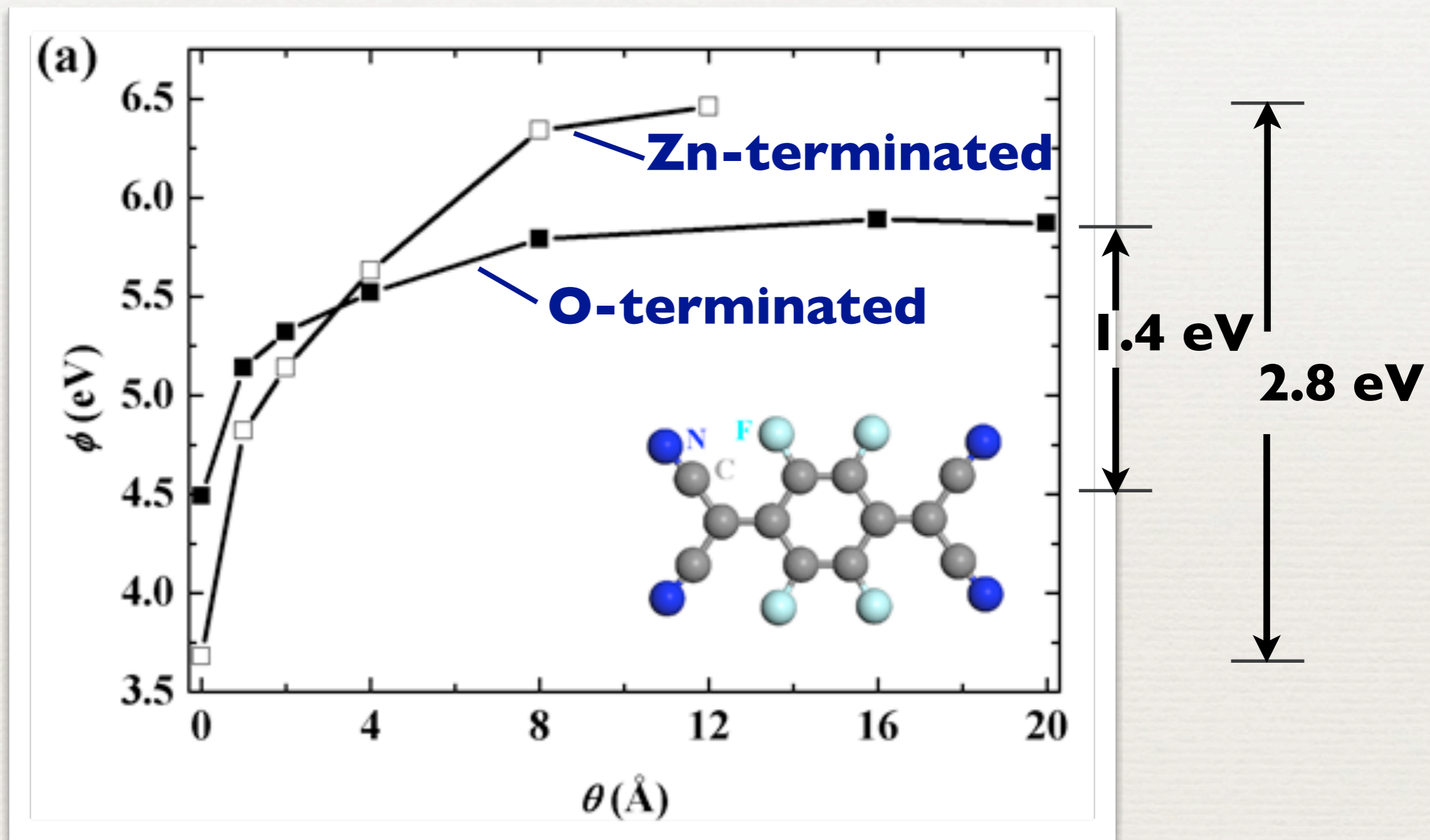
F4TCNQ@ZnO - photoemission

F4TCNQ@ZnO(00-10)



R. Schlesinger, et al., Phys. Rev. B **87**, 155311 (2013)

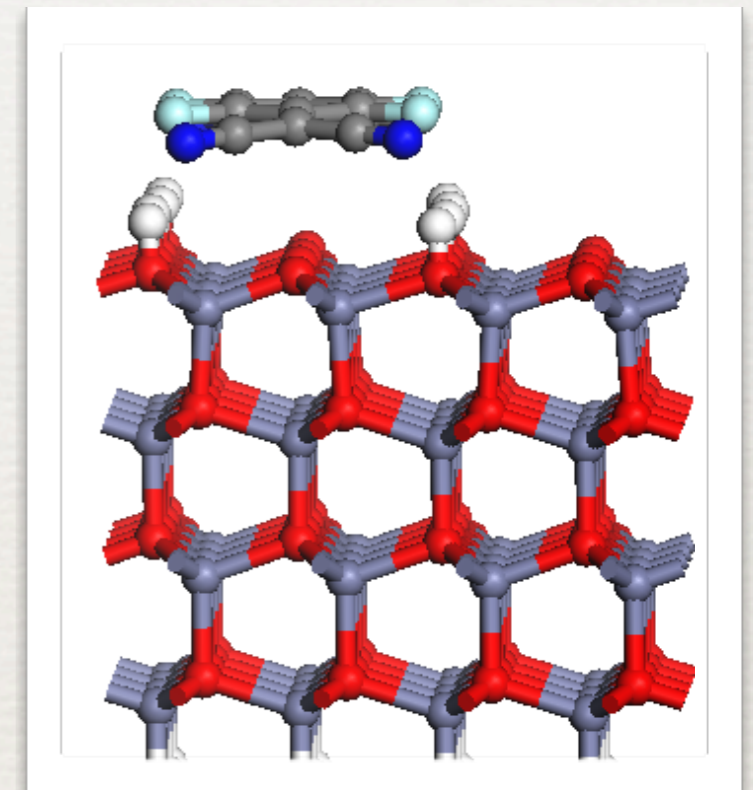
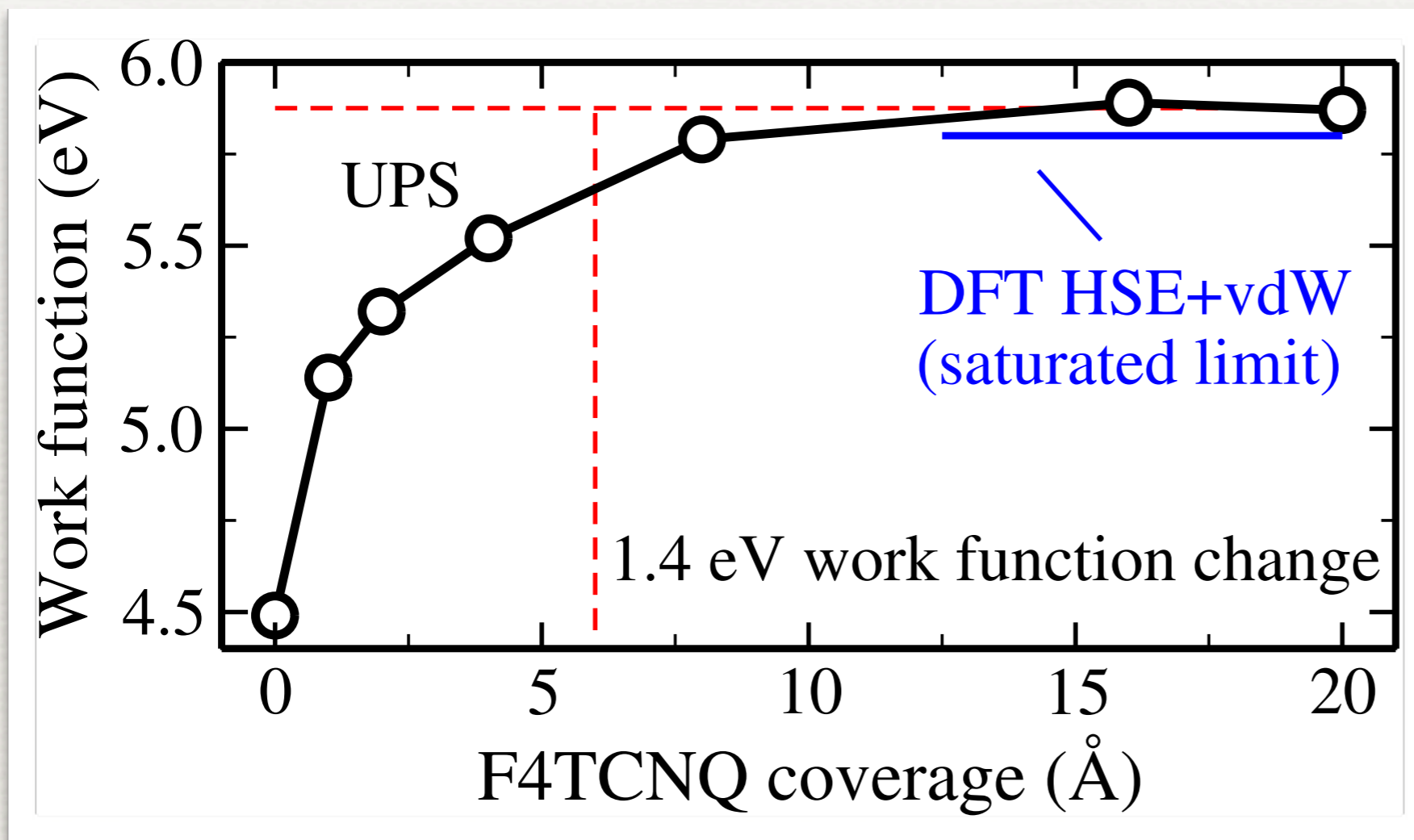
F4TCNQ@ZnO - photoemission



