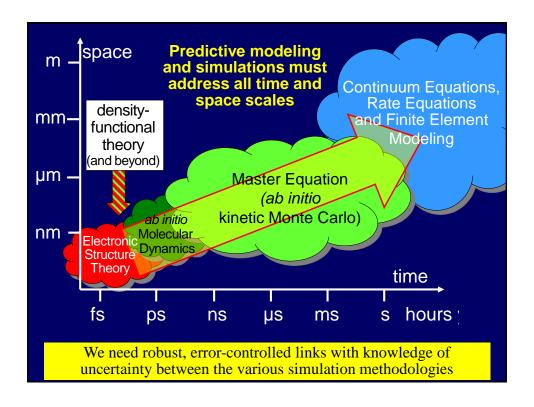
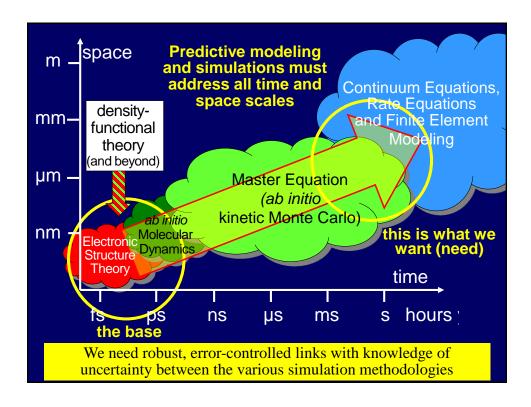
Max-Planck-EPFL Center for Molecular Nanoscience & Technol July 27 - 31, 2015 in Schloss Ringberg, Germany							
	Monday	Tuesday	Wednesday	Thursday	Friday		
8:45 - 9:55		Silke Biermann - Electronic structure calculations using dynamical mean field theory	Ivano Tavernelli - Trajectory- based nonadiabatic dynamics using time-dependent density functional theory	Cecile Hebert - Investigation of molecules at surfaces and chemical reactions by transmission electron microscopy: Is a dream becoming true?			
9:55 - 11:05		Matthieu Verstraete - Ab initio approaches to electron transport	Olle Hellman - Phonons and anharmonics	Andrea Cepellotti – Thermal Transport in 2D Materials	Alec Wodtke - The dynamics of molecular interactions and chemical reactions at metal surfaces: Testing the foundations of theory		
11:05		Coffee break	Coffee break	Coffee break	Coffee break		
11:25- 12:35		Carsten Baldauf - Molecular dynamics of peptides in isolation and computation on physical observables	Matthias Scheffler - Big-Data Analytics for Materials Science: Concepts, Challenges, and Hype	Markus Elstner -Multiscale simulations of biological structures and processe	Examinations		
12:35		Lunch	Lunch	Lunch	Lunch		
14:15 - 15:25		Tom Rizzo - Biomolecules in isolation - challenges and benchmarks for theory	Christian Carbogno – Thermal Conductivities from First Principles Molecular Dynamics				
15:25	Coffee break	Coffee break	Coffee break				
15:50 - 17:00	Matthias Scheffler - Electronic Structure Theory: Introduction and Overview	Michele Ceriotti - Representing and understanding patterns in materials and molecules	Sergey Levchenko – Defects In Solids at realistic conditions	Group discussion			
17:00 - 18:10	Claudia Draxi - Beyond density-functional theory: GW and the Bethe- Salpeter equation	Klaus Kern - Molecular nanostructures at surfaces	Karsten Jacobsen - Computational Screening of Light- Absorbing Materials				
18:30	Dinner break	Dinner break	Dinner break	Conference Dinner			





Modeling Materials Properties and Functions: The Many-Body Schrödinger Equation

The Many-Body Schrödinger Equation
$$(\hat{T}^e + \hat{T}^{ion} + \hat{V}^{e-e} + \hat{V}^{e-ion} + \hat{V}^{ion-ion}) \Psi = E \Psi$$
 With:
$$\Psi(\mathbf{r_1}, \cdots \mathbf{r_N}; \mathbf{R_1}, \cdots \mathbf{R_M})$$

$$\hat{T}^e = \sum_{k=1}^N \frac{\mathbf{p}_k^2}{2m} \qquad \hat{T}^{ion} = \sum_{I=1}^M \frac{\mathbf{p}_I^2}{2M_I}$$

$$\hat{V}^{e-e} = \frac{1}{2} \frac{1}{4\pi\epsilon_0} \sum_{k \neq k'}^{N,N} \frac{e^2}{|\mathbf{r_k} - \mathbf{r_{k'}}|}$$

$$\hat{V}^{ion-ion} = \frac{1}{2} \frac{1}{4\pi\epsilon_0} \sum_{I \neq I'}^{M,M} \frac{Z_I Z_{I'}}{|\mathbf{R_I} - \mathbf{R_{I'}}|}$$

$$\hat{V}^{e-ion}(\mathbf{r_k}, \mathbf{R_I}) = \sum_{k=1}^N \sum_{I=1}^M \mathbf{v_I^{ion}}(|\mathbf{R_I} - \mathbf{r_k}|)$$

Modeling Materials Properties and Functions: The Many-Body Schrödinger Equation

$$(\hat{T}^e + \hat{T}^{ion} + \hat{V}^{e-e} + \hat{V}^{e-ion} + \hat{V}^{ion-ion})\Psi = E\Psi$$
With: $\Psi(\mathbf{r_1}, \cdots \mathbf{r_N}; \mathbf{R_1}, \cdots \mathbf{R_M})$

Dirac: "The underlying physical laws necessary for the mathematical theory of a large part of physics and the whole of chemistry are thus completely known, and the difficulty is only that the exact application of these laws leads to equations much too complicated to be soluble. It therefore becomes desirable that approximate practical methods of applying quantum mechanics should be developed, which can lead to an explanation of the main features of complex atomic systems without too much computation."

