

SUPPLEMENTAL INFORMATION: Steps or Terraces? Dynamics of Aromatic Hydrocarbons Adsorbed at Vicinal Metal Surfaces

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DFT CALCULATIONS

Density-functional theory calculations were performed using the all-electron full-potential FHI-aims code and numerically tabulated atom-centered orbitals (NAOs) basis set [1]. We employed self-consistent convergence criterion for the total energy, force and charge density of 10^{-4} eV, 10^{-2} eV/Å, and 10^{-5} Å⁻³, respectively. Relativistic effects for metal atoms were treated by using the scaled zeroth-order regular approximation (ZORA). Different slabs were separated by 20 Å of vacuum to eliminate the interactions between image atoms.

The transition state for diffusion processes was searched by using the LST/QST algorithm [2], as implemented in the plane-wave basis set CASTEP code [3]. The valence electrons were expanded in plane waves with a cutoff energy of 700 eV. For slab calculations, we used a $3 \times 3 \times 1$ k -points mesh.

LOW COVERAGE CONFIGURATIONS

At low coverage, benzene (Bz) molecule was placed in the atop, hcp, fcc, and bri sites with 0° and 30° rotations. Naphthalene (Np) molecules were placed with an addition of the second aromatic ring to the 0° case, namely with the long axis molecule parallel to the Cu step edge. Note that we have systematically considered 70 initial structures for Bz/Cu(443) and 40 structures for Np/Cu(443) systems, according to their distances d between the center of the molecule and the adjacent step edges.

Figure 1 shows the stable adsorption structures for the Np/Cu(443) system, in analog to Figure 1c in the main text, which indicates molecule–substrate binding heights for Bz/Cu(443).

MONOLAYER AND MULTILAYER CONFIGURATIONS

High-coverage adsorption models considered in the present work are shown in Figure 2. All these structures were created based on the most stable Bz-S1 and Np-S1 structures (see Figures 1a and 1b in the main text). Notably, there are 3 pathways from the complete monolayer (3 molecules) to the initial bilayer (4 molecules), 6 path-

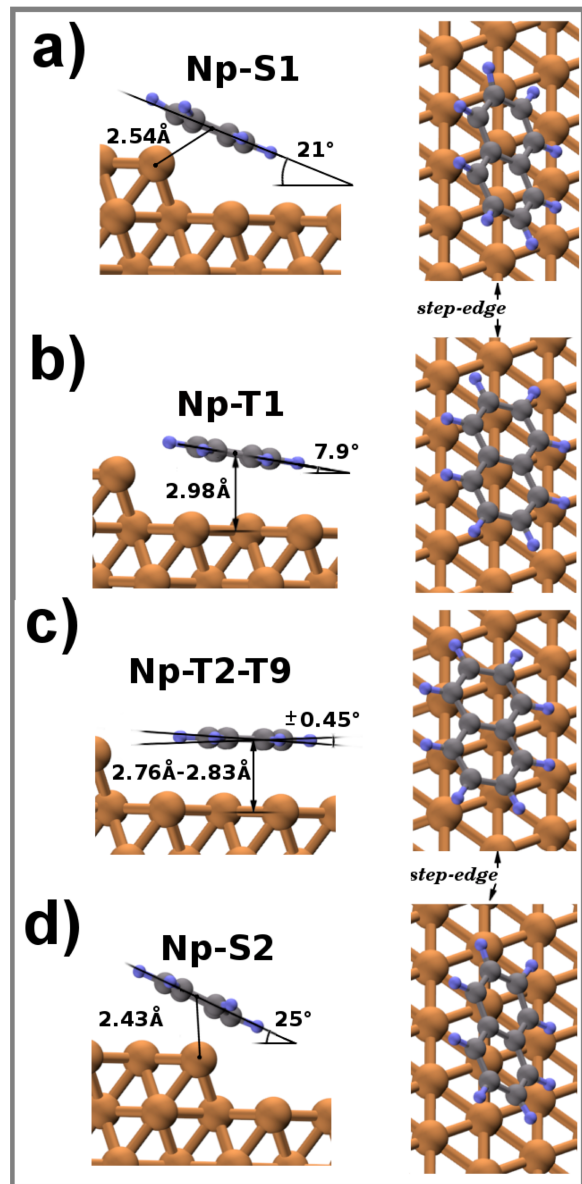


FIG. 1. Fully relaxed naphthalene structures from the PBE+vdW^{surf} method.

ways to the next bilayer coverage (5 molecules), and 3 pathways to the complete bilayer (6 molecules).