- **extraordinarily large work function changes**

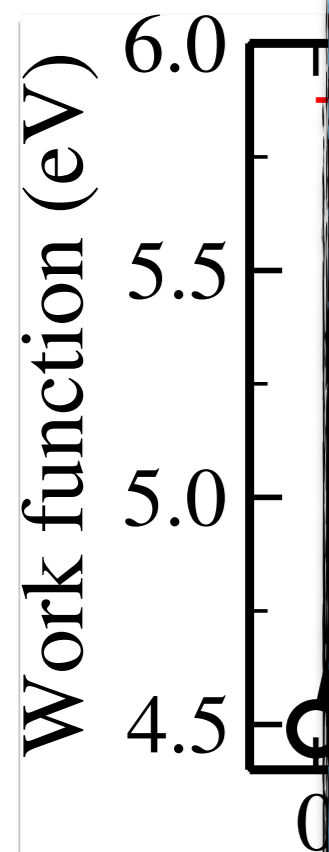
R. Schlesinger, et al., Phys. Rev. B **87**, 155311 (2013)

Workfunction of F4TCNQ@ZNO

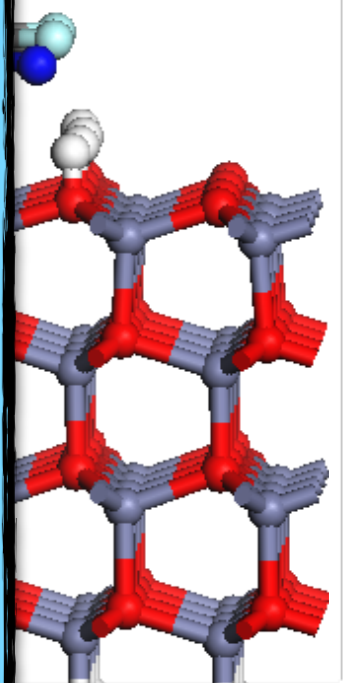


- DFT-HSE workfunction in good agreement with UPS

Workfunction of F4TCNQ@ZNO



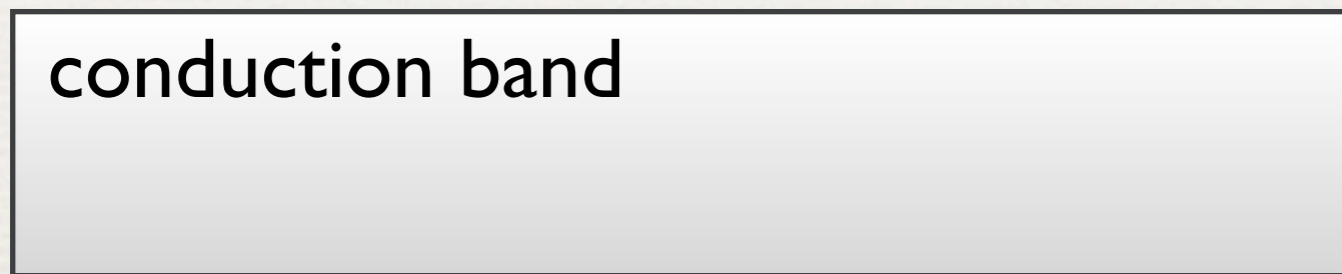
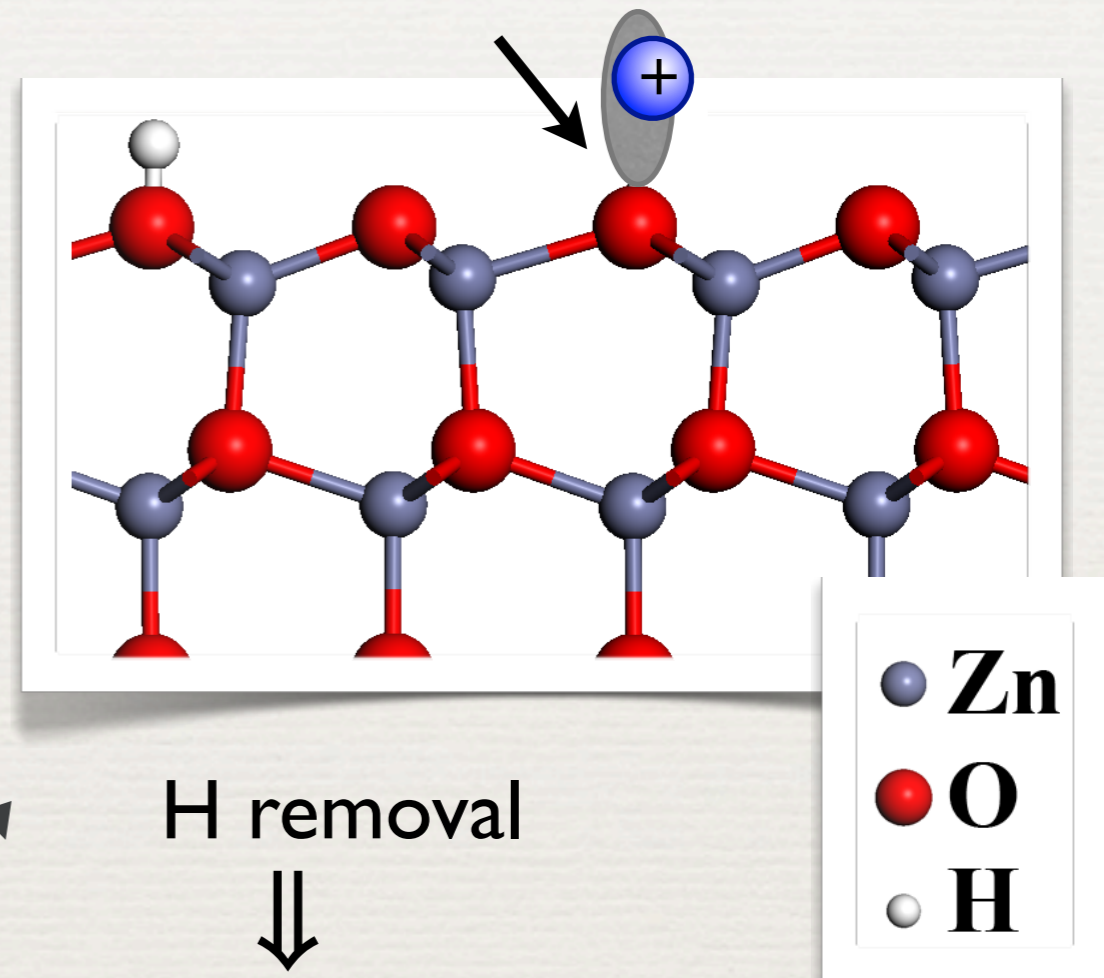
- work function change:
 - theory: ~ 3.5 eV
 - exper.: ~ 1.4 eV
- potential reasons for deviation:
 - photo voltage effect
 - deep defects pin band bending
 - hydrogen concentration at interface differs