Proceedings of the Royal Society of London. Ser. A, Vol. 123, No. 792 (6 April 1929)

Born-Oppenheimer Approximation

$$\Psi(\mathbf{r_1},\cdots\mathbf{r_N};\mathbf{R_1},\cdots\mathbf{R_M}) = \sum_{\nu} \Lambda_{\nu}(\{\mathbf{R_I}\}) \Phi_{\nu} \, {}_{\{\mathbf{R_I}\}}(\{\mathbf{r_k}\})$$

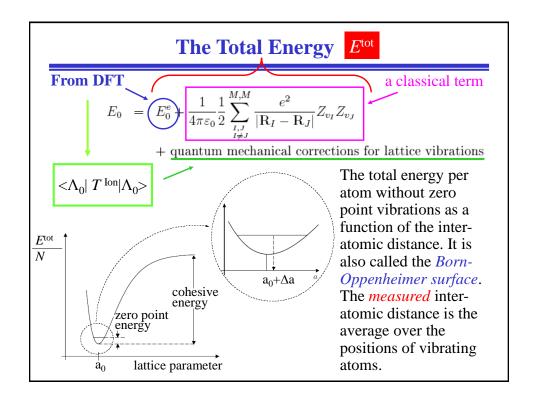
Where Φ_{v} are solutions of the "electronic Hamiltonian":

$$H_{\{\mathbf{R}_{\mathbf{I}}\}}^{e} \Phi_{\nu, \{\mathbf{R}_{\mathbf{I}}\}}(\{\mathbf{r}_{\mathbf{k}}\}) = E_{\nu, \{\mathbf{R}_{\mathbf{I}}\}}^{e} \Phi_{\nu, \{\mathbf{R}_{\mathbf{I}}\}}(\{\mathbf{r}_{\mathbf{k}}\})$$

$$H^{e} = T^{e} + V^{e-e} + V^{e-ion}$$

frequently (commonly) applied approximations:

- neglect non-adiabatic coupling (terms of order m/M_I)
- keep only Λ_0
- the dynamics of electrons and nuclei decouple



Some Limits of the Born-Oppenheimer Approximation

It does not account for correlated dynamics of ions and electrons. For example:

- polaron-induced superconductivity
- dynamical Jahn-Teller effect at defects in crystals
- some phenomena of diffusion in solids
- non-adiabaticity in molecule-surface scattering and chemical reactions
- relaxation and transport of charge carriers (e- or h)
- etc.

Some Limits of the Born-Oppenheimer Approximation

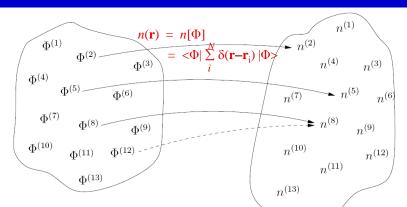
It does not account for correlated dynamics of ions and electrons. For example:

These limits can be severe. Nevertheless, we will use the BO approximation in the following.

How can we solve:

$$H^e_{\{\mathbf{R_I}\}}\Phi_{\nu,\{\mathbf{R_I}\}}(\mathbf{r_k}) = E^e_{\nu,\{\mathbf{R_I}\}}\Phi_{\nu,\{\mathbf{R_I}\}}(\mathbf{r_k})$$
 by $H^e = T^e + V^{e-e} + V^{e-ion}$

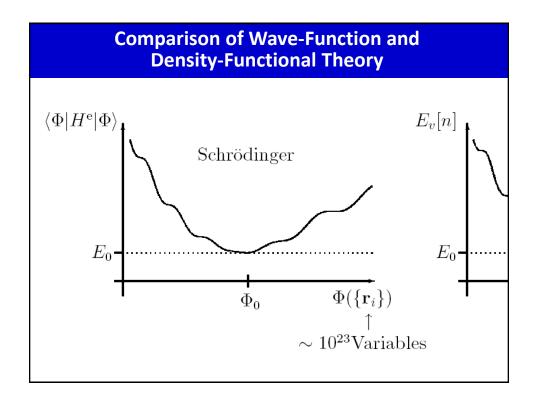
The Hohenberg-Kohn Theorem (1964)