The total binding energy (*Total bE*) is defined as, $Total\ bE_n = Sys_n - surf - n \times m$, where Sys_n , $surf$, and m denote the total energies of the entire system with n molecules, the clean metal substrate, and the isolated

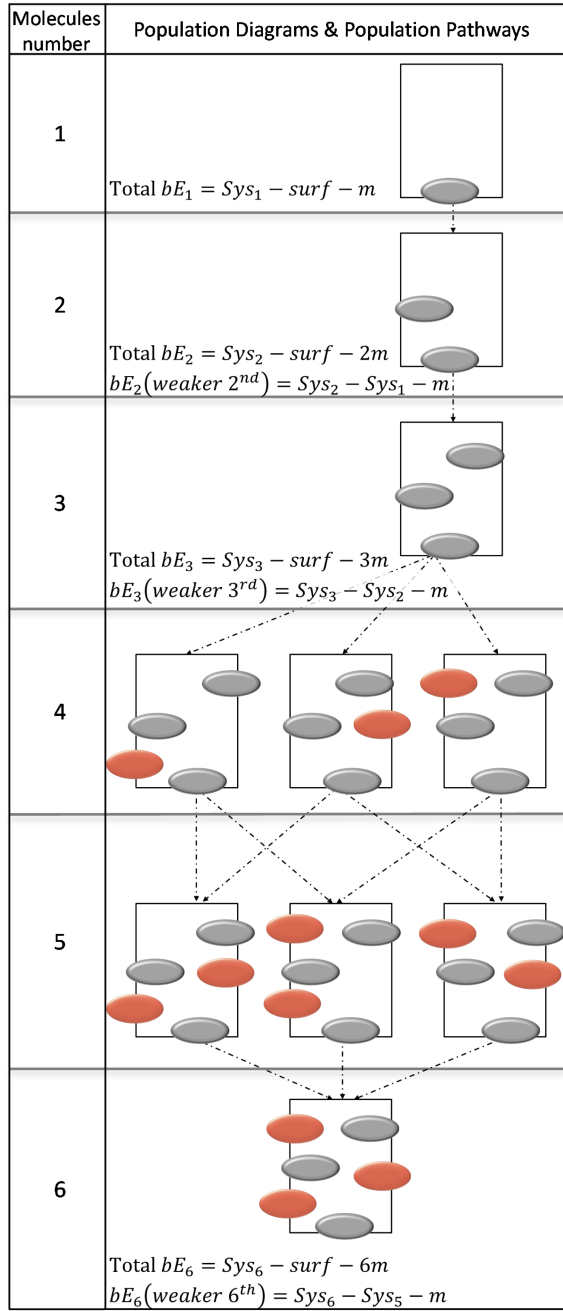


FIG. 2. Schematic plots for the population pathways evolved in the adsorption process for the mono- and bilayer coverages.

molecule, respectively. The stability of the last added molecule, n^{th} , is measured by $bE_n(weaker\ n^{th})$, which is defined as $bE_n(weaker\ n^{th}) = Sys_n - Sys_{(n-1)} - m$. The computed binding energies for Bz and Np on the Cu(443)

surface at different coverages are shown in Tables I to III.

TABLE I. Binding energies calculated by PBE+vdW^{surf} for benzene and naphthalene configurations at low coverages at the Cu(443) surface.

System	Benzene B-E (eV)	Naphthalene B-E (eV)
9×1	-1.34	-2.02
8×1	-1.34	-2.00
7×1	-1.32	-2.00
6×1	-1.32	-2.05
5×1	-1.31	-2.02
4×1	-1.31	-1.99

TABLE II. Binding energies calculated by PBE+vdW^{surf} for benzene configurations at monolayer and bilayer coverages at the Cu(443) surface.

System	Molecule number	Total B-E (eV)	Weaker mol. B-E (eV)
	1	-1.32	-1.32
	2	-2.29	-0.98
	3	-3.14	-0.85
	4	-3.69	-0.55
	5	-4.55	-0.73
4×1	6	-4.47	-0.64

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TABLE III. Binding energies calculated by PBE+vdW^{surf} for naphthalene configurations at monolayer and bilayer coverages at the Cu(443) surface.

System	Molecule number	Total B-E (eV)	Weaker mol. B-E (eV)
	1	-2.06	-2.06
	2	-3.50	-1.44
	3	-4.83	-1.34
	4	-5.67	-0.84
	5	-6.74	-0.90
6×1	6	-7.71	-0.97