- DFT-ISE workfunction in good agreement with UPS

H-deficient ZnO-O 2x1-H surface

O dangling bond

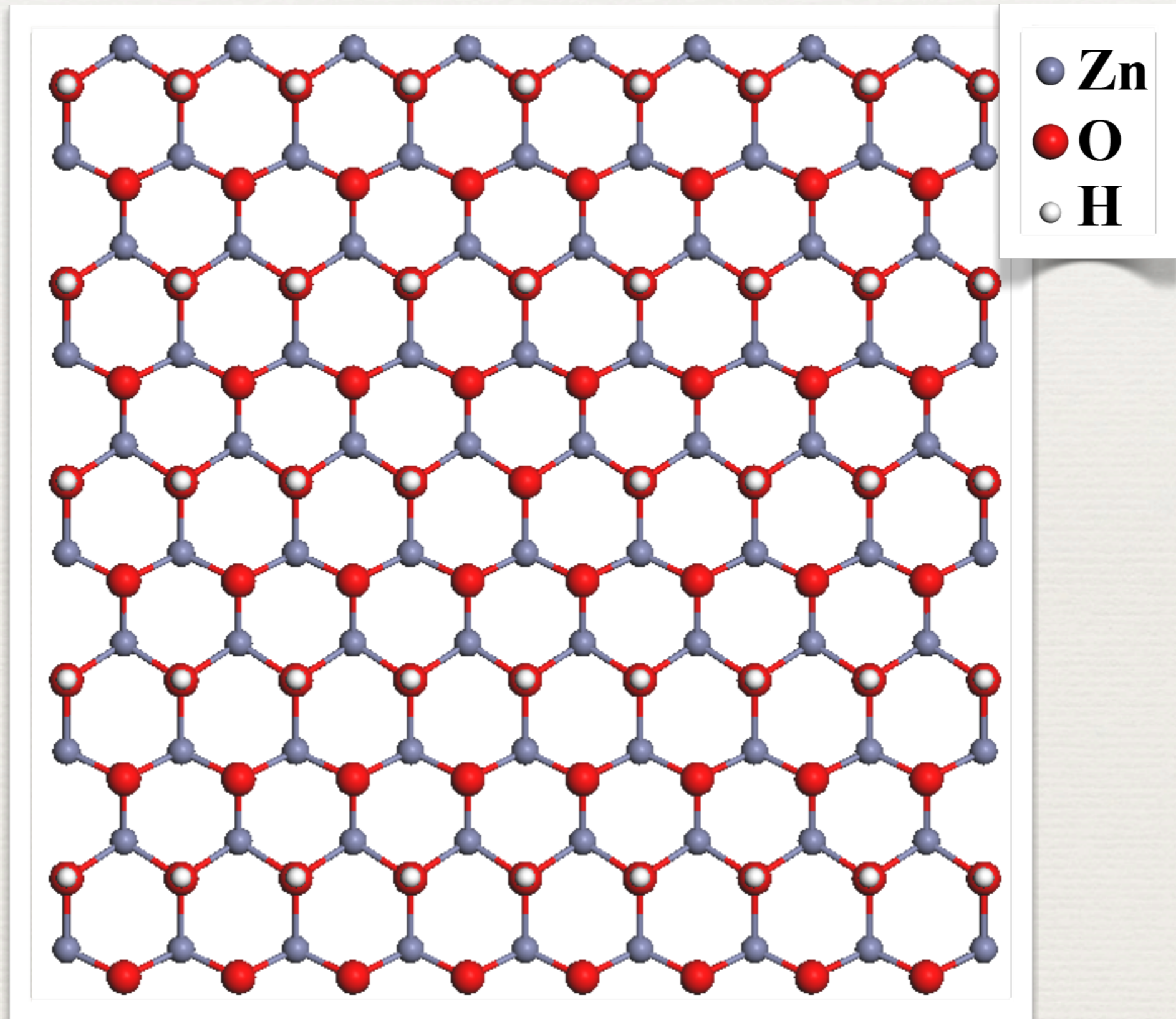


H removal

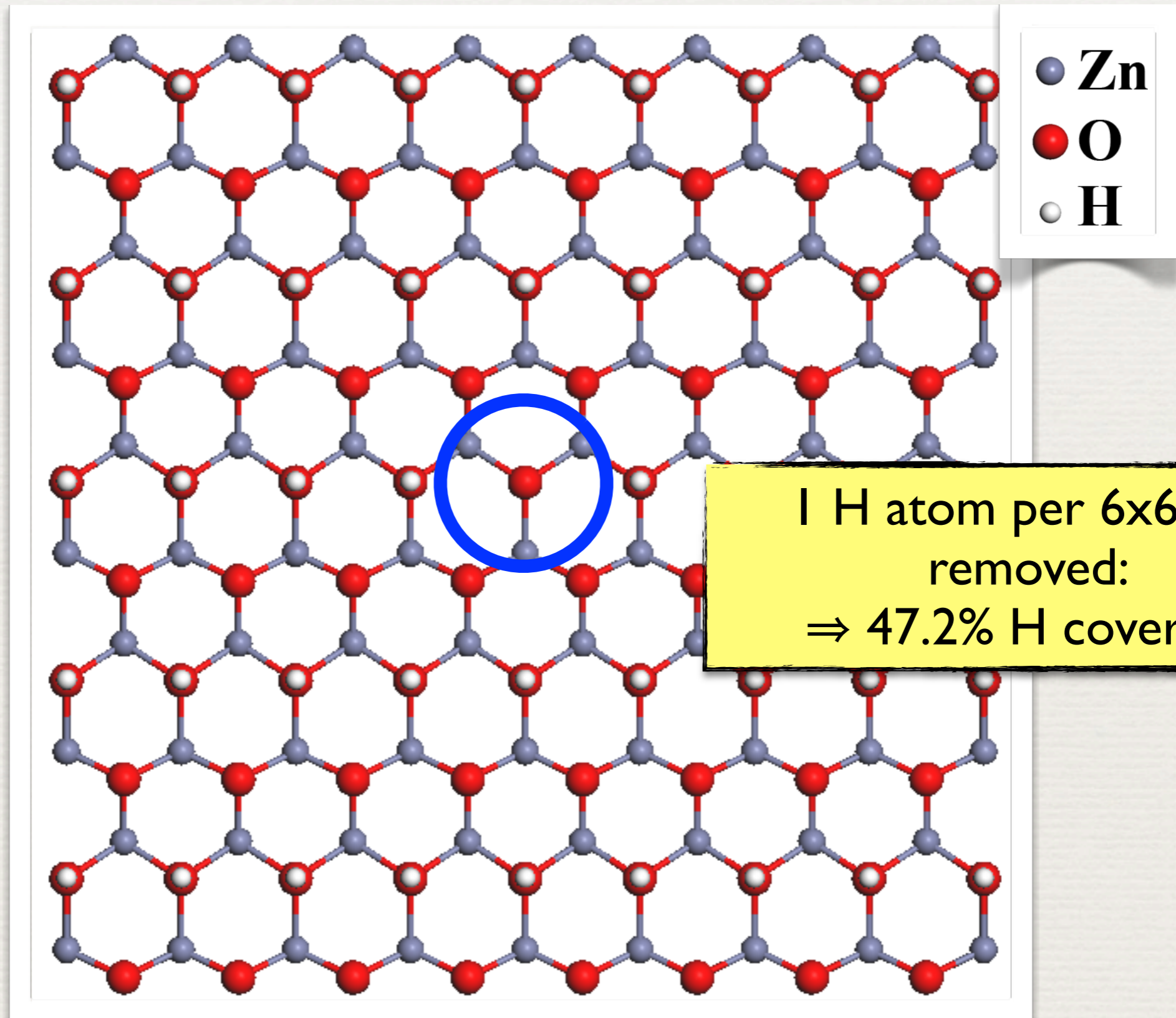


□ O dangling bonds

H-deficient ZnO-O 2x1-H surface

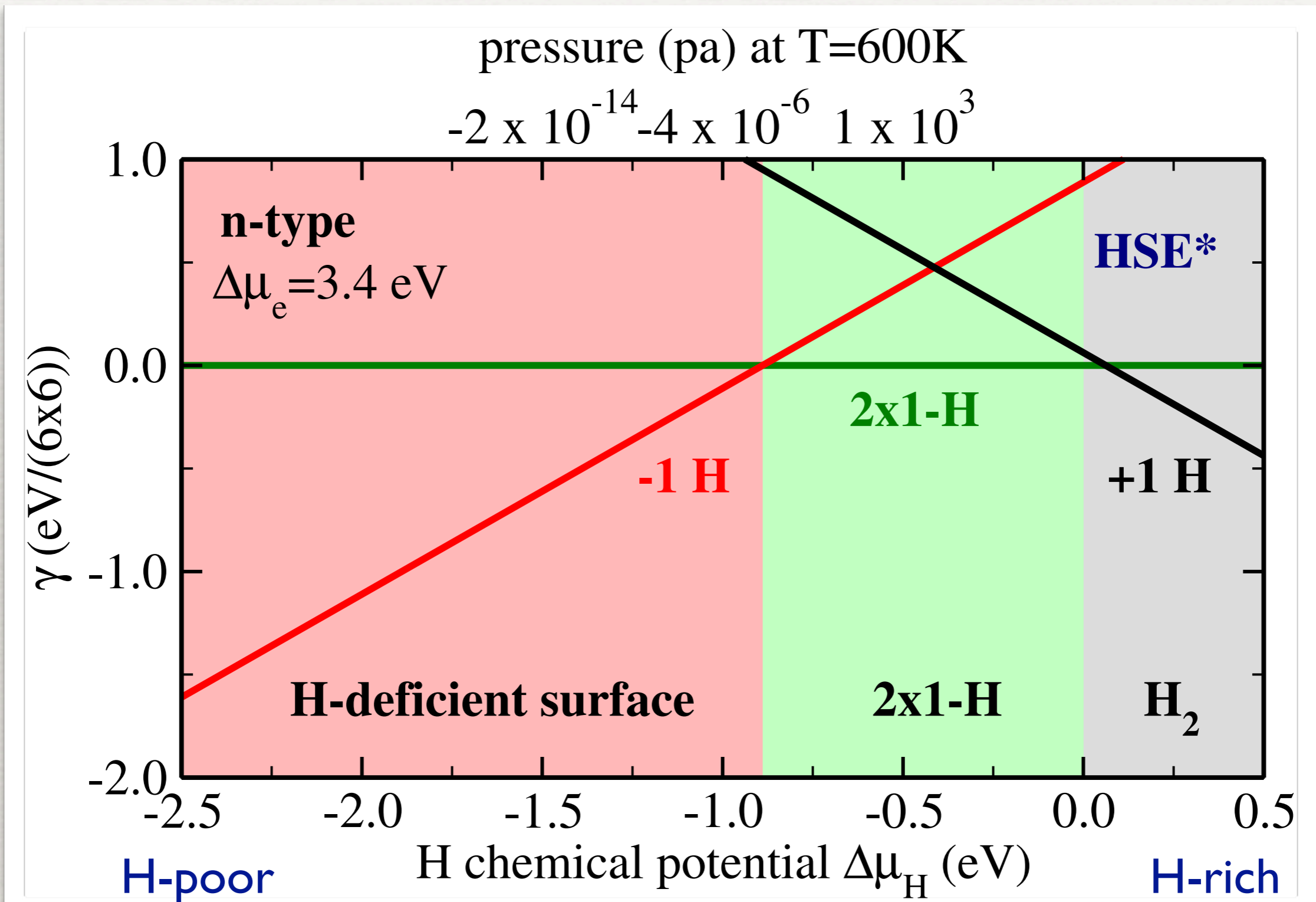


H-deficient ZnO-O 2x1-H surface



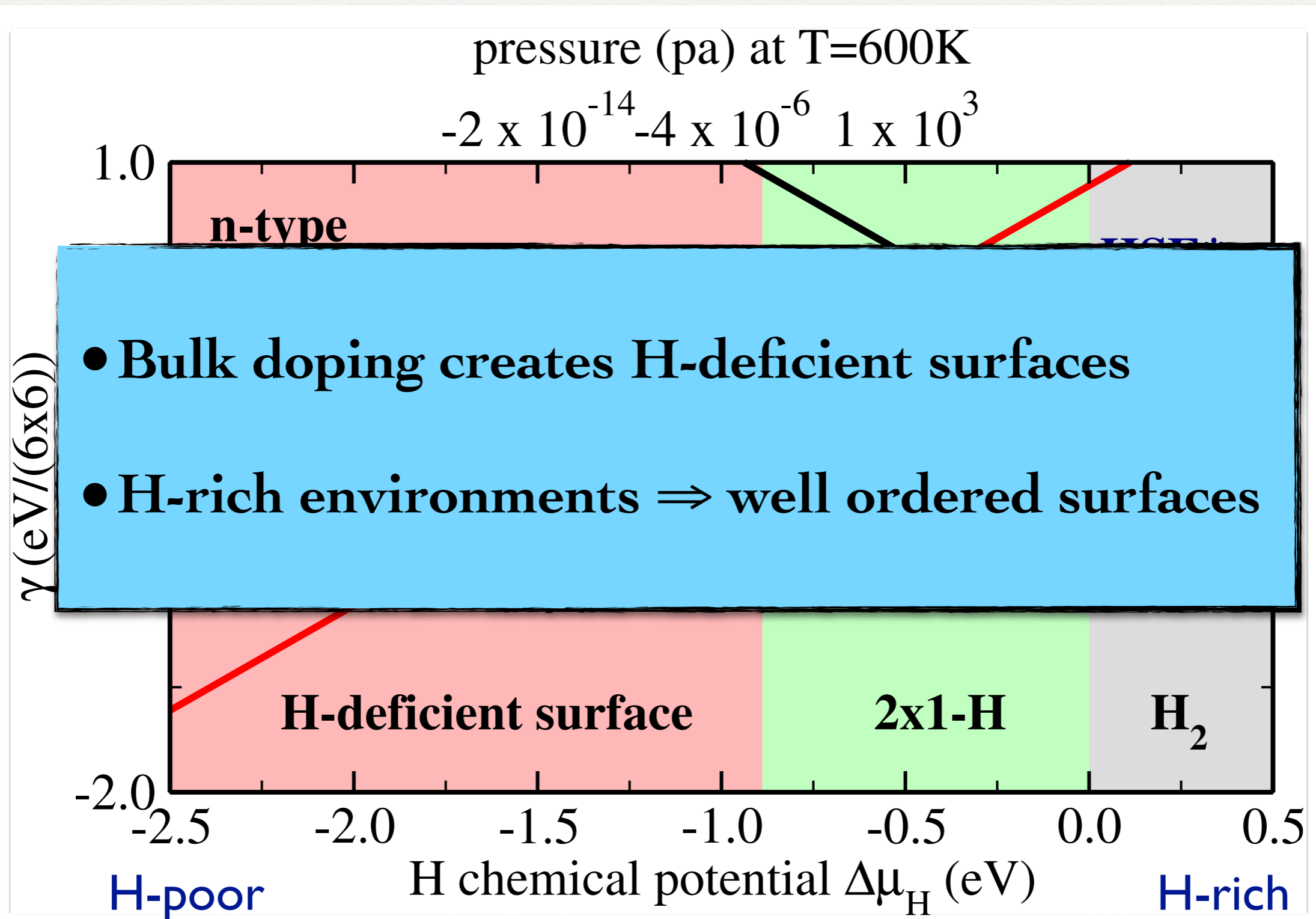