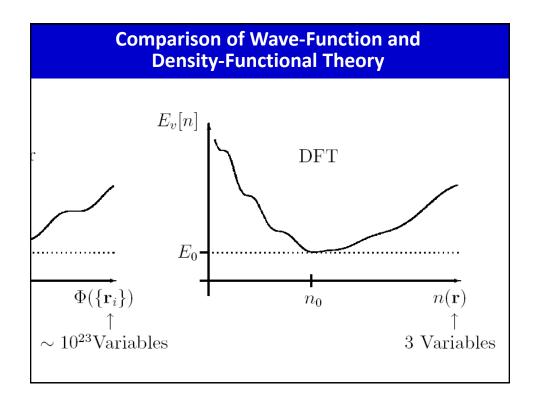


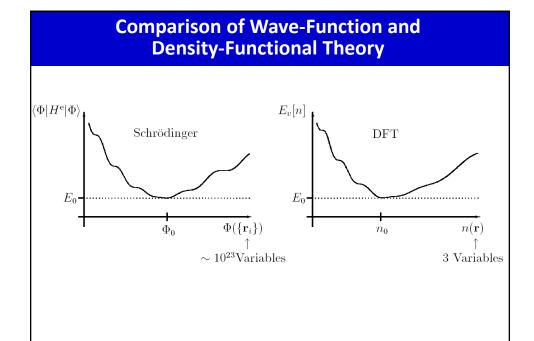
Set of non-degenerate ground-state wave functions Φ of arbitrary N-electron Hamiltonians.

Set of particle densities $n(\mathbf{r})$ belonging to non-degenerate N-electron ground states.

The dashed arrow is not possible. Thus, here is a one-to-one correspondence between Φ and $n(\mathbf{r})$.







Density Functional Theory

The energy of the ground state of a many-electron system : $E_0\left(\{\mathbf{R}_I\}\right) = \mathrm{Min}_\Phi <\Phi|H^e|\Phi>$

Hohenberg and Kohn (1964): The functional

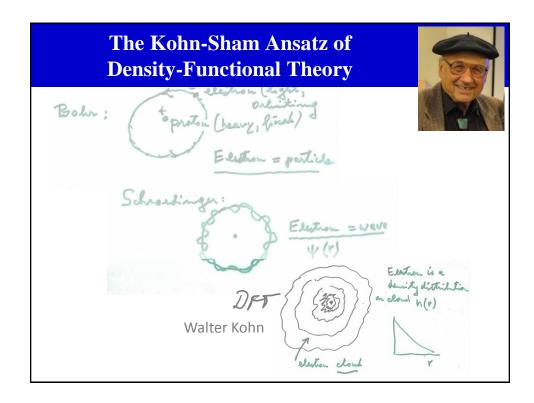
$$n(\mathbf{r}) = n[\Phi] = \langle \Phi | \sum_{i} \delta(\mathbf{r} - \mathbf{r}_{i}) | \Phi \rangle$$

can be inverted, i.e.,

$$\Phi(\mathbf{r}_1, \mathbf{r}_2, \dots, \mathbf{r}_N) = \Phi[n(\mathbf{r})] .$$

This implies:

$$E_0\left(\left\{\mathbf{R}_I\right\}\right) = \operatorname{Min}_{n(\mathbf{r})} E_{\left\{\mathbf{R}\right\}}\left[n\right]$$



The Kohn-Sham Ansatz of Density-Functional Theory



 Kohn-Sham (1965): Replace the original many-body problem by an independent electron problem that can be solved!

$$E_{v}[n] = T_{s}[n] + \int v(\mathbf{r}) n(\mathbf{r}) d^{3}\mathbf{r} + E^{\text{Hartree}}[n] + E^{xc}[n]$$

- With $T_{\rm s}$ [n] the kinetic energy functional of independent electrons, and $E^{\rm xc}[n]$ the unknown functional.
- The challenge is to find useful, approximate xc functionals.

Kohn and Sham (1965):

$$E_v[n] = T_s[n] + \int v(\mathbf{r})n(\mathbf{r})d^3\mathbf{r} + E^{\text{Hartree}}[n] + E^{\text{xc}}[n]$$
with
$$E^{\text{Hartree}}[n] = \frac{1}{2} \frac{e^2}{4\pi\varepsilon_0} \iint \frac{n(\mathbf{r})n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d^3\mathbf{r} d^3\mathbf{r}'$$

And $T_s[n]$ the functional of the kinetic energy of **noninteracting** electrons. $E^{xc}[n]$ contains all the unknowns.

At fixed electron number N the variational principle gives

$$\delta \left\{ E_v[n] - \mu \left(\int n(\mathbf{r}) d^3 \mathbf{r} - N \right) \right\} = 0$$

$$\frac{\delta E_v^{[n]}}{\delta_n} = \mu = \frac{\delta T_s[n]}{\delta n(\mathbf{r})} + v^{\text{eff}}(\mathbf{r})$$
Kohn-Sham equation

Kohn and Sham (1965):

$$v^{\text{eff}}(\mathbf{r}) = v(\mathbf{r}) + \frac{e^2}{4\pi\varepsilon_0} \int \frac{n(\mathbf{r}')}{|\mathbf{r} - \mathbf{r}'|} d^3\mathbf{r}' + \frac{\delta E^{\text{xc}}[n]}{\delta n(\mathbf{r})}$$

Because $T_s[n]$ is the functional of non-interacting particles we effectively restrict the allowed densities to those that can be written as

This implies:

$$n(\mathbf{r}) = \sum_{i=1}^{N} |\varphi_i(\mathbf{r})|^2$$
 Kohn-Sham

$$\left\{-\frac{\hbar^2}{2m}\nabla^2+v^{\rm eff}(\mathbf{r})\right\}\varphi_{i,i}(\mathbf{r})=\epsilon_i^{}\varphi_{i}^{}(\mathbf{r}) \qquad \text{equation}$$

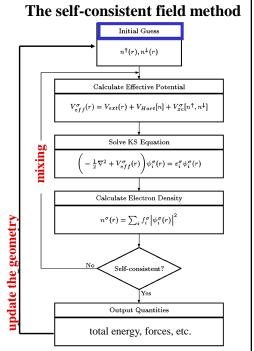
 $v^{\rm eff}(\mathbf{r})$ depends on the density that we are seeking.