ZnO-O in contact with H-reservoir

- H-deficient surfaces stabilized by n-type conditions

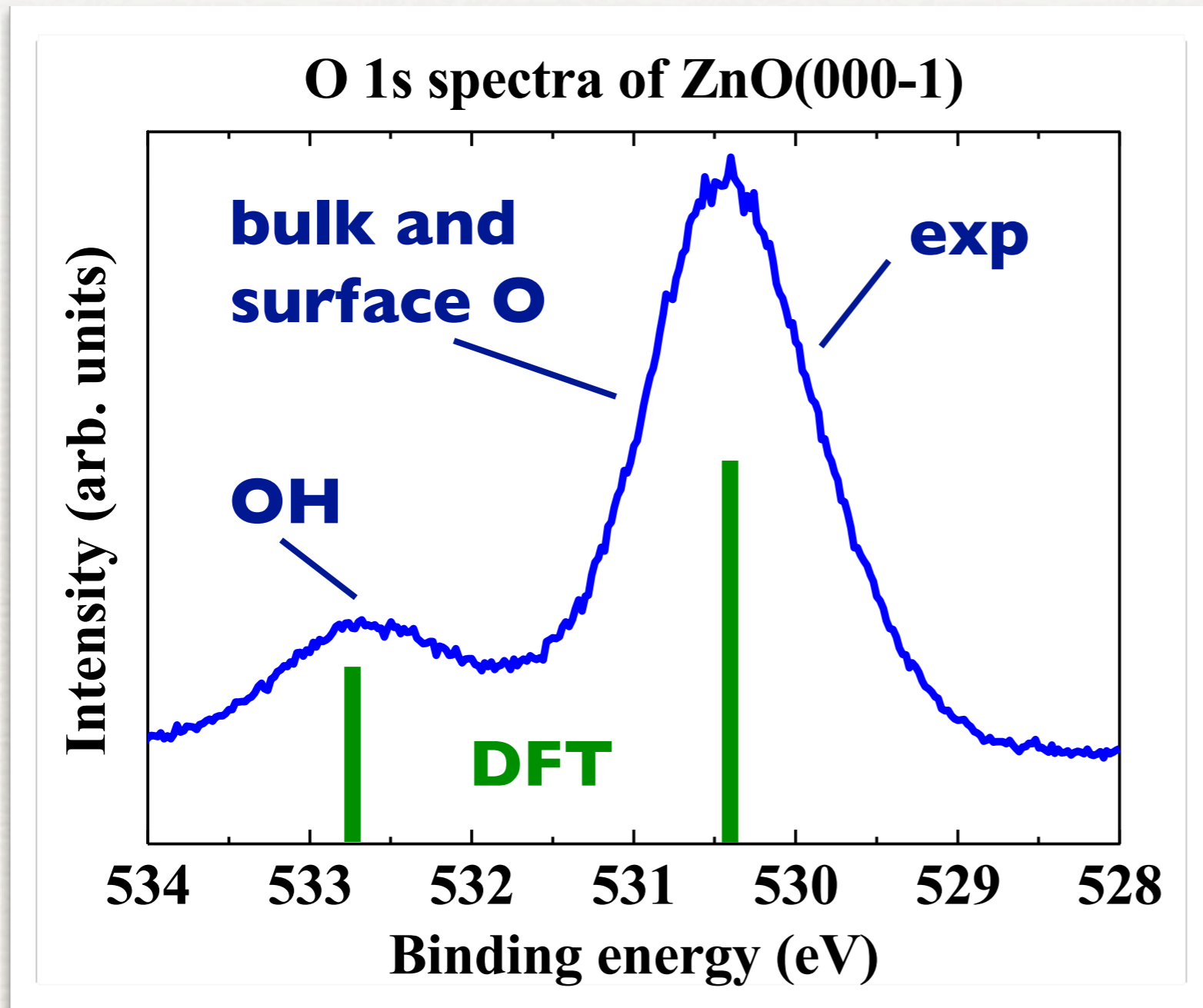


ZnO-O in contact with H-reservoir

- H-deficient surfaces stabilized by n-type conditions



Experimental evidence



	surf. core level shift
Exp.	2.0 eV
DFT	2.3 eV

peak weights:

- H-coverage ~30-40%

PBE Δ self-consistent field (SCF) calculations reveal:

- **only H-terminated structures match exp. core level shifts**

HIOS are fascinating systems

Charge transfer

Level alignment

Van der Waals interaction

Surface structure

Adsorption geometry

Defects

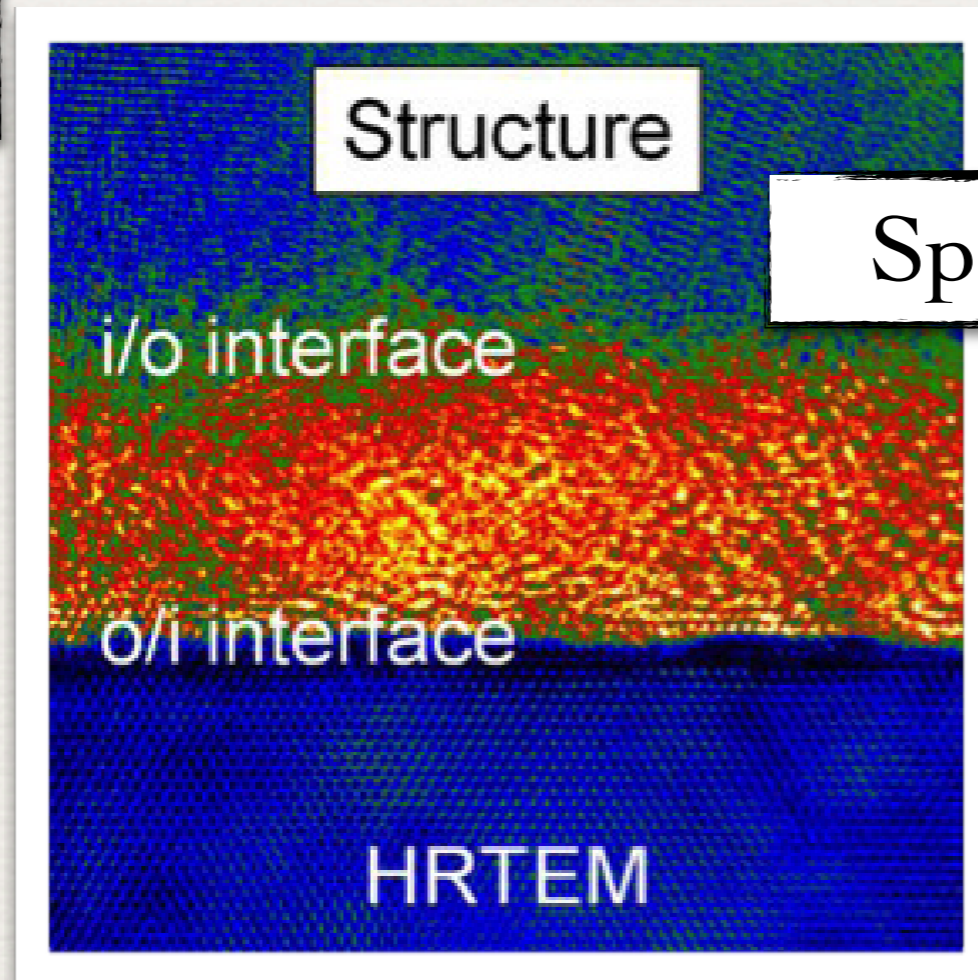
Structure

Space-charge layers

i/o interface

o/i interface

HRTEM



Atomistic understanding of HIOS

Charge transfer

Level alignment

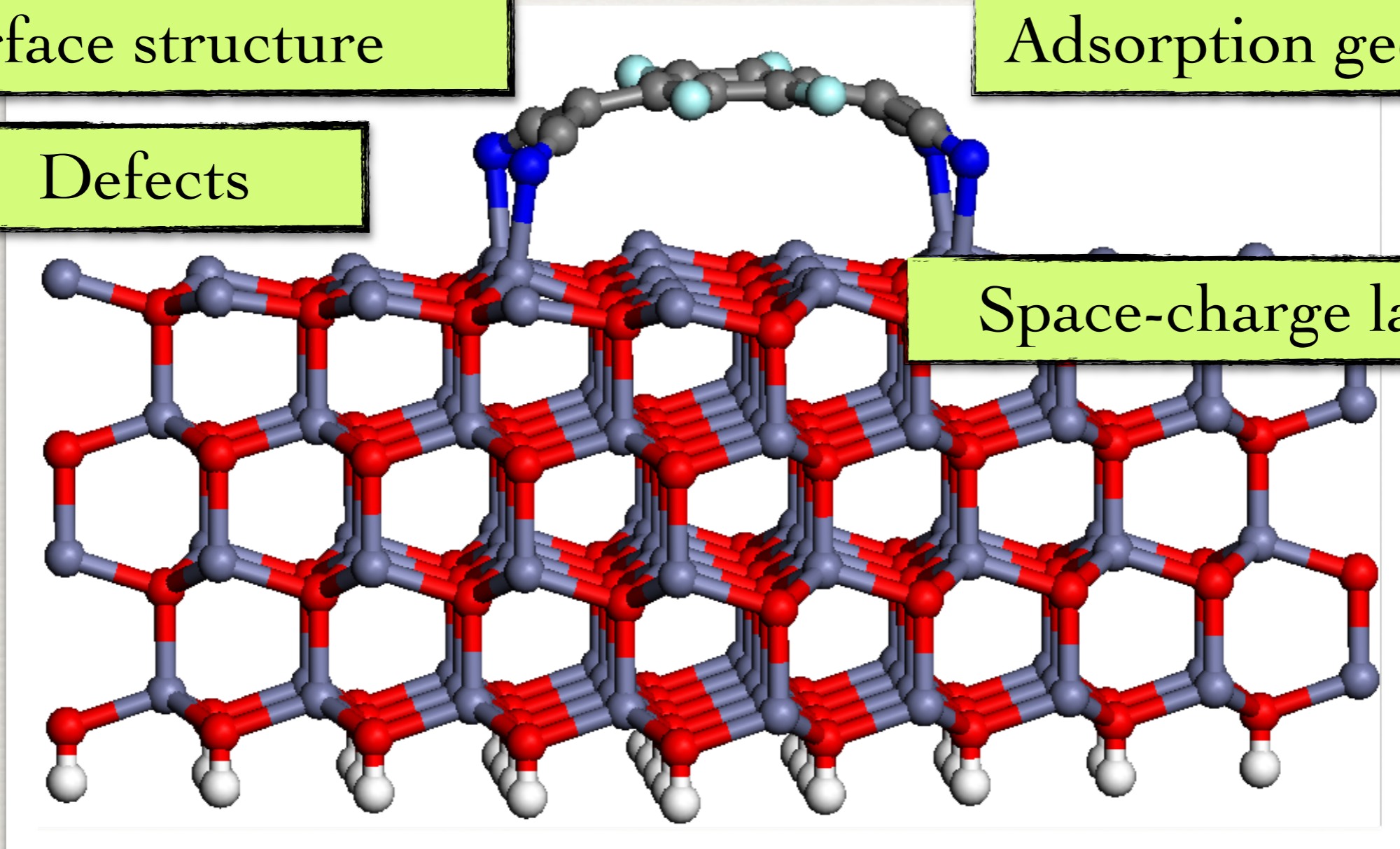
Van der Waals interaction

Surface structure

Adsorption geometry

Defects

Space-charge layers



Atomistic understanding of molecules@surfaces

Charge transfer

Level alignment

Van der Waals interaction

Surface structure

Adsorption geometry

Defects

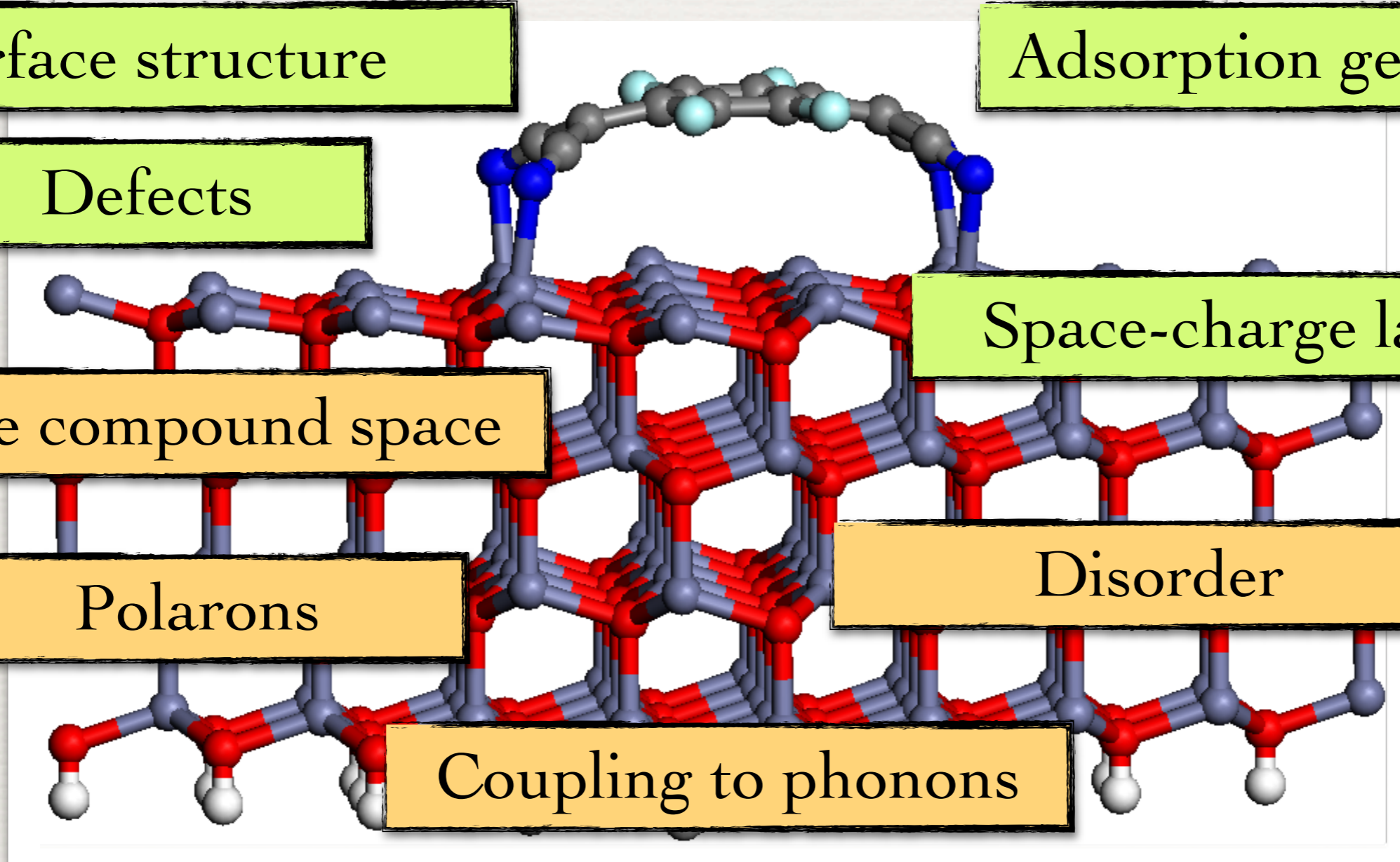