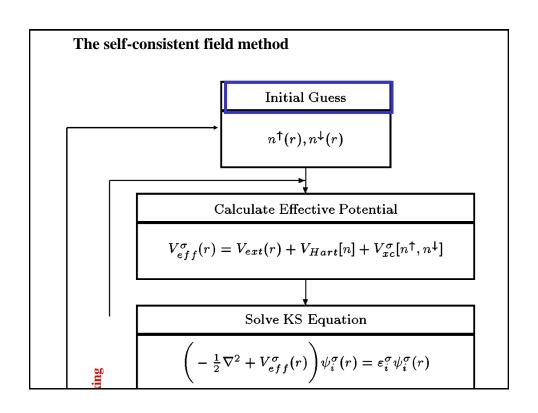
$$T_s[n] = \sum_{k=1}^{N} \langle \varphi_k | -\frac{\hbar^2}{2m} \nabla^2 | \varphi_k \rangle ,$$

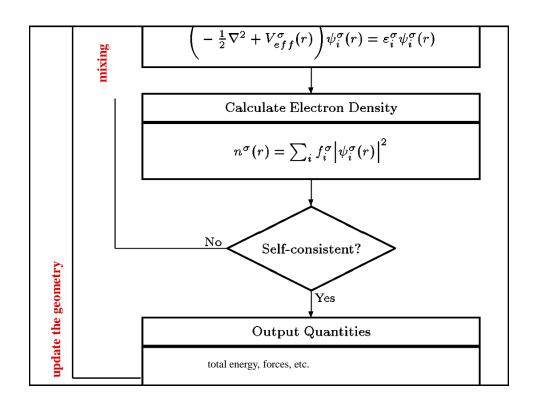
$$= \sum_{k=1}^{N} \epsilon_k - \int v^{\text{eff}}[n^{\text{in}}](\mathbf{r}) n(\mathbf{r}) d^3 \mathbf{r}$$

Solving the Kohn-Sham Equations

- Structure, types of atoms
- · Guess for input
- Solve KS Eqs.
- New density
- Self-consistent?
- Output:
 - Total energy, force, ...
 - KS eigenvalues









 T_s , $E^{\rm Hartree}$, and $E^{\rm xc}$ are all *universal* functionals in $n({\bf r})$, i.e., they are independent of the special system studied. (general theory: see the work by Levy and Lieb)

$$E^{\text{xc}}[n] = \int \epsilon^{\text{xc}}[n] n(\mathbf{r}) d^3\mathbf{r} = E^{\text{xc-LDA}}[n] + O(\nabla n)$$

$$E^{\text{xc-ipellium}}(n) = \int \epsilon^{\text{xc-ipellium}}(n) n(\mathbf{r}) d^3\mathbf{r}$$

$$E^{\text{xc-ipellium}}(n) = \int \epsilon^{\text{xc-ipellium}}(n) n(\mathbf{r}$$

The xc Functional

 T_s , $E^{\rm Hartree}$, and $E^{\rm xc}$ are all *universal* functionals in $n({\bf r})$, i.e., they are independent of the special system studied. (general theory: see the work by Levy and Lieb)

$$E^{\text{xc}}[n] = \int \epsilon^{\text{xc}}[n] n(\mathbf{r}) d^3\mathbf{r} = E^{\text{xc-LDA}}[n] + O(\nabla n)$$

$$E^{\text{xc-LDA}} = \int \epsilon^{\text{xc-jellium}}(n) \ n(\mathbf{r}) \ d^3\mathbf{r}$$

$$\leftarrow \text{Wigner (1938)}$$

$$\epsilon^{\text{xc-jellium}}(n)$$

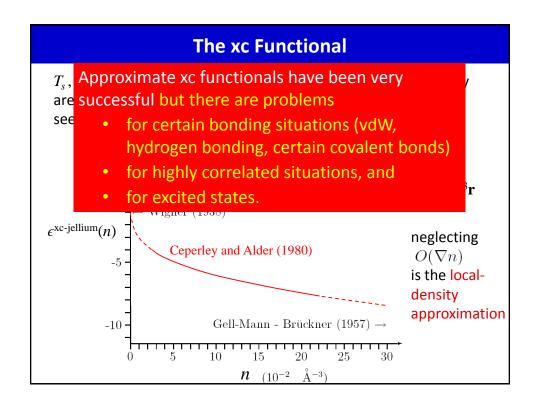
$$Ceperley \text{ and Alder (1980)}$$

$$Gell-Mann - Brückner (1957) \rightarrow 0$$

$$neglecting$$

$$O(\nabla n)$$
is the local-density approximation
$$n = n$$

$$n =$$



Certainties about Density Functional Theory

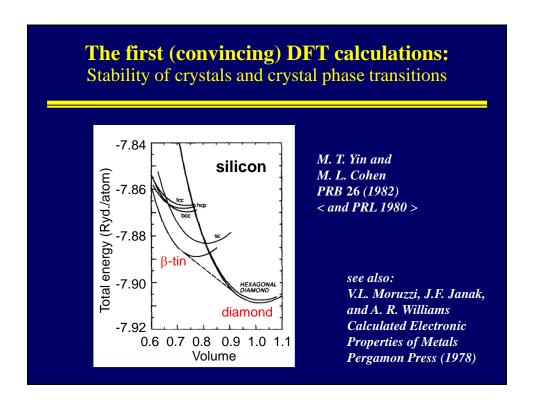
- 1. DFT in principle: It is exact; a universal $E^{xc}[n]$ functional "exists".
- 2. DFT in practice: It is probably not possible to write down $E^{xc}[n]$ as a closed mathematical expression. We need approximations.

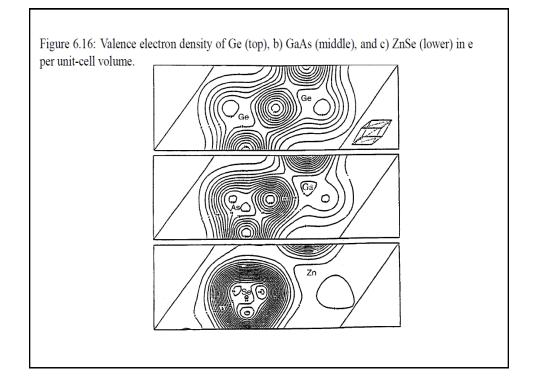
The success of DFT proves that "simple" approximations to the exchange-correlation functional can provide good results – if one knows what one is doing.

Certainties about Density Functional ory

- 1. DFT in principle: It is exact; a universely $E^{xc}[n]$ functional "exists".
- 2. DFT in practice: It is problem to write down $E^{\text{xc}}[n]$ as a close the matical expression. We need approximately $E^{\text{xc}}[n]$

The success of Dress oves that "simple" approximations to the exchange relation functional can provide good results — if the success of Dress oves that "simple" approximations to the exchange relation functional can provide good results — if the success of Dress oves that "simple" approximations to the exchange relation functional can provide good results — if the success of Dress oves that "simple" approximations to the exchange relation functional can provide good results — if the success of Dress oves that "simple" approximations to the exchange relation functional can provide good results — if the success of Dress over the success over the success of Dress over the success over







MOST CITED PAPERS IN APS (FROM 1893)

	Journal	# cites	Title	Author(s)
1	PRB (1988)	39190	Development of the Colle-Salvetti Correlation-Energy	Lee, Yang, Parr
2	PRL (1996)	25452	Generalized Gradient Approximation Made Simple	Perdew, Burke, Ernzerhof
3	PRA (1988)	22904	Density-Functional Exchange-Energy Approximation	Becke
4	PR (1965)	20142	Self-Consistent Equations Including Exchange and Correlation	Kohn and Sham
5	PRB (1996)	13731	Efficient Iterative Schemes for Ab Initio Total-Energy	Kresse and Furthmu ll er
6	PRB (1976)	13160	Special Points for Brillouin-Zone Integrations	Monkhorst and Pack
7	PRB (1992)	10876	Accurate and Simple Analytic Representation of the Electron	Perdew and Wang
8	PRB (1999)	10007	From Ultrasoft Pseudopotentials to the Projector Augmented	Kresse and Joubert
9	PRB (1990)	9840	Soft Self-Consistent Pseudopotentials in a Generalized	Vanderbilt
10	PR (1964)	9789	Inhomogeneous Electron Gas	Hohenberg and Kohn
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12	PRB (1992)	9786	Atoms, Molecules, Solids, and Surfaces - Applications of the	Perdew, Chevary,
13	PRB (1986)	9313	Density-Functional Approx. for the Correlation-Energy	Perdew
14	PR (1934)	9271	Note on an Approximation Treatment for Many-Electron Systems	Moller and Plesset
15	PRB (1994)	9100	Projector Augmented-Wave Method	Blochi
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20	PRB (1993)	6925	Ab initio Molecular Dynamics for Liquid Metals	Kresse and Hafner
21	PR (1961)	6467	Effects of Configuration Interaction on Intensities and Phase Shifts	Fano
22	PR (1957)	6260	Theory of Superconductivity	Bardeen, Cooper, Schrieffer

list was compiled ~2010

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Perdew's Dream: Jacob's Ladder in **Density-Functional Theory**

The exchange-correlation functional our favorite

occupied $\psi_i(\mathbf{r})$, $\tau(\mathbf{r})$,

 $\nabla n(\mathbf{r})$, $n(\mathbf{r})$,

unoccupied $\psi_i(\mathbf{r})$, EX + cRPA, as given by ACFD

hybrids (B3LYP, PBEO, HSE, ...) meta-GGA (e.g., TPSS, SCAN)

Generalized Gradient Approximation Local-Density Approximation

Kohn-Sham kinetic-energy density τ(**r**):