Space-charge layers

Large compound space

Polarons

Disorder

Coupling to phonons



Acknowledgements

Fritz-Haber-Institut, Berlin

Yong Xu
Oliver Hofmann
Norina Richter
Sergey Levchenko
Matthias Scheffler

IBM Research Zurich

Nikolaj Moll

Humboldt University Berlin

R. Schlesinger
J. Frisch
J. Niederhausen
S. Blumstengel
F. Henneberger
N. Koch

Helmholtz-Center Berlin - BESSY II

S. Winkler
A. Vollmer

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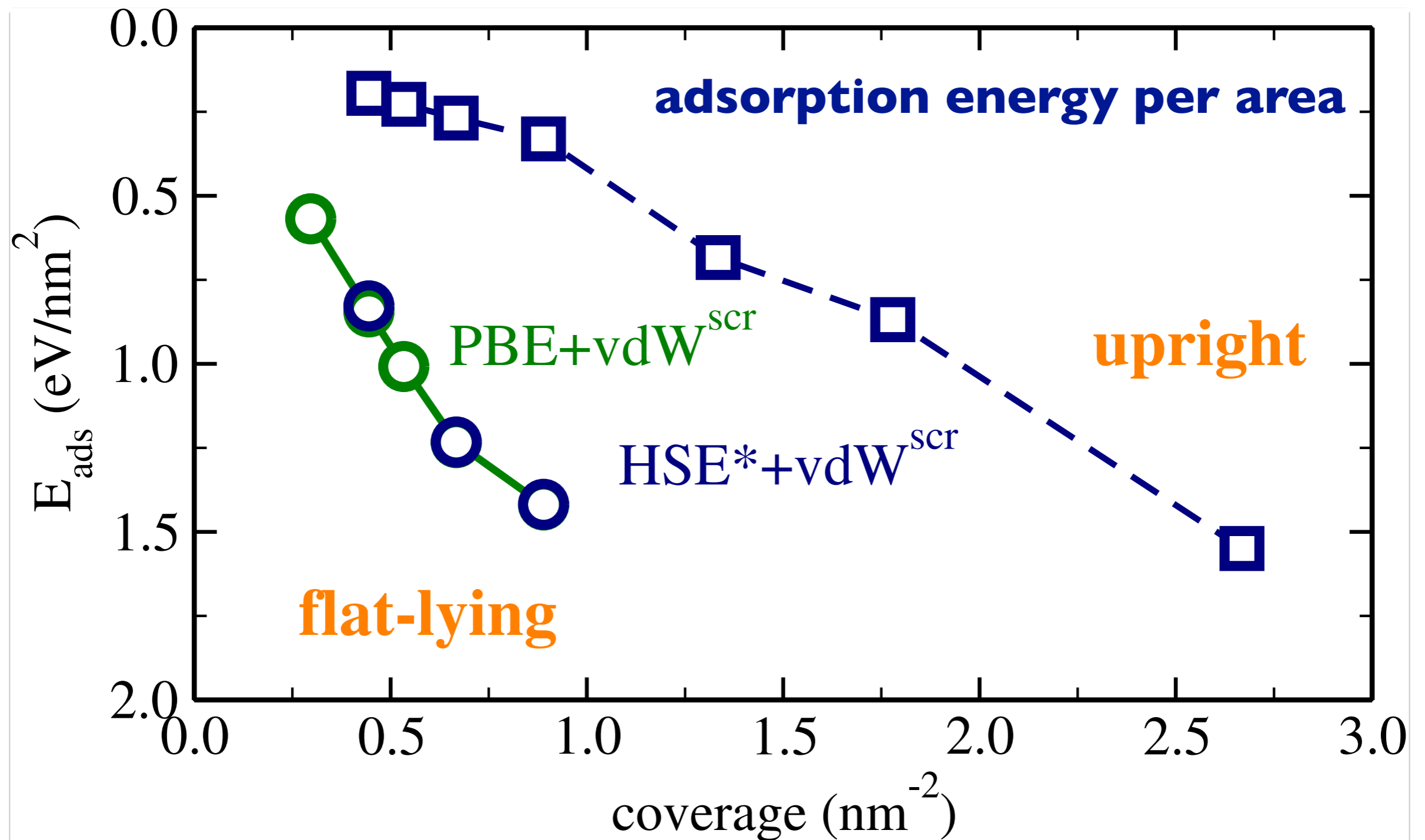


Alexander von Humboldt
Stiftung/Foundation



Thank you!

Adsorption geometries and energies



- monolayer is composed of flat-lying molecules
- molecular orientation in layer random: disorder (no impact on electronic structure)

Step 2: contribution from space charge region

Step 2a:

bulk doping concentration

- translate Fermi energy into doping dependence
(standard semiconductor text book expressions)

$$\Delta\epsilon_F \rightarrow N_D$$

$$\frac{N_D}{1 + 2 \exp\left(\frac{\Delta\epsilon_F}{k_B T}\right) \exp\left(\frac{E_d}{k_B T}\right)} = N_C \frac{2}{\sqrt{\pi}} F_{1/2}\left(\frac{\Delta\epsilon_F}{k_B T}\right)$$

**donor binding energy
(30 meV)**

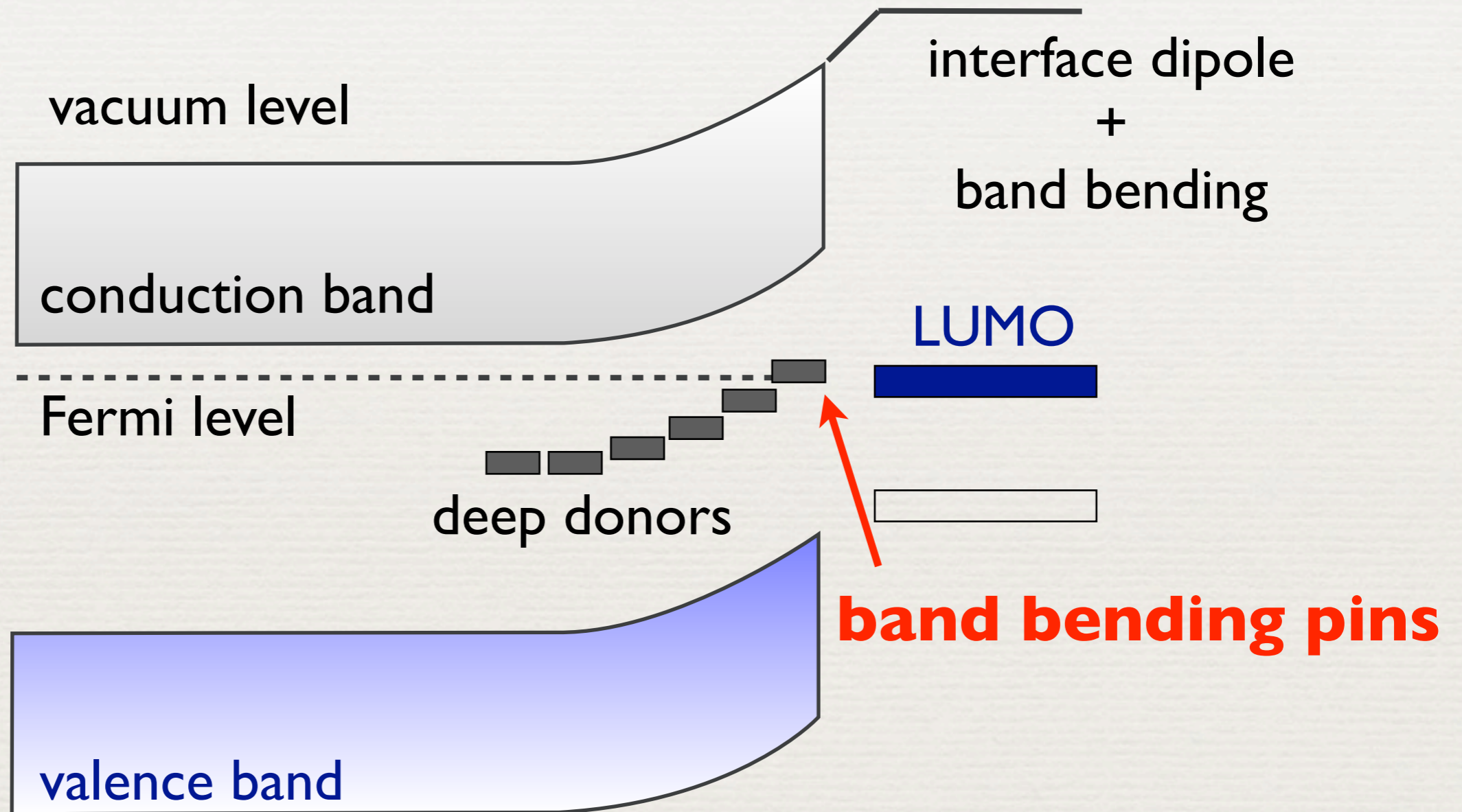
**conduction band
effective mass
(0.27 m_e)**

$$N_C = 2 \left(\frac{2\pi m^* k_B T}{h^2} \right)^{\frac{3}{2}}$$

$$F_{1/2}(\eta_F) = \int_0^{\infty} \frac{\eta^{1/2} d\eta}{1 + \exp(\eta - \eta_F)}$$