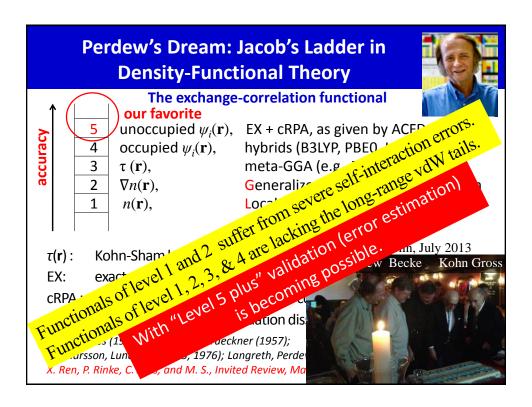
exact exchange: EX:

cRPA: random-phase approximation for co

ACFD: adiabatic connection fluctuation dis-

Bohm, Pines (1953); Gell-Mann, Brueckner (1957); Gunnarsson, Lundqvist (1975, 1976); Langreth, Perde X. Ren, P. Rinke, C. Joas, and M. S., Invited Review, Ma





"Level 5 plus" Viewed in the Many-Body Framework

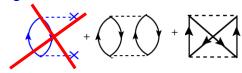
Perturbation theory:

 $H = H^0 + H'$ with $H^0/\phi_n \rangle = E_n^{(0)}/\phi_n \rangle$ and $/\phi_n \rangle =$ Slater det. $/\phi_0 \rangle =$ ground state, $/\phi_{i,a} \rangle =$ single excitations, $/\phi_{ii,ab} \rangle =$ double exci.

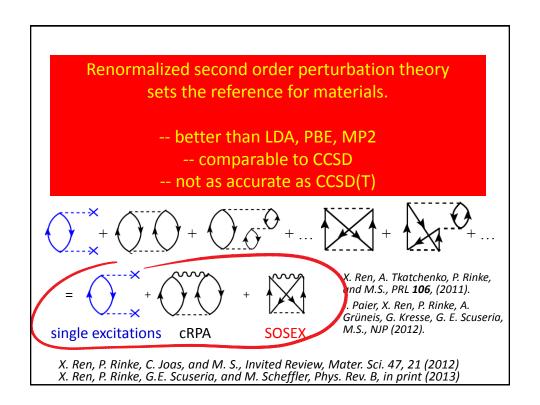
$$E_0^{(0)} = \langle \phi_0 | H^0 | \phi_0 \rangle, \quad E_0^{(1)} = \langle \phi_0 | H' | \phi_0 \rangle$$

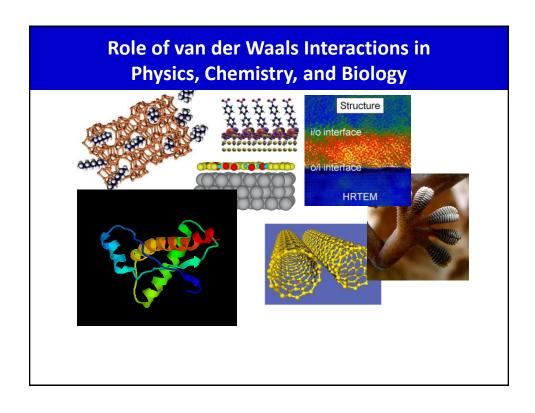
$$E_{0}^{(2)} = \sum_{n \neq 0} \frac{|\langle \phi_{0} | H' | \phi_{n} \rangle|^{2}}{E_{0}^{(0)} - E_{n}^{(0)}} = \sum_{i, \ a} \frac{|\langle \phi_{0} | H' | \phi_{i, \ a} \rangle|^{2}}{E_{0}^{(0)} - E_{i, \ a}^{(0)}} + \sum_{ij, \ ab} \frac{|\langle \phi_{0} | H' | \phi_{ij,ab} \rangle|^{2}}{E_{0}^{(0)} - E_{ij,ab}^{(0)}}$$
single excitations double excitations

Using HF input, this is Møller-Plesset perturbation theory, MP2



X. Ren, P. Rinke, C. Joas, and M. S., Invited Review, Mater. Sci. 47, 21 (2012) X. Ren, P. Rinke, G.E. Scuseria, and M. Scheffler, Phys. Rev. B, in print (2013)





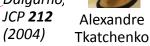
C_6 Coefficients in the TS Scheme (schematic)

S. Grimme: add tails $E_{\text{vdW}} = \sum_{l,J}^{\prime} C_{6,l,J} / R_{l,J}^{6}$

$$C_{6AB} = \frac{3}{\pi} \int_0^\infty \alpha_A(i\omega) \alpha_B(i\omega) d\omega$$

 $C_{6AA}^{\text{eff}} = \frac{\eta_A^{\text{eff}}}{\eta_A^{\text{free}}} \left(\frac{\kappa_A^{\text{free}}}{\kappa_A^{\text{eff}}}\right)^2 \left(\frac{V_A^{\text{eff}}}{V_A^{\text{free}}}\right)^2 C_{6AA}^{\text{free}}$

Chu and Dalgarno,



- on the fly -

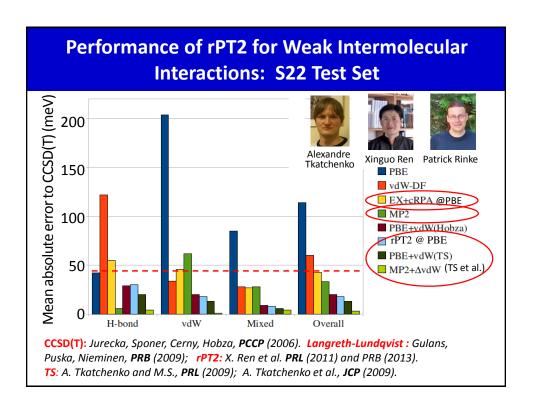
Calculated by DFT C_6 is a functional of the density, $C_6 = C_6[n]$.

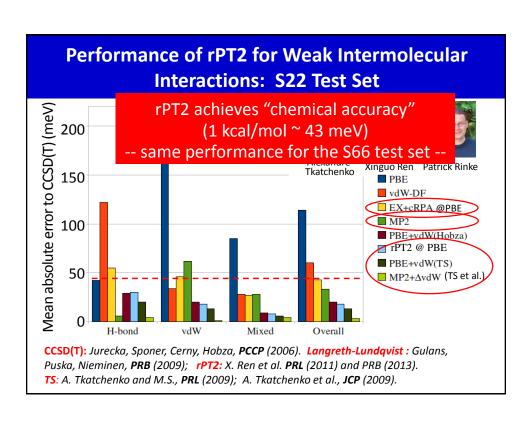
For details, see:

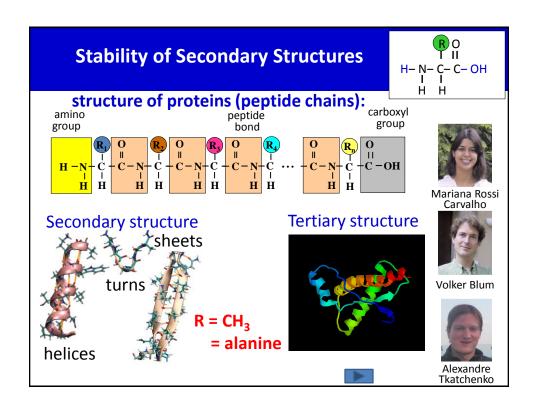
DFT+vdW: A. Tkatchenko and M.S., PRL 102 (2009).

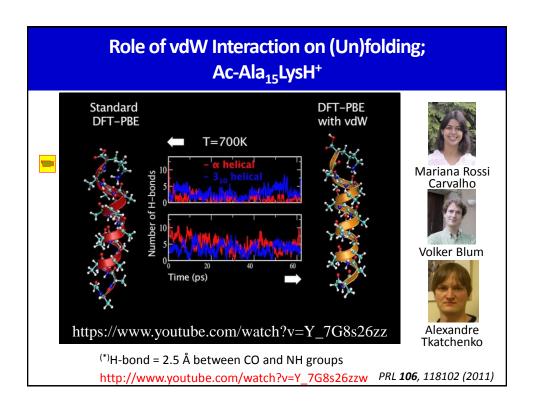
MP2+ΔvdW: A. Tkatchenko, R. DiStasio, M. Head-Gordon, and M.S., JCP (2009).

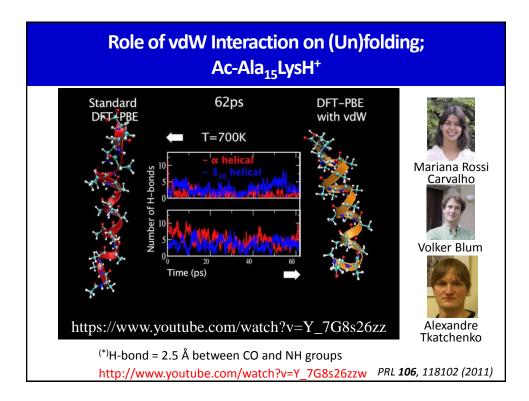
Excellent description – as long as the "pair-wise model" holds.











What Did We Learn from Studying Polypeptides?

Density-functional theory (PBE+vdW) is able to:

- Predict geometry and properties, analyze stability and unfolding:
 - hydrogen bonding,
 - vdW interaction, and
 - vibrational entropy.
- Verification of structure predictions against experiment (vibrational spectroscopy).

M. Rossi, V. Blum, et al., JPCL 1 (2010).

A Tkatchenko, M Rossi, V Blum, J Ireta, M.S., PRL 106, 118102 (2011)

http://www.youtube.com/watch?v=Y 7G8s26zzw

Ab initio Atomistic Thermodynamics

Assumption of thermodynamic equilibrium

This means that any information about time-scales is lost. We look at the system after infinite time.

Note that in reality the kinetics can be very slow; an extreme example is C-diamond \leftrightarrow graphite.

Sometimes it is important (and possible) to define a *constrained equilibrium*.

Thermodynamics (A Brief Reminder)

Thermodynamic potentials:

Internal energy $U(S, V) = E^{\text{tot}} + U^{\text{vib}}$ Enthalpy H(S, p) = U + pV

(Helmholtz) free energy F(T, V) = U - TS

Gibbs free energy G(T, p) = U - TS + pV

 E^{tot} contains energy contributions from chemical binding (structure) as well as electronic excitations, if present. U^{vib} is the energy of vibrations. For polyatomic systems E^{tot} and U^{vib} are best calculated by DFT.