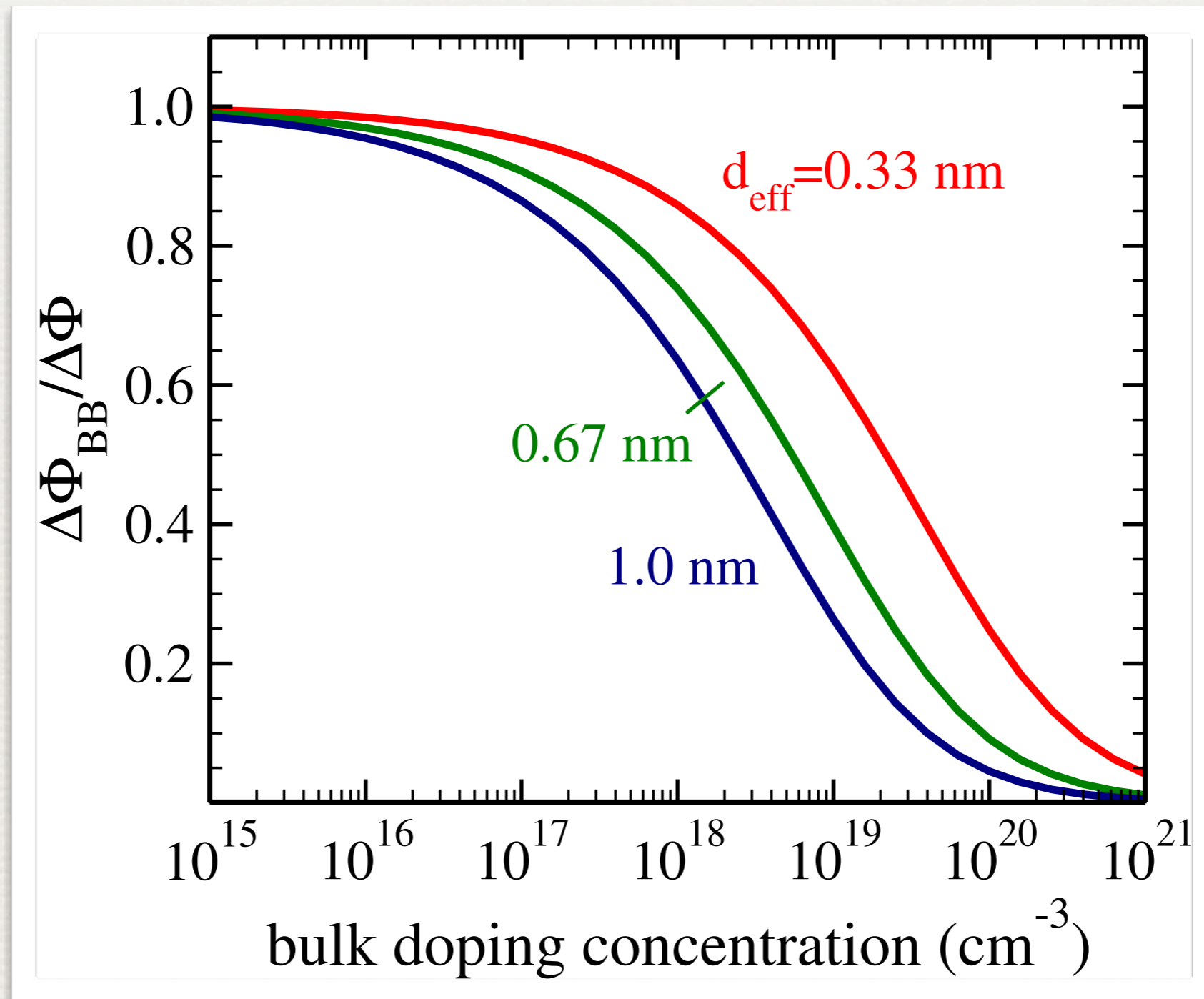
F4TCNQ@ZnO schematically

low doping + deep donors \Rightarrow band bending pins



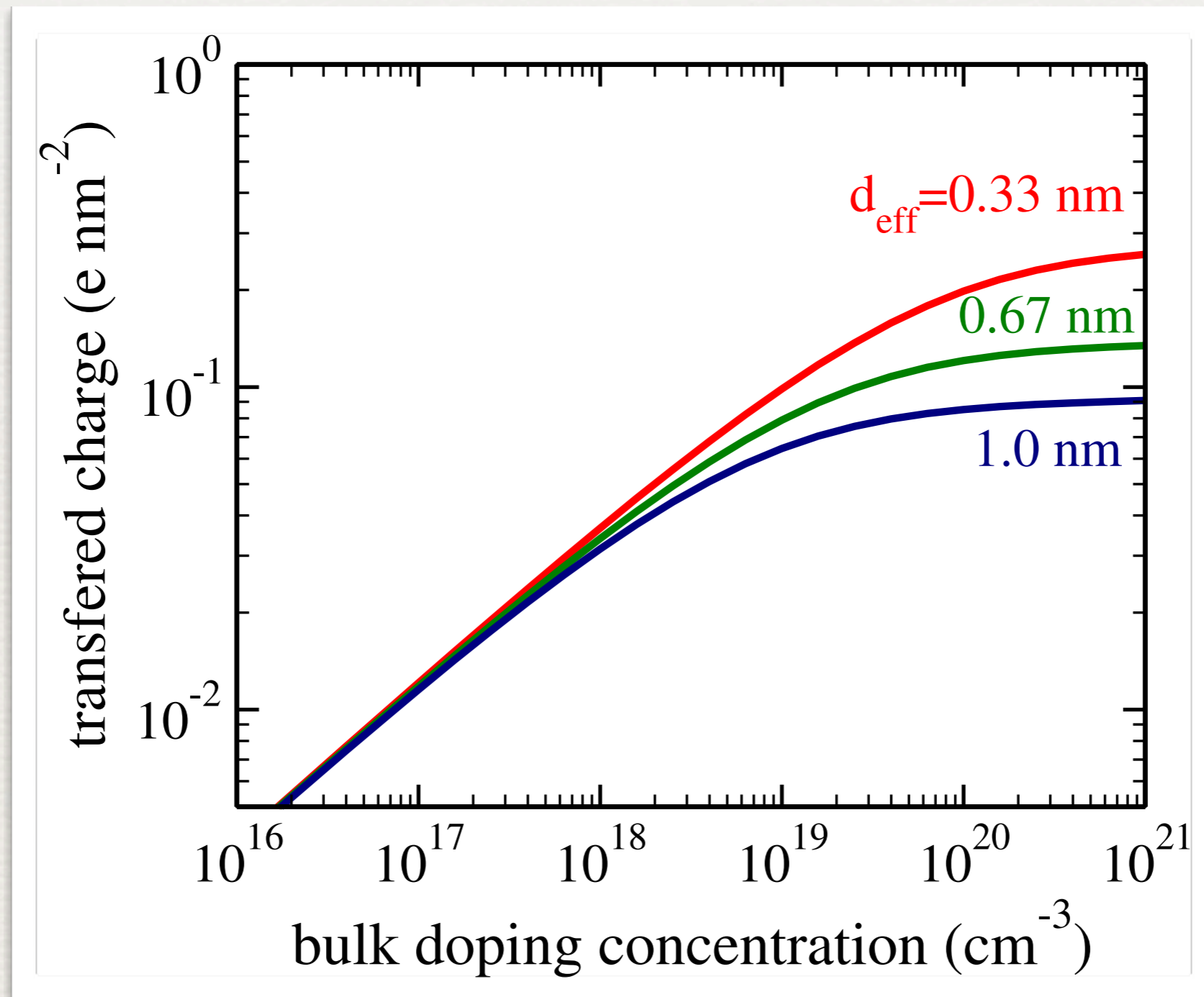
F4TCNQ@ZnO schematically

$$\Delta\Phi = \Delta\Phi_{ID} + \Delta\Phi_{BB} = \frac{\delta q}{\epsilon_0} d_{eff} + \frac{\delta q^2}{2\epsilon_0\epsilon N_D}$$



F4TCNQ@ZnO schematically

$$\Delta\Phi = \Delta\Phi_{ID} + \Delta\Phi_{BB} = \frac{\delta q}{\epsilon_0} d_{eff} + \frac{\delta q^2}{2\epsilon_0\epsilon N_D}$$



F4TCNQ@ZnO schematically

$$\Delta\Phi = \Delta\Phi_{ID} + \Delta\Phi_{BB} = \frac{\delta q}{\epsilon_0} d_{eff} + \frac{\delta q^2}{2\epsilon_0\epsilon N_D}$$