At thermodynamic equilibrium the corresponding thermodynamic potential assumes its minumum.

Chemical potential $\mu = (\partial G / \partial N)_{T, p} =$

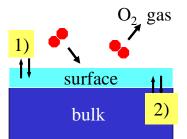
= the cost to remove a particle from the system.

Ab Initio Atomistic Thermodynamics Effect of a Surrounding Gas Phase on Surface Structure and Composition

- C. M. Weinert and M. Scheffler, Mat. Sci. Forum 10-12, 25 (1986).
- E. Kaxiras et al., Phys. Rev. B 35 9625 (1987)
- M. Scheffler and J. Dabrowski, Phil. Mag. A 58, 107 1988).
- K. Reuter and M. Scheffler, Phys. Rev. B 65, 035406 (2002)

Surface in Contact with a One-Component Gas Phase – Example: O₂ @ Pd

$$\gamma_{\rm \, surf.} = 1/A$$
 [$G\left(N_{\rm O},\, N_{\rm Pd}\right) - N_{\rm O}\,\,\mu_{\rm O}$ - $N_{\rm Pd}\,\,\mu_{\rm Pd}$]



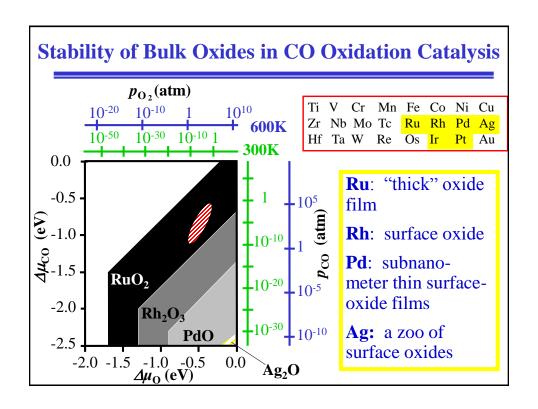
 $N_{\rm O} = \#$ of O atoms in the system $N_{\rm Pd} = \#$ of Pd atoms (substrate) in the system

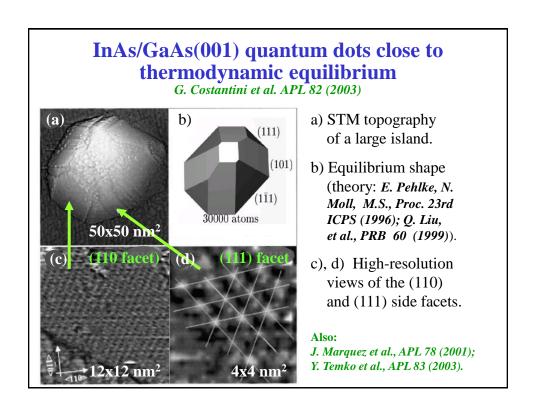
Concept of thermodynamic reservoirs:

1) The environment can give and take O atoms at an energy $\mu_{\rm O} = 1/2 \, \mu_{\rm O}$,

$$\mu_{\rm O} (T, p) = \frac{1}{2} \mu_{\rm O_2} (T, p^0) + \frac{1}{2} k_B T \ln(p/p^0)$$

2) Also the bulk of the substrate is practically of infinite size and acts as a reservoir for the Pd atoms: $\mu_{Pd} = g_{Pd}^{bulk}$





Summary Ab Initio Atomistic Thermodynamics

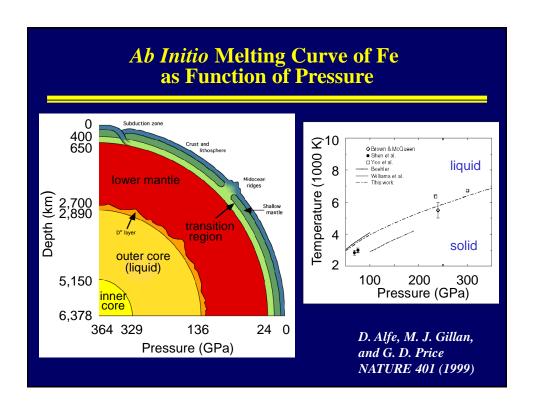
Calculate free energies by DFT.

Exploit thermodynamic equilibria and the concept of thermal reservoirs (atomic chem. potentials).

- Concentration of point defects at finite T
- Surface structure and composition in realistic environments
- Order-order and order-disorder phase transitions

Limitations:

- -- The accuracy of the xc functional (with respect to k_BT)
- -- "only" thermodynamic equilibrium



Summary and Outlook: Interacting Electrons Determine the Properties and Function of Real Materials

Important arenas for future theoretical work:

- Non-adiabatic effects, dissipation
- Transport (electrons, ions, heat)
- Thermodynamic phase transitions, e.g. melting
- Modeling the kinetics, e.g. of catalysts or crystal growth (self-assembly and self-organization) – in realistic environments
- Molecules and clusters (incl. bio molecules) in solvents, electrochemistry, fuel cells, external fields
- Correlated systems, e.g. *f*-electron chemistry
- Big-data analytics (statistical learning, compressed sensing, etc.